

# Direct N<sub>2</sub>O emission factors for synthetic N-fertilizer and organic residues applied on sugarcane for bioethanol production in Central-Southern Brazil

MARCOS SIQUEIRA NETO<sup>1</sup>, MARCELO V. GALDOS<sup>2</sup>, BRIGITTE J. FEIGL<sup>1</sup>, CARLOS E. P. CERRI<sup>3</sup> and CARLOS C. CERRI<sup>1</sup>

<sup>1</sup>Centro de Energia Nuclear na Agricultura, Universidade de São Paulo (CENA/USP), Av. Centenário, 303, P.O. Box. 96, 13400-970 Piracicaba, SP, Brazil, <sup>2</sup>Laboratório Nacional de Ciência e Tecnologia do Bioetanol (CTBE), R. Giuseppe Máximo Scalfaro, 10.000, P.O. Box. 6170, 13083-970 Campinas, SP, Brazil, <sup>3</sup>Escola Superior de Agricultura Luiz de Queiroz, Universidade de São Paulo (ESALQ/USP), Av. Pádua Dias, 11, 13400-970 Piracicaba, SP, Brazil

## Abstract

The production and use of biofuels have increased rapidly in recent decades. Bioethanol derived from sugarcane has become a promising alternative to fossil fuel for use in automotive vehicles. The ‘savings’ calculated from the carbon footprint of this energy source still generates many questions related to nitrous oxide (N<sub>2</sub>O) emissions from sugarcane cultivation. We quantified N<sub>2</sub>O emissions from soil covered with different amounts of sugarcane straw and determined the direct N<sub>2</sub>O emission factors of nitrogen fertilizers (applied at the planting furrows and in the topdressing) and the by-products of sugarcane processing (filter cake and vinasse) applied to sugarcane fields. The results showed that the presence of different amounts of sugarcane straw did not change N<sub>2</sub>O emissions relative to bare soil (control). N-fertilizer increased N<sub>2</sub>O emissions from the soil, especially when urea was used, both at the planting furrow (plant cane) and during the regrowth process (ratoon cane) in relation to ammonium nitrate. The emission factor for N-fertilizer was 0.46 ± 0.33%. The field application of filter cake and vinasse favored N<sub>2</sub>O emissions from the soil, the emission factor for vinasse was 0.65 ± 0.29%, while filter cake had a lower emission factor of 0.13 ± 0.04%. The experimentally obtained N<sub>2</sub>O emission factors associated with sugarcane cultivation, specific to the major sugarcane production region of the Brazil, were lower than those considered by the IPCC. Thus, the results of this study should contribute to bioethanol carbon footprint calculations.

**Keywords:** bioethanol, carbon footprint, filter cake, N-fertilizer, *Saccharum officinarum* L., vinasse

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## Introduction

N<sub>2</sub>O is a long-lived trace gas that is naturally present in the atmosphere. This gas is able to absorb infrared radiation and relay it in the form of thermic energy. The heating potential of this gas is 298 times higher than CO<sub>2</sub> (IPCC, 2001), and it participates directly in stratospheric ozone depletion (Ravishankara *et al.*, 2009). The atmospheric N<sub>2</sub>O concentration has steadily remained at 270 ppbv since the last glacial period (Flückiger *et al.*, 1999). However, by 2013, the concentration increased to 325.9 ppbv, with an average absolute increase during the last 10 years of 0.82 ppbv yr<sup>-1</sup> (WMO, 2014). The major N<sub>2</sub>O producers are the soil microorganisms that are responsible for N transformation (Bouwman, 1998). Agricultural activity with large N inputs from the

Haber–Bosch synthesis is the most significant N<sub>2</sub>O source to the atmosphere (Butterbach-Bahl *et al.*, 2013).

Brazil's 2nd National Communication to the Framework Convention of the United Nations on Climate Change estimated that 84% (456.8 Gg) of the total N<sub>2</sub>O emissions were from agricultural soils, of which 17% (77.8 Gg) were from the use of synthetic fertilizers and crop residues (Brazil, 2010).

Currently, Brazilian biofuel production is based on sugarcane. In 2013, bioethanol production was 21 million m<sup>3</sup> (UNICA, 2014) from a planted area of 9.1 million ha, of which 2.1 million ha were from new areas and reform/planting (CANASAT, 2014). The success of bioethanol production is due to the hardiness of the sugarcane crop, enabling regrowth after harvesting, high tillering production and positive energy balance (Macedo *et al.*, 2008). The expansion of sugarcane production in the central-southern region of Brazil is due to the increase in the domestic fleet of flex fuel vehicles

Correspondence: Dr Marcos Siqueira Neto, tel./fax +55 19 3429 4726, e-mail: msiqueir@usp.br

and the export demand for bioethanol (Rudorff *et al.*, 2010).

The uncertainty associated with the emissions of non-CO<sub>2</sub> gases affects the carbon footprint calculation of the bioethanol derived from sugarcane. N<sub>2</sub>O emissions from the use of different N-sources (fertilizer, by-products and crop residues) are considered to be major negative contributions to the 'savings' of this biofuel compared with fossil fuels (Smeets *et al.*, 2009). This consideration is because the estimate of N<sub>2</sub>O emissions uses a single default emission factor based only on the amount of applied N, which ignores the complex interactions between the microorganisms responsible for N<sub>2</sub>O production and environmental factors (IPCC, 2006).

The most common fertilization management practices on sugarcane plantations in the central-southern region of Brazil are N applied to planting furrows for reform and topdressing for sugarcane ratoons. Synthetic N-sources, urea and ammonium nitrate, account for over 85% of the fertilizer used in sugarcane cultivation. In addition to synthetic fertilizers, by-products of the sugarcane agribusiness production phase (filter cake and vinasse) can cause considerable N input. The application of these byproducts to the crop is common and has been extensively studied as a means of nutrient cycling for plant growth, potentially reducing input costs (Prado *et al.*, 2013).

The filter cake is obtained during the sugar manufacture, with every ton of sugarcane processed generating close to 30 kg (Veiga *et al.*, 2006) from the clarification of the juice obtained during milling. This composite is rich in phosphorus, in addition to calcium, magnesium, sulfur and micronutrients (Fravet *et al.*, 2010). The filter cake is applied to the planting furrow in doses between 15 and 40 ton ha<sup>-1</sup>. The vinasse is considered to be the major residue of the ethanol production; every liter of ethanol produced generates between 10 and 18 L of vinasse (Freire & Cortez, 2000). Vinasse is rich in organic matter and stands out as a potassium source, in addition to other minerals (Silva *et al.*, 2014). The vinasse is commonly applied as fertirrigation directly onto sugarcane straws in the ratoons at a dose of 150 m<sup>3</sup> ha<sup>-1</sup>.

The different sources (synthetic or organic), amounts applied, the application forms (furrow or topdressing, single or combined) and edaphoclimatic factors influence the resulting N<sub>2</sub>O emissions and thereby necessitate different emission factors for each N-fertilizer management process on a sugarcane plantation (Choudhary *et al.*, 2001; Bouwman *et al.*, 2002; Khalil *et al.*, 2004; Stehfest & Bouwman, 2006; Denmead *et al.*, 2010; Aguilera *et al.*, 2013; Gu *et al.*, 2013).

To increase the knowledge of the contribution of N<sub>2</sub>O emissions of sugarcane production to bioethanol carbon footprint calculations, we quantified N<sub>2</sub>O emissions

from soil covered with different amounts of sugarcane straw and determined the direct N<sub>2</sub>O emission factors of N-fertilizers (applied at planting furrows and topdressing) and by-products of sugarcane agribusiness (filter cake and vinasse) applied to sugarcane fields in the south central region of Brazil.

## Materials and methods

### Site description

The experiments were conducted in an area of 4.83 ha (22°36'35.7"S 47°36'06.5"W) that has been cultivated continuously with sugarcane since 1971. This area is located in the southwest region of Brazil, municipality of Piracicaba, São Paulo State. The regional climate is classified as Köppen's Cwa – mesothermal humid subtropical, with a dry winter and a hot and wet summer. The mean annual precipitation is 1400 mm yr<sup>-1</sup>, and the mean annual temperature is 22.5°C. The soil is classified as clayey Oxisol, a Typic Acrustox (Soil Survey Staff, 1999). Soil samples were taken to characterize the physical and chemical properties, shown in Table 1.

The experimental area was cultivated with sugarcane variety RB 86-7515. At the time of the experiments, the crop was in the third growth cycle, or second ratoon. Sugarcane has been harvested mechanically without straw burning in this area since 2009.

Three experiments were sequentially installed: the first evaluated N<sub>2</sub>O emissions from different quantities of straw deposited on the ground, the second experiment evaluated emissions due to the application of N-fertilizer and vinasse to ratoon sugarcane, and the third trial measured N<sub>2</sub>O emissions from the application of N-fertilizer and filter cake in the planting furrow during the reform of the sugarcane plantation.

**Table 1** Soil physic-chemical characteristics in the experimental area with sugarcane cultivation in the Central-southern part of the Brazil

Soil parameter	Soil layer (m)	
	0.0–0.1	0.1–0.2
Texture (g kg <sup>-1</sup> )		
Clay	676	684
Silt	101	93
Sand	223	223
Bulk density (g dm <sup>-3</sup> )	1.18	1.17
pH (CaCl <sub>2</sub> )	4.2	4.2
Total C (g kg <sup>-1</sup> )	15.1	14.5
Total N (g kg <sup>-1</sup> )	1.2	1.1
Avail. P (mg dm <sup>-3</sup> )	8	16
K <sup>+</sup> (mmol <sub>c</sub> dm <sup>-3</sup> )	2.7	2.1
Ca <sup>++</sup> (mmol <sub>c</sub> dm <sup>-3</sup> )	12	12
Mg <sup>++</sup> (mmol <sub>c</sub> dm <sup>-3</sup> )	4	4
CEC (mmol <sub>c</sub> dm <sup>-3</sup> )	70.3	76.0
BS (%)	26	24

CEC, cation exchanged capacity; BS, base saturation.

### Experimental design and treatments

The first experiment started with the quantification of the newly deposited plant residues on the soil surface by the sugarcane harvested on July 6, 2012. To that end, we collected all the straw in ten 1 m<sup>2</sup> quadrants randomly distributed in the area. All of the collected materials were dried at 60°C until a constant weight was reached. Total C and N were determined using an elemental analyzer (LECO© CN 2000®, St. Joseph, Michigan). The amount of straw produced was 15 Mg ha<sup>-1</sup>, with a total C content of 41.2% and 0.82% N. The field experiment was installed on August 8 and 9, 2012. We used a completely randomized design with four treatments (quantities of straw) and five replications. The quantities of straw were equivalent to (i) a full dose, 15 Mg ha<sup>-1</sup> dry mass straw; (ii) a 66% dose, 10 Mg ha<sup>-1</sup> dry mass straw; (iii) a 33% dose, 5 Mg ha<sup>-1</sup> dry mass straw; and (iv) no dose (control). The sample chambers were installed in the inter-row, in an area over 450 m<sup>2</sup> (an useful area of twelve 32-m-long rows of sugarcane in 1.2-m intervals). Gas sampling to determine N<sub>2</sub>O emissions began the following day and was performed weekly between August 2012 and March 2013, totaling 176 days.

The second experiment, designed to determine N<sub>2</sub>O emissions from N-fertilizer and vinasse application to sugarcane ratoons, was installed on November 12 and 13, 2012. We delimited a plot of 0.25 ha of the total area (an useful area of twenty five 32-m-long rows of sugarcane in 1.2-m intervals), and the experiment was allocated using a completely randomized design consisting of eight situations – six treatment (mineral and organic N-sources in two doses), an interaction between sources (mineral and organic) and a control, each with five replicates. N<sub>2</sub>O emissions from fertilizer were evaluated from two N-sources (ammonium nitrate at 35% of N and urea at 46% N) in doses of 80 and 120 kg N ha<sup>-1</sup>. The vinasse treatments were equivalent to doses of 150 and 300 m<sup>3</sup> ha<sup>-1</sup> of vinasse (6.45 g C L<sup>-1</sup> and 0.57 g N L<sup>-1</sup>) applied directly on the straw, and there was additionally a control treatment (without N-fertilizer or vinasse). In addition, to verify the interaction between sources (N-fertilizer plus vinasse) that usually occurs along the regrowth process, we evaluated a treatment of an 80 kg N ha<sup>-1</sup> dose of urea followed by 150 m<sup>3</sup> ha<sup>-1</sup> of vinasse. To apply the exact N-fertilizer quantity of each source, the doses were weighed on a precision balance and surface-applied in a band about 0.1 m from the plant row directly inside the chambers on the straw. The vinasse application was made using a garden watering can to an area of 1 m<sup>2</sup> bordering the row where N-fertilizer was applied. Gas sampling to determine N<sub>2</sub>O emissions began the day after the experiment installation and was performed from November to December 2012, first daily for 15 days, and then every 2–3 days following this period, for a total of 30 days.

The third experiment, designed to determine the N<sub>2</sub>O emissions from N-fertilizer and filter cake application in the sugarcane planting furrow, began with the conventional preparation of the soil for sugarcane planting on a 0.25-ha plot (an useful area of ten 40-m-long planting furrows of 1.2-m intervals). This preparation consisted of manually harvesting the sugarcane without burning the straw followed by soil tillage operations for the physical destruction of the stumps, plowing to 30 cm

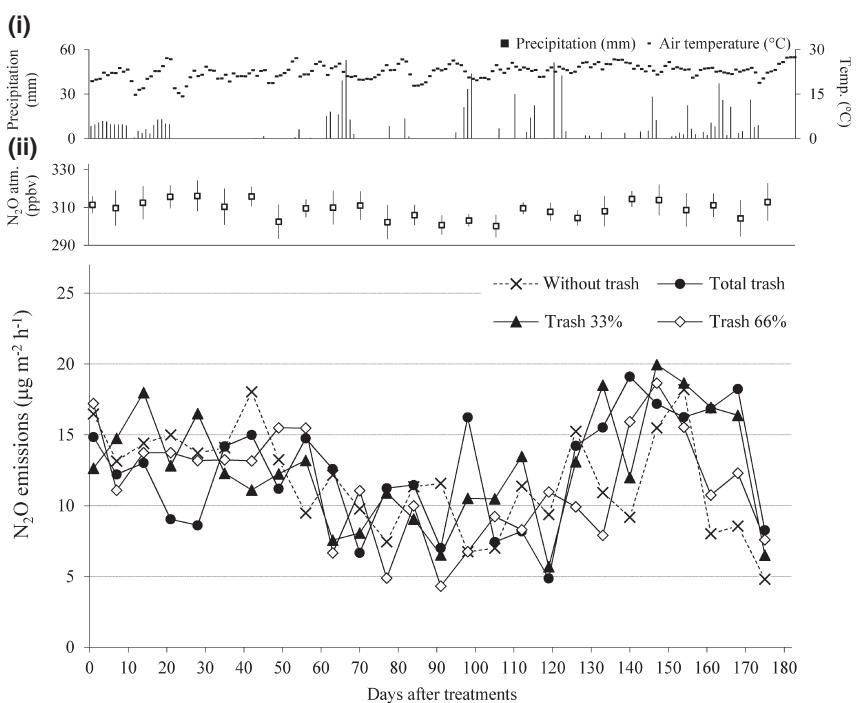
and making the planting furrows. These three operations were performed three days prior to the experiment installation, which occurred on April 29, 2013. The experimental design was completely randomized consisting of six situations with five replications. To evaluate N<sub>2</sub>O emissions from N-fertilizer, two N-sources (ammonium nitrate and urea) were tested at a single dose of 60 kg N ha<sup>-1</sup>. To determine N<sub>2</sub>O emissions from the application of filter cake, doses equivalent to 25 and 40 Mg ha<sup>-1</sup> (wet mass) of filter cake (29.9% C and 1.67% N in dry weight) were applied to the planting furrow, in addition to a control without N-fertilizer or filter cake. In addition to these treatment conditions, another treatment was performed to evaluate the interaction between the mineral and organic sources of nitrogen that usually occurs in the planting furrow. Therefore, in the bottom furrow were applied an equivalent to 25 Mg ha<sup>-1</sup> of filter cake plus an equivalent to 60 kg N ha<sup>-1</sup> urea. To apply the exact N-fertilizer or filter cake (wet mass) quantity, the doses were weighed on a precision balance and applied directly to 1 m of a planting furrow. Sugarcane stalks were placed on top of the different N-sources and the furrows were then covered with soil, as is usually done during the planting process. After the furrow was covered, the chambers were installed on the row. Gas sampling to determine N<sub>2</sub>O emissions began the day after experiment installation and was performed from April to June 2013, daily for 15 days, every 2–3 days for the next 30 days, and every 5–10 days after this period, for a total of 60 days.

### N<sub>2</sub>O sampling and analysis

The static chambers used to collect N<sub>2</sub>O emitted by the soil and added materials consisted of two parts, a base and a lid. The dimensions of the base were 45 (width) × 70 (length) × 30 (height) cm; it was buried in the soil between 5 and 7 cm deep. Air samples of the chamber's headspace were collected with a nylon syringe of 20 mL (Becton Dickinson Ind. Surgical Inc.) at four fixed time intervals (0, 10, 20 and 30 minutes) to determine the N<sub>2</sub>O concentration. The measurement of N<sub>2</sub>O concentration in the syringes was performed using a Shimadzu© GC-2014® (Kyoto, Japan) gas chromatograph with a packed Porapak™ Q column (80–100 mesh) maintained at 82°C to separate molecular gases. N<sub>2</sub>O was quantified using an electron capture detector (ECD) operating at 325°C. The N<sub>2</sub>O fluxes were calculated by the linear change in the amount of N<sub>2</sub>O in the chambers as a function of sampled time. During the sampling period, we also monitored environmental temperature and precipitation as well as ambient N<sub>2</sub>O concentration to check the order of magnitude of the N<sub>2</sub>O concentration in the chambers.

### Result analysis

In the N<sub>2</sub>O concentrations obtained for each sampling time, critical limits (lower and upper) were calculated by the box chart type (quartile). Thus, the 'outlier' values were discarded (missing subplots). We performed classical statistical analyses of the soil N<sub>2</sub>O emission results for each treatment to verify the frequency and distribution of the data. The cumulative



**Fig. 1** N<sub>2</sub>O emissions (µg m<sup>-2</sup> h<sup>-1</sup>) from different levels of sugarcane trash added to the soil surface. In the top figures (i and ii) represented environmental information during the experiment: (i) Precipitation (mm) and air temperature (°C); (ii) N<sub>2</sub>O atmospheric concentration (ppmv).

(integral) N<sub>2</sub>O emissions were calculated considering the period (days after application of nitrogen source) at which N<sub>2</sub>O emission of N treatments no longer presented a difference significantly higher ( $P < 0.05$ ) than the control treatment. Analysis of variance was performed using the Kruskal–Wallis test for the values of the cumulative N–N<sub>2</sub>O emissions for each treatment compared with their respective controls. Means of the cumulative N<sub>2</sub>O and emissions factors in different doses (same source – mineral or organic) were statistically separated using the Student–Newman–Keuls test ( $P < 0.05$ ).

#### Calculating the N<sub>2</sub>O emission factor

The emission factor for N<sub>2</sub>O was calculated only when the treatments with N-sources showed significant differences from their respective controls. In other words, the difference in the integral ( $\Sigma_{N_2O} - \Sigma_{Co}$ ) must always be positive.

To calculate the N<sub>2</sub>O emission factor (EF) due to the application of N-fertilizer or by-products, we used the methodology described in Guidelines for National Inventories of Greenhouse Gases (IPCC, 2006), according to Eqn (1):

$$EF = \left( \frac{\Sigma_{N_2O} - \Sigma_{Co}}{Napll.} \right) \quad (1)$$

Where  $\Sigma_{N_2O}$  is the total emission of N<sub>2</sub>O in the chamber from the treatment when a source containing N was applied,  $\Sigma_{Co}$  is the total N<sub>2</sub>O emission in the chamber from the control area, and Napll. is the amount of nitrogen applied as fertilizer or by-product in the chamber.

## Results

### N<sub>2</sub>O emissions from soil amended with different amounts of sugarcane straw

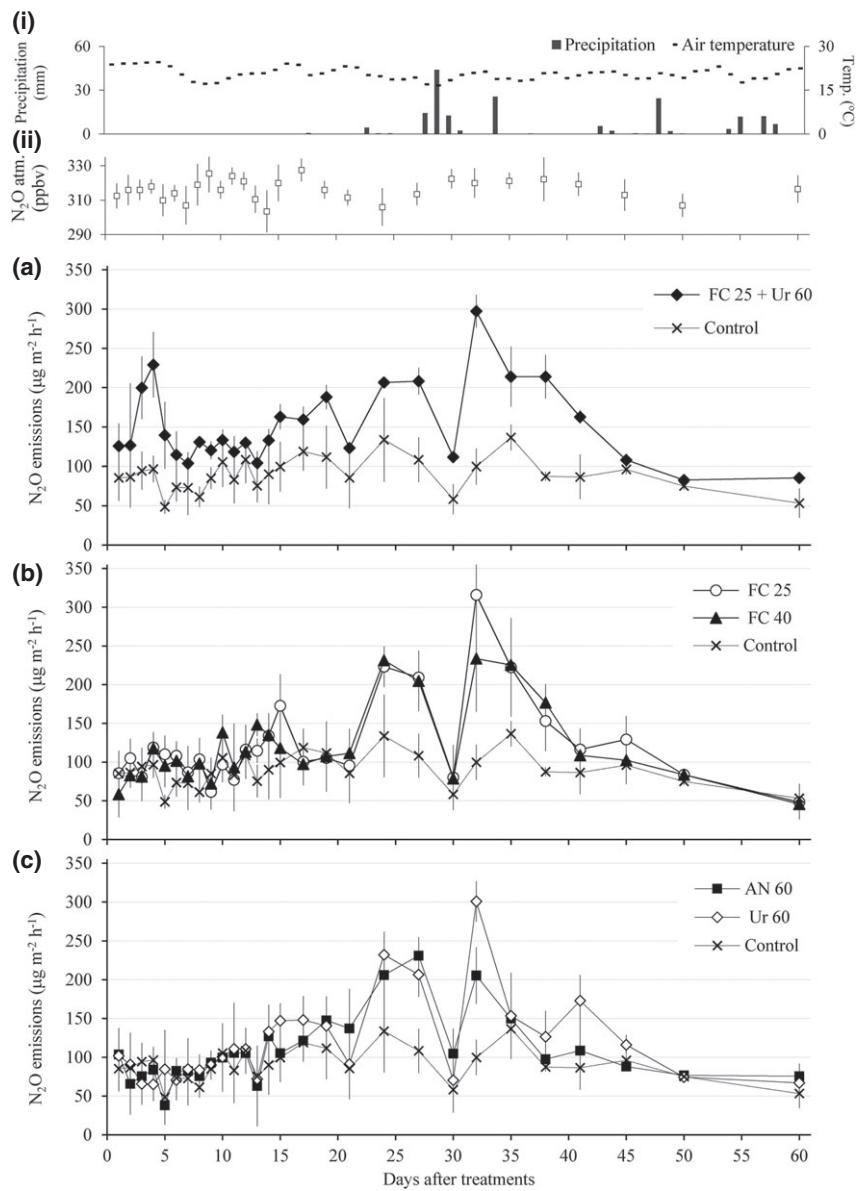
During the sampling period, the average environmental temperature was  $22.8 \pm 2.5^\circ\text{C}$ , and 86 days with precipitation events were recorded, totaling 1025.7 mm (Fig. 1i). The atmospheric concentration of N<sub>2</sub>O was  $309.8 \pm 5.1$  ppbv (Fig. 1ii).

The presence of different amounts of straw and even the absence of crop residues did not alter N<sub>2</sub>O emissions (Fig. 1). On average, treatments emitted  $\sim 12 \mu\text{g N}_2\text{O m}^{-2} \text{h}^{-1}$  (minimum = 0.0 to maximum =  $40.6 \mu\text{g N}_2\text{O m}^{-2} \text{h}^{-1}$ ) with an accumulated value of  $\sim 30 \text{ mg N-N}_2\text{O m}^{-2}$  in 176 days. No significant differences were found among treatments (Table 2).

In this experiment, emission factors were not determined because the results showed that straw should not be considered a N<sub>2</sub>O source, in that treatments (quantities of straw) showed no significant cumulative emissions compared with the control (without straw).

### Soil N<sub>2</sub>O emissions derived from the sugarcane planting process

During the period of the experiment, the average temperature was  $20.5 \pm 2.0^\circ\text{C}$  and 29 days with precipitation



**Fig. 2** N<sub>2</sub>O emissions ( $\mu\text{g m}^{-2} \text{h}^{-1}$ ) from filter cake (FC 25 = 25 Mg  $\text{ha}^{-1}$   $\approx$  195 kg N  $\text{ha}^{-1}$ ) and N-urea fertilizer (Ur 60 = 130 kg  $\text{ha}^{-1}$   $\approx$  60 kg N  $\text{ha}^{-1}$ ) applied in the furrow to the sugarcane plantation (a). From different doses of the Filter cake: FC 25 and FC 40 (40 Mg  $\text{ha}^{-1}$   $\approx$  310 kg N  $\text{ha}^{-1}$ ) (b). From different synthetic N-sources Ammonium nitrate – AN 60 (170 kg  $\text{ha}^{-1}$   $\approx$  60 kg N  $\text{ha}^{-1}$ ) and Urea – Ur 60 (c). In the top figures (i and ii) represented environmental information during the experiment: (i) Precipitation (mm) and air temperature (°C); (ii) N<sub>2</sub>O atmospheric concentration (ppmv).

events were recorded, for a total of 173.6 mm (Fig. 2i). The mean atmospheric N<sub>2</sub>O concentration was  $316.0 \pm 16.1$  ppbv (Fig. 2ii).

Figure 2a shows the daily average N<sub>2</sub>O emissions resulting from the treatment combining filter cake (FC 25 Mg  $\text{ha}^{-1}$ ) with N-fertilizer urea (Ur 60 kg N  $\text{ha}^{-1}$ ), compared with the control. N<sub>2</sub>O emissions from the organic and synthetic nitrogen sources (FC + Ur) were higher than the control from the start and throughout the sampling period, maintaining an average emission

level of  $130 \mu\text{g N}_2\text{O m}^{-2} \text{h}^{-1}$ , higher than the  $90 \mu\text{g N}_2\text{O m}^{-2} \text{h}^{-1}$  observed for the control.

On the 3rd day after treatment (DAT), N<sub>2</sub>O emissions increased in treatment FC + Ur and reached an emission three times greater than the control by the 4th DAT. This increase in N<sub>2</sub>O emissions was observed exclusively in the treatment with the combined of organic and synthetic sources. Other treatments in which synthetic (urea and ammonium nitrate) or organic (filter cake) sources were applied alone did not increase N<sub>2</sub>O emissions before the

**Table 2**  $\text{N}_2\text{O}$  emissions (mean  $\pm$  standard deviation and range), cumulative N- $\text{N}_2\text{O}$  emissions from different levels to the sugarcane trash deposited on the soil surface

Treatments	Mean $\pm$ SD*	Range†	Cumulative N- $\text{N}_2\text{O}$ ‡
	$\mu\text{g N}_2\text{O m}^{-2} \text{h}^{-1}$	$\text{mg N-N}_2\text{O m}^{-2}$	
Without trash	11.7 $\pm$ 7.0	1.1–34.1	28.7 $\pm$ 4.3 ns
Trash 33%	12.6 $\pm$ 7.5	1.2–33.8	30.6 $\pm$ 3.0
Trash 66%	11.6 $\pm$ 10.0	0.6–39.7	28.0 $\pm$ 3.0
Total trash	12.5 $\pm$ 11.1	0.5–40.6	29.9 $\pm$ 3.7

\*Mean  $\pm$  SD = standard deviation.

†Range = Minimum and maximum.

‡Cumulative N- $\text{N}_2\text{O}$  = sampled time = 175 daysns = Cumulative means without significant differ in the Student–Newman–Keuls test ( $P < 0.05$ ).

20th DAT, when the first precipitation occurred since the beginning of the experiment (Fig. 2b and c).

At the 20th DAT, the first rainfall of the period occurred (0.5 mm – 18th DAT) and emissions began to respond to rainfall. The lower  $\text{N}_2\text{O}$  emissions at the 30th DAT can be attributed to consecutive high rainfall events (~70 mm) that occurred the day before sampling. After the effect of the heavy rainfall ceased, the highest average emission of  $\text{N}_2\text{O}$  was recorded at the 32th DAT (~300  $\mu\text{g N}_2\text{O m}^{-2} \text{h}^{-1}$ ), approximately three times the emission observed for the control. After the 40th DAT, emission values of the treatment with N application decreased until they were similar to the control.

**Table 3**  $\text{N}_2\text{O}$  emissions (mean  $\pm$  standard deviation and range), cumulative N- $\text{N}_2\text{O}$  emissions and  $\text{N}_2\text{O}$  emissions factor from N-sources (mineral and organic) applied in the sugarcane planting

Treatments (sources/levels)	N <sub>2</sub> O emissions			N <sub>2</sub> O emission factor	
	Mean $\pm$ SD*	Range†	Cumulative N- $\text{N}_2\text{O}$ ‡	N-source/level§	N-source¶
	$\mu\text{g N}_2\text{O m}^{-2} \text{h}^{-1}$	$\text{mg N-N}_2\text{O m}^{-2}$	% N-applied		
Control	89.3 $\pm$ 31.0	12.9–190.8	80.3 $\pm$ 9.7 <sup>B</sup>		
Mineral source					
Ammonium nitrate (60 kg N $\text{ha}^{-1}$ )	110.9 $\pm$ 53.7	10.8–268.6	108.1 $\pm$ 4.9 <sup>A</sup>	0.44 $\pm$ 0.08 <sup>ns</sup>	0.48 $\pm$ 0.12 <sup>a</sup>
Urea (60 kg N $\text{ha}^{-1}$ )	119.3 $\pm$ 60.8	15.5–335.4	113.0 $\pm$ 9.5 <sup>A</sup>	0.52 $\pm$ 0.15	
Organic source					
Filter cake (25 Mg $\text{ha}^{-1}$ )	123.3 $\pm$ 60.3	12.2–332.6	112.0 $\pm$ 8.8 <sup>A</sup>	0.17 $\pm$ 0.02 <sup>ns</sup>	0.13 $\pm$ 0.04 <sup>b</sup>
Filter cake (40 Mg $\text{ha}^{-1}$ )	123.5 $\pm$ 57.8	14.0–326.5	114.6 $\pm$ 4.7 <sup>A</sup>	0.10 $\pm$ 0.03	
Filter cake (25 Mg $\text{ha}^{-1}$ ) + urea (60 kg N $\text{ha}^{-1}$ )	156.7 $\pm$ 75.3	10.4–463.7	132.5 $\pm$ 14.7 <sup>A</sup>	0.21 $\pm$ 0.05	

\*Mean  $\pm$  SD = standard deviation.

†Range = Minimum and maximum.

‡Cumulative  $\text{N}_2\text{O}$  = sampled time in the sugarcane planting = 60 days.§ $\text{N}_2\text{O}$  emissions factor from each N-sources and level applied in the sugarcane planting.¶N<sub>2</sub>O emissions factor from N-source applied in the sugarcane planting. Different capital letter to the Cumulative N- $\text{N}_2\text{O}$  (N-source and dose  $\times$  control) and small letter to the Emission Factor (N-source/level and N-source) with significant differ in Student–Newman–Keuls test ( $P < 0.05$ ); ns, not significant.

Cumulative  $\text{N}_2\text{O}$  emissions from the filter cake application were significantly higher than the control (40%), but the two doses (25 and 40 Mg  $\text{ha}^{-1}$ ) did not show a significant difference in  $\text{N}_2\text{O}$  emissions (Fig. 2b).

In general, the pattern of  $\text{N}_2\text{O}$  emissions from soils treated with ammonium nitrate and urea fertilizers were similar (Table 3). However, more pronounced peaks were observed in the treatment with the urea application (Fig. 2c).

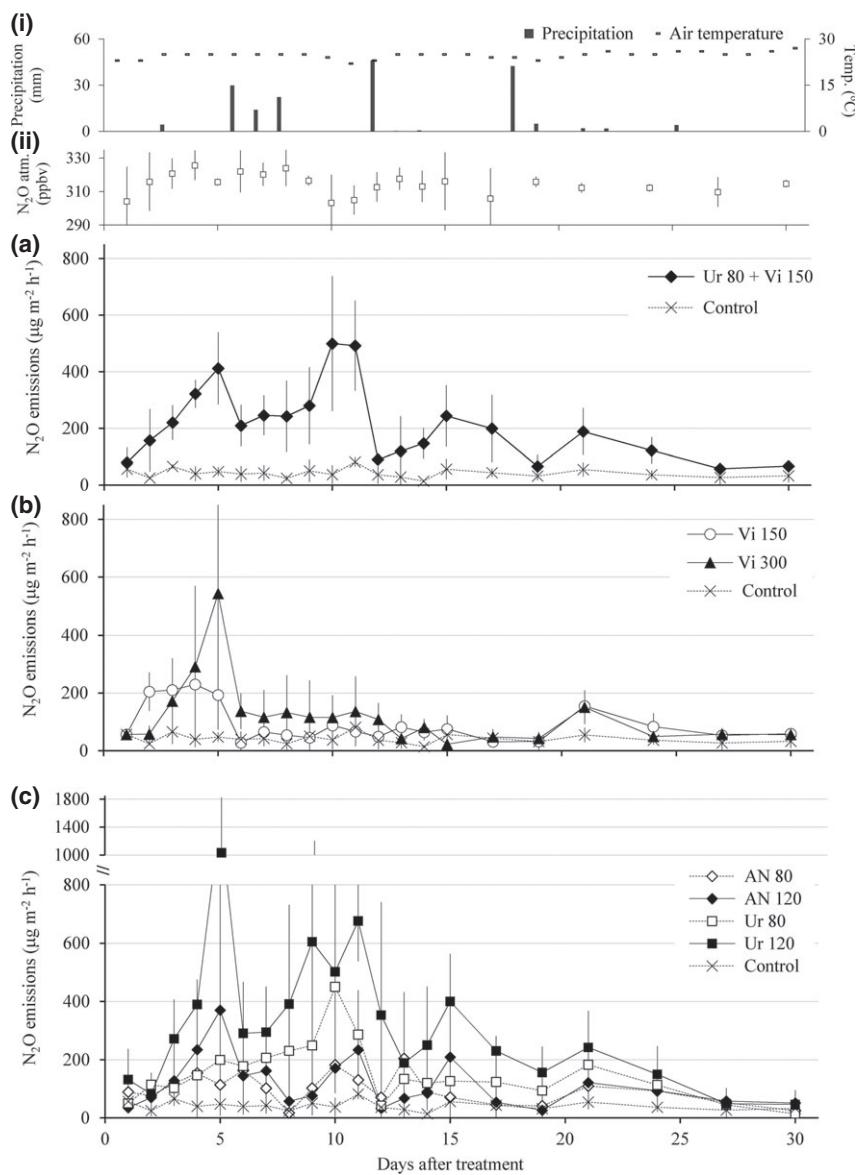
The accumulated values showed no significant difference between the N-sources, but they were 36 and 40% (ammonium nitrate and urea, respectively) higher than the control (Table 3).

Table 3 shows that not only were the average  $\text{N}_2\text{O}$  emissions similar, but the ranges of presented values were also similar. The accumulation for the period from the different doses of filter cake was on average 40% higher than the control.

The use of synthetic N-fertilizer during sugarcane planting generated an emission factor of 0.48%, while the filter cake application showed an average value three times lower (0.13%). The combination of organic and synthetic sources (FC + Ur), which is the most common condition, showed an emission factor of 0.21% (Table 3).

#### *N<sub>2</sub>O emissions from soil for ratoon sugarcane*

During this experimental period, the average temperature was 24.7  $\pm$  1.2°C, and 12 days with precipitation events were recorded, totaling 172.3 mm (Fig. 3i). The



**Fig. 3** N<sub>2</sub>O emissions ( $\mu\text{g m}^{-2} \text{h}^{-1}$ ) from Urea – Ur 80 ( $174 \text{ kg ha}^{-1} \approx 80 \text{ kg N ha}^{-1}$ ) and vinasse – Vi 150 ( $150 \text{ m}^3 \text{ ha}^{-1} \approx 195 \text{ kg N ha}^{-1}$ ) applied in the sugarcane ratoon (a). From different doses of the vinasse – Vi 150 and Vi 300 ( $300 \text{ m}^3 \text{ ha}^{-1} \approx 390 \text{ kg N ha}^{-1}$ ) (b). From different synthetic N-sources and doses: Ammonium nitrate – AN 80 ( $225 \text{ kg ha}^{-1} \approx 80 \text{ kg N ha}^{-1}$ ) and AN 120 ( $340 \text{ kg ha}^{-1} \approx 120 \text{ kg N ha}^{-1}$ ); and Urea – Ur 80 and Ur 120 ( $260 \text{ kg ha}^{-1} \approx 120 \text{ kg N ha}^{-1}$ ) (c). In the top figures (i and ii) represented environmental information during the experiment: (i) Precipitation (mm) and air temperature (°C); (ii) N<sub>2</sub>O atmospheric concentration (ppmv).

atmospheric N<sub>2</sub>O concentration was  $314.5 \pm 18.1 \text{ ppbv}$  (Fig. 3ii).

Figure 3a presents the daily N<sub>2</sub>O emissions from the treatment with the application of N-fertilizer urea (Ur 80 kg N ha<sup>-1</sup>) with vinasse (Vi 150 m<sup>3</sup> ha<sup>-1</sup>) compared with the control. N<sub>2</sub>O emissions from the combination of synthetic and organic N-sources (Ur + Vi) were higher than the control starting from the 2nd DAT and maintained a median level of  $188 \mu\text{g N}_2\text{O m}^{-2} \text{ h}^{-1}$ , while the control emission was  $41 \mu\text{g N}_2\text{O m}^{-2} \text{ h}^{-1}$ .

This means that the combination of organic and synthetic sources was 4.5 times higher than the control.

On the 5th DAT, N<sub>2</sub>O emissions from the Ur + Vi treatment were nine times higher than the control value. This high value of N<sub>2</sub>O emissions early in the experimental period was observed in all treatments in which synthetic (urea and ammonium nitrate) and organic (vinasse) sources were applied alone (Fig. 3b and c). The maximum values in the range of results clearly expresses the magnitude of N<sub>2</sub>O emissions using differ-

**Table 4**  $\text{N}_2\text{O}$  emissions (mean  $\pm$  standard deviation and range), cumulative  $\text{N-N}_2\text{O}$  emissions and  $\text{N}_2\text{O}$  emissions factor from  $\text{N}$ -sources (mineral and organic) applied in the sugarcane ratoon

Treatments (sources/levels)	$\text{N}_2\text{O}$ emissions		$\text{N}_2\text{O}$ emission factor	
	Mean $\pm$ SD*	Range†	Cumulative $\text{N-N}_2\text{O}$ ‡	$\text{N}$ -source/level§
	$\mu\text{g N}_2\text{O m}^{-2} \text{h}^{-1}$		$\text{mg N-N}_2\text{O m}^{-2}$	% N-applied
Control	41.9 $\pm$ 24.6	5.2–103.2	17.9 $\pm$ 1.0 <sup>B</sup>	
<i>Mineral source</i>				
Ammonium nitrate (80 kg $\text{N ha}^{-1}$ )	97.5 $\pm$ 77.5	5.0–478.3	39.1 $\pm$ 5.8 <sup>A</sup>	0.25 $\pm$ 0.07 <sup>ns</sup>
Ammonium nitrate (120 kg $\text{N ha}^{-1}$ )	119.2 $\pm$ 112.8	4.9–530.2	47.3 $\pm$ 8.6 <sup>A</sup>	0.23 $\pm$ 0.07
Urea (80 kg $\text{ha}^{-1}$ )	153.4 $\pm$ 145.0	2.5–998.4	61.8 $\pm$ 12.9 <sup>A</sup>	0.52 $\pm$ 0.15 <sup>ns</sup>
Urea (120 kg $\text{ha}^{-1}$ )	323.9 $\pm$ 329.7	3.8–2188.4	122.0 $\pm$ 28.1 <sup>A</sup>	0.68 $\pm$ 0.24 <sup>a</sup>
<i>Organic source</i>				
Vinasse (150 $\text{m}^3 \text{ha}^{-1}$ )	97.3 $\pm$ 77.7	3.8–312.1	38.6 $\pm$ 4.4 <sup>A</sup>	0.77 $\pm$ 0.16 <sup>ns</sup>
Vinasse (300 $\text{m}^3 \text{ha}^{-1}$ )	125.5 $\pm$ 170.8	3.5–1117.9	46.9 $\pm$ 19.9 <sup>A</sup>	0.54 $\pm$ 0.37
Urea (80 kg $\text{N ha}^{-1}$ ) + vinasse (150 $\text{m}^3 \text{ha}^{-1}$ )	212.6 $\pm$ 156.5	3.6–831.3	83.6 $\pm$ 21.3 <sup>A</sup>	0.59 $\pm$ 0.19

\*Mean  $\pm$  SD = standard deviation.

†Range = Minimum and maximum.

‡Cumulative  $\text{N}_2\text{O}$  = sampled time in the sugarcane ratoon = 30 days.

§ $\text{N}_2\text{O}$  emissions factor from each  $\text{N}$ -sources and level applied in the sugarcane ratoon.

¶ $\text{N}_2\text{O}$  emissions factor from  $\text{N}$ -source applied in the sugarcane ratoon. Different capital letter to the Cumulative  $\text{N-N}_2\text{O}$  ( $\text{N}$ -source and dose  $\times$  control) and small letter to the Emission Factor ( $\text{N}$ -source/level and  $\text{N}$ -source) with significant differ in Student-Newman-Keuls test ( $P < 0.05$ ); ns, not significant.

ent  $\text{N}$ -sources (synthetic and organic) applied on the surface (Table 4). The reduction in  $\text{N}_2\text{O}$  emissions between the 6th and 9th DAT can be related to consecutive high rainfall events ( $\sim 66 \text{ mm}$ ) that occurred during the sampling period. The same occurred at 12 (46 mm) and 19 DAT (42 mm), when precipitation events occurred the day before sampling. At the 10th DAT, a second increase in  $\text{N}_2\text{O}$  emissions was observed in the treatment with  $\text{N}$  application at a magnitude thirteen times higher than the control. Two other increases in emissions were observed: one at the 15th DAT, four times higher than the control, and another at the 21th DAT, three times higher than the control. After the 24th DAT,  $\text{N}_2\text{O}$  emissions from the treatment with  $\text{N}$  application dropped to levels similar to the control.

The vinasse applied to the ratoon produced cumulative emissions 110% and 160% (for 150 and 300  $\text{m}^3 \text{ha}^{-1}$ , respectively) higher than the control. Although  $\text{N}_2\text{O}$  emissions were higher as a result of the applied dose, there was no significant difference between them (Fig. 3b, Table 4).

The patterns of  $\text{N}_2\text{O}$  emissions from  $\text{N}$ -fertilizers (ammonium nitrate and urea) were quite different (Fig. 3c, Table 4). Ammonium nitrate had lower emissions than urea. The cumulative emissions for the different doses of ammonium nitrate showed differences of more than 115 and 160% (80 and 120 kg  $\text{N ha}^{-1}$ , respectively) compared with the control. However, the increase in the applied dose did not cause a proportional increase in  $\text{N}_2\text{O}$  emissions, and no showed significant difference.

$\text{N}_2\text{O}$  emissions derived from urea were high in both applied doses. The highest dose had an accumulated value about twice the lowest dose. Thus, the source urea emitted 240 and 570% (80 and 120 kg  $\text{N ha}^{-1}$ , respectively) more than the control.

By the sugarcane ratoon, the use of ammonium nitrate presented an emission factor of 0.24%, while the source urea generated an emission factor almost three times higher (0.68%). The vinasse application presented an emission factor close to that observed for urea (0.65%). The combination of synthetic and organic sources (Ur + Vi), which is the most normal field condition, presented an emission factor of 0.59% (Table 4).

## Discussion

### $\text{N}_2\text{O}$ emissions from soil and sugarcane straws

$\text{N}_2\text{O}$  emissions from different quantities of sugarcane straw did not cause measurable changes. For the conditions of this study, our goal was to determine whether during the course of decomposition of sugarcane straw, it presents itself as a source for soil  $\text{N}$ , which would contribute significantly to  $\text{N}_2\text{O}$  emissions, because from the first cut, the straw covering the soil remains throughout the conduction period after the ratoon regrowth, in a cycle of deposition and decomposition.

Carmo *et al.* (2012) found increased  $\text{N}_2\text{O}$  emissions with an increasing amount of sugarcane residues deposited on the ground; however, this study included appli-

cation of N-fertilizer and vinasse. Malhi & Lemke (2007) found no significant differences in N<sub>2</sub>O emissions due to the maintenance or removal of different crop residues. The authors also commented that due to the high variability in the results, it is difficult to determine the interaction between factors (e.g., soil water content and temperature) that may influence the N<sub>2</sub>O emissions, even with the application of the N-fertilizer. Wang *et al.* (2008) reported a reduction between 24 and 30% in N<sub>2</sub>O emissions due to removal of the sugarcane straw and attributed this finding to the importance of organic compounds in the regulation of N<sub>2</sub>O fluxes.

Our study evaluated only sugarcane straw as a variable, without introducing other N-sources. Sugarcane straw is a plant residue with a high C:N ratio (approximately 50–80). Moreover, sugarcane straw has a large amount of lignin and polyphenols that reduce the decomposition rate (Abiven *et al.*, 2005) and the availability of the minimal N (0.8%) is slow (Fortes *et al.*, 2012; Leal *et al.*, 2013). Huang *et al.* (2004) evaluated the influence of crop residue application with different C:N ratios on N<sub>2</sub>O emissions, including sugarcane straw. This study showed that increasing the C:N ratio of the plant residues incorporated into the soil promoted reduction of N<sub>2</sub>O emissions.

Currently, it is known that the amount of straw added by sugarcane is approximately 14.1 (7.4 to 24.3) Mg ha<sup>-1</sup> yr<sup>-1</sup>; however, there are different methods of managing the straw in the field (Leal *et al.*, 2013). In the central-southern region of Brazil, it is common to bunch the excess straw between the rows of crops because the producers claim that the high amount deposited along the cycles of regrowth promotes a shallower root system, which reduces the length of the crop cycle. The growth in the production of second generation ethanol can increase an industrial demand for excess straw remaining in field (Goldemberg, 2008).

It is worth mentioning that the N<sub>2</sub>O emission was approximately 12 µg N<sub>2</sub>O m<sup>-2</sup> h<sup>-1</sup>, a low value for agricultural soils. However, one must consider that the experiment was conducted at the beginning of a regrowth cycle, after exportation of nutrients by the crop (stalks) with no contribution of N-sources. In an experiment to assess N<sub>2</sub>O emissions from elephant grass cultivation for biomass production, Morais *et al.* (2013) found N<sub>2</sub>O emissions ranging from 0.1 to 10 µg m<sup>-2</sup> h<sup>-1</sup>, which increased due to soil management and N-fertilizer application.

Throughout the experimental period (~180 days), interventions related to sugarcane management in the area were made only to meet the goals of the experiment (i.e., herbicide application), without the normal crop management practices in the region (i.e., topdressing N-fertilizer or vinasse application), so as to priori-

tize only the N<sub>2</sub>O emissions from sugarcane straw. Additionally, one must note that the determinations were made between crop rows (Cai *et al.*, 2012). The ground in this area has a high clay content (> 60%), and the third regrowth is sugarcane, which means that it has high soil bulk density (maximum 1.52 g cm<sup>-3</sup>) as a result of machinery traffic and consequently lower pore space for N<sub>2</sub>O production and gas exchange.

#### *N<sub>2</sub>O emissions from organic and synthetic N-sources in sugarcane cultivation*

The N<sub>2</sub>O emissions in the control plots at sugarcane planting (~ 90 µg N<sub>2</sub>O m<sup>-2</sup> h<sup>-1</sup>) were considered to be high for agricultural soils. This result is most likely due to soil disruption in the reform area cultivation due to tillage operations (physical destruction of stumps, plowing, and opening and closing of the furrows) causing the incorporation of crop residues and aeration into the soil surface layer, which favors the priming effect and soil organic matter mineralization (La Scala *et al.*, 2006; Silva-Olaya *et al.*, 2013). These processes accelerate the labile-C availability in the soil, and thus, the set of microbiological processes enabled by the formation of anaerobic sites with high CO<sub>2</sub> concentrations together with the mineralization-N increase and N<sub>2</sub>O emissions from the soil (Rochette, 2008; Morais *et al.*, 2013). The same effect was observed by Morais *et al.* (2013), who observed higher N<sub>2</sub>O emissions after soil plowing of between 30 and 100 µg m<sup>-2</sup> h<sup>-1</sup> (maximum 450 µg m<sup>-2</sup> h<sup>-1</sup>) for elephant grass management. In the same period, the authors also found higher CO<sub>2</sub> and inorganic N concentrations.

The increase in N<sub>2</sub>O emissions at the beginning of the measurements in the experiment involving sugarcane planting that occurred for the trial with a combination of organic and synthetic sources (filter cake + urea) shows the effect of the interaction between the source; however, it has not been possible to establish with certainty whether this synergistic effect was due to a reaction between the filter cake and urea, or, more simplistically, as soil moisture was low, the filter cake moisture would have initiated the urea transformation, favoring N<sub>2</sub>O emissions.

In the regrowth crop, the N<sub>2</sub>O emissions control had a value (~ 40 µg N<sub>2</sub>O m<sup>-2</sup> h<sup>-1</sup>) that is considered normal for cultivated agricultural soils in tropical regions (Gomes *et al.*, 2009; Kachenchart *et al.*, 2012; Morais *et al.*, 2013). The increase in N<sub>2</sub>O emissions during the initial period of sampling is most likely associated with transformations in soil N because the application of the fertilizer+vinasse (Vi + Ur) promoted the same effect on the soil in sugarcane planting as filter cake plus urea. The urea and vinasse provided N-reactive and a faster increase in soil labile-C and humidity by vinasse.

Ball *et al.* (1999) showed that high doses of N-fertilizer promoted changes in  $N_2O$  emissions for a period of up to six weeks. Morais *et al.* (2013) showed that 80% of the  $N_2O$  emissions due to N-fertilizer occurred in the 60 days after the N-fertilizer was applied.

$N_2O$  emissions from vinasse were high when compared to the control or to other sources containing nitrogen (e.g. ammonium nitrate). This was most likely due to the available N form in this by-product in addition to the high organic content readily available to microorganisms in the soil, in which the higher microbial activity produced anaerobic sites with high  $N_2O$  production.

During the experiments, the  $N_2O$  emissions were largely influenced by precipitation, both the lack and excess thereof. There was no irrigation during planting; thus, the lack of moisture hindered the availability of N-fertilizer. In low soil moisture conditions, there was little development of the microbiota responsible for the N transformation (immobilization, mineralization and denitrification). The onset of precipitation and the resulting increased soil moisture promoted the development of microorganisms, increasing the availability of the reactive forms of soil N.

Regardless of the situation (planting or regrowth), the entry of water into the porous soil system promoted the 'expulsion' of gases contained in the pore spaces (physical movement) generated by soil disturbances at planting and in the macro- and mesopores formed between the particles and aggregates in the soil of regrowth.

Excessive rainfall on consecutive days, even in tropical soils with good drainage, causes most of the pore space to temporarily fill with water (~100% saturation of WFPS), thereby decreasing microbial activity and reducing greenhouse gas emissions (Bouwman, 1998). Hence, at higher WFPS (> 80%), the major product of the denitrifying community is  $N_2$  (complete denitrification) (Davidson *et al.*, 2000). Leaching may also occur because  $N_2O$  has a high solubility in water (Heincke & Kaupenjohann, 1999).

#### *Emissions factor from N applied in sugarcane cultivation*

For the synthetic N-fertilizer applied in the sugarcane, the overall EF was  $0.46 \pm 0.33\%$ . The ammonium nitrate has a lower emission factor than the urea. This behavior was most evident during the sugarcane regrowth and surface N application. This difference in  $N_2O$  emissions between sources is related to changes occurring in the soil related to the transformation of each source especially changes in soil pH. When applied, the breakdown of urea by the enzyme urease leads to a rapid pH increase (Lara-Cabeza & Souza, 2008); this brief increase in the soil pH is sufficient to promote the conversion of ammonium to nitrate (Wickramasinghe *et al.*, 1985) and,

consequently, start the denitrification. However, the conversion of ammonium to nitrate promotes soil acidification, which hinders nitrification (Page *et al.*, 2002), also affecting the denitrifying community (Daum & Schenk, 1998).

Carmo *et al.* (2012) calculated EF for N-fertilizer in sugarcane as 1.1% for N applied at planting as urea at the same dose as in this study ( $60 \text{ kg N ha}^{-1}$ ) and, for the N applied to the topdressing, the EF was 0.76% for a dose of  $120 \text{ kg N ha}^{-1}$ , a factor very similar to that determined in our experiment (0.83%). Lisboa *et al.* (2011) presented an emission factor of 3.8% for N-fertilizer in calculating the carbon footprint of Brazilian ethanol, based mostly on literature review. Morais *et al.* (2013) evaluated urea application in elephant grass, which has a similar management as sugarcane involving harvest and regrowth cycles. In this situation, the authors found an EF of 0.51% after three cycles.

The EF for filter cake and vinasse did not show the expected behavior, or, the increasing the dose did not increase  $N_2O$  emissions linearly. Thus, the highest dose showed a smaller EF than the lowest dose, this occurs because the EF calculation is based on the amount of N applied.

The filter cake applied at sugarcane planting had a lower EF relative to other synthetic and organic sources, even with similar cumulative emissions to synthetic sources used in planting (0.13%). This occurred because even with a low N content (~ 2%), the amount of filter cake applied at planting (25 and  $40 \text{ Mg ha}^{-1}$ ) was high. However, the amount of N applied was not available to denitrifying organisms in the soil, and when the EFs were calculated, the values were low.

The vinasse applied generated a high emission factor (0.65%) analogous to that observed for urea application. N contained in the vinasse, together with the application form of fertirrigation, favored  $N_2O$  production in the soil. In a study of short duration with application of  $200 \text{ m}^{-3} \text{ ha}^{-1}$  of vinasse, Oliveira *et al.* (2013) calculated an EF of 0.44%.

The EF calculated from the sources combination filter cake + urea at planting and urea + vinasse in covering regrowth did not produce values representing a summation of sources when individually evaluated, but rather demonstrated an interaction between sources leading to a specific emissions factor.

Carmo *et al.* (2012) evaluated emissions from mixing organic and synthetic sources for the supply of N to the crop of sugarcane planting and regrowth. The specific EF were much higher for the combination planting (filter cake + urea); EF was 1.1% vs. the 0.21% observed in our study. In the regrowth crop period (urea + vinasse), the difference was even greater, with an EF of the 3.0%, vs. 0.59% in our study.

Schils *et al.* (2008) studied the effect of the combination of fertilizer with an organic source (liquid slurry) in ryegrass in the Netherlands and found an EF between 0.13 and 0.17% in the application of calcium ammonium nitrate, an EF of 0.12% in the application of liquid slurry, and values of 0.35 to 0.39% when a mix of fertilizer and slurry was applied. Dambreville *et al.* (2008), in a maize crop in the Brittany region (France), found an EF between 0.01 and 0.23% from liquid pig slurry application alone and of 0.01% for ammonium nitrate.

In the literature, N<sub>2</sub>O emission factors from N-fertilizer application in sugarcane cultivation mostly come from studies conducted in Australia, with values ranging from 1.0 to 21% (Weier, 1999; Galbally *et al.*, 2005; Wang *et al.*, 2008; Allen *et al.*, 2010; Denmead *et al.*, 2010). In terms of commercial sugarcane cultivation in Australia, the nitrogen cycle peculiarities, which reflect on the direct and indirect N<sub>2</sub>O emissions and consequently on the EFs, are related especially to the use of irrigation, drainage and presence of the seasonal fluctuations in the groundwater levels that directly influence soil oxygenation (and gas diffusion) against water-filled pore space (WFPS%). Abbasi & Adams (2000) found that for the same amount of N-fertilizer applied, the EF responded differently to increased WFPS%, with EFs of 0.15, 0.40 and 2.8%, respectively, for 63, 71 and 84% WFPS% values.

In the central-southern region of Brazil, where more than 85% of the Brazilian sugarcane is cultivated (UNICA, 2014), the management does not include the irrigation, or the soil does not report the occurrence of this specific phenomenon of poor drainage or seasonal fluctuations in the groundwater.

As shown by Smeets *et al.* (2009), our results indicate that the use of the default value of IPCC (2006) for the conditions of sugarcane cultivation in the central-southern region of Brazil overestimates the contribution of direct N<sub>2</sub>O emissions from N-fertilizer (synthetic or organic). Our results not only supply measured values for EFs from N-applied but also help to reduce the variance of the values. Combining these results with other values for this region (Carmo *et al.*, 2012; Morais *et al.*, 2013; Oliveira *et al.*, 2013) should facilitate eliminating the uncertainties associated with these emission sources.

Other studies showed that sugarcane usually does not respond to increasing rates of N (Reis Jr *et al.*, 2000; Franco *et al.*, 2011). This is due to the contribution of biological nitrogen fixation (BNF) by groups of diazotrophic endophytes responsible for 25–60% of the N supply to the crop (Reis Junior *et al.*, 2000; Boddey *et al.*, 2001; Resende *et al.*, 2003). Thus, the N management in sugarcane varies considerably with respect to sources and doses. Evaluating different variety behavior to reduce the external N supply (synthetic and/or organic) and

the possibility of finding the best economic level and not 'standardized' quantities of fertilizers based on expected yields can effectively reduce N<sub>2</sub>O emissions.

In the case of greenhouse gas inventories from biofuels, we suggest a top-down approach using EFs evaluated for specific situations (Tier 2 – IPCC, 2006) in addition to extrapolating N<sub>2</sub>O emission values to regional scales to include more sophisticated mathematical models with a bottom-up approach (Del Grosso *et al.*, 2008) that seek a better understanding of the N cycle, including the influence of other factors (e.g., water, temperature, aeration, labile-C availability and plant N demand), rather than only the amount of N-applied and amount of N<sub>2</sub>O emitted.

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