



XXXI B-MRS Meeting 2024
September 29th to October 3rd

PROCEEDINGS

Sociedade Brasileira de Pesquisa em Materiais

Proceedings of the XXII B-MRS Meeting

Santos, SP 2024

Copyright © 2024 para os autores

Conteúdo, revisão textual e gramatical: Resposanbilidade dos respectivos autores.

Todos os direitos reservados 2024

A reprodução não autorizada desta publicação, no todo ou em parte,
constitui violação de direitos autorais (Lei 9.610/98).

ISBN: 978-85-63273-63-5

Hydrogel beads formed from the sisal lignocellulosic fibers' deconstruction: applications in the purification of water polluted with heavy metals

Giani de Vargas Brião¹, Bianca Groner Queiroz¹, Luiz Antonio Ramos¹, Derval dos Santos Rosa², Elisabete Frollini¹

¹Instituto de Química de São Carlos - Universidade de São Paulo (*Departamento de Físico-Química*) , ²Universidade Federal do ABC (*Centro de Engenharia, Modelagem e Ciências Sociais Aplicadas*)

e-mail: giani@usp.br

Hydrogels made from sisal biomass, which consists of cellulose, hemicelluloses, and lignin, were produced, and assessed for their ability to sorb toxic metals. The process of creating hydrogel beads from sisal fibers involved the following steps: (i) fibers deconstruction using bis(ethylenediamine)copper(II) hydroxide solution (CUEN, 1M); (ii) dropping the resulting gel in calcium chloride solution at 80°C (CaCl₂-80) and 120°C (CaCl₂-120); (iii) crosslinking with citric acid (CA) at 80°C (CaCl₂-CA-80) and 120°C (CaCl₂-CA-120). The beads were characterized by sorption capacity, scanning electron microscopy (SEM), Fourier transform infrared spectroscopy (FTIR), and compressive properties. The beads' sorption capacity (q/mmol/g), total and for each metal, was obtained by putting them in contact with simulated wastewater containing several toxic metals for 24 h under agitation and analyzed via ion chromatography. The FTIR spectra of CaCl₂-CA-80 and CaCl₂-CA-120 indicated the esterification of the hydroxyl groups (band at 1724 cm⁻¹). CA crosslinked beads presented smoother surfaces than their counterparts, CaCl₂ crosslinked ones. The crosslinking process with CA was expected to increase Young's modulus of the spheres. However, the reaction at 80°C barely affected the beads' modulus, while the one at 120°C resulted in beads with a lower modulus than those not crosslinked. These findings suggest that CA caused partial hydrolysis of the bead components. On the contrary, CA crosslinking positively impacted the sorption capacity of Cu(II) and Cr(VI), while the sorption of Ni(II), Zn(II), Mn(II) and Cd(II) was not significantly affected. This is ongoing research, and the initial outcomes indicated a potential application of hydrogels derived from the sisal's fibers deconstruction for removing heavy metals from wastewater.

Funding: FAPESP, #2023/07447-2.