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Influence of antimony contant as a dopant in IrO_x/ATO catalysts for oxygen evolution reaction

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Proton exchange membrane water electrolyzer (PEMWE) stands out as a promising method for its production and storage of hydrogen. Iridium-based catalysts are the most promissing for oxygen evolution reaction (OER) [1]. However, due to the high cost and scarcity of this noble metal, it is necessary to optimize its use in PEM. This work investigated the activity of amorphous iridium oxide (IrOx) supported on antimony-doped tin oxide (ATO) with different amounts of Sb (3-10%) as a catalyst for OER. The electrocatalysts were characterized by XRD, EDS and TEM. Electrochemical characterization and OER were evaluated in a three-electrode cell with 0.1 mol L^{-1} H₂SO₄ electrolyte. The (Fig. 1) shows that despite the antimony amount, the IrOx/ATO catalyst activity was lower than of IrOx, demonstrating that changes in antimony content do not greatly affect electrocatalyst activity.

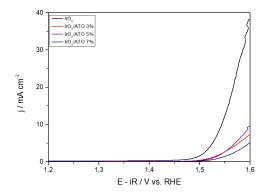


Figure 1: iR-corrected OER curve showing the comparative activity of IrOx and IrOx/ATO catalysts, in $0.1 \text{ mol } L^{-1} \text{ H}_2\text{SO}_4$ electrolyte.

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References:

[1] L. Moriau, M. Smiljanić et al., ChemCatChem, vol. 14, no. 20, 2022.