

Potential of storing gas with high CO₂ content in salt caverns built in ultra-deep water in Brazil

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Abstract: Salt caverns have been identified as one of the best options for the underground storage of gases due to salt rock's excellent sealing capabilities and interesting mechanical properties, such as self-healing when damaged or cracked. It is feasible to build salt caverns in the Brazilian pre-salt ultra-deep water environment for gas storage. However, the peculiar geology of the Brazilian province considered here is characterized by the stratification of thick layers of halite with intercalations of carnallite and tachyhydrite salt rock, whose creep strain rate is almost two orders of magnitude higher than halite's creep strain rate under the same conditions of temperature and pressure. Computational mechanics is being used for the design of offshore salt caverns opened by dissolution mining for the storage of natural gas. The challenge presented in this paper requires the storage of natural gas with high CO₂ content offshore in ultra-deep water (2140 m) in salt caverns. If the economics proves feasible, this offshore gas storage station will be the first of its kind in the world. A technical feasibility rock mechanics study of giant salt caverns, 450 m high by 150 m in diameter, has shown that one cavern can store 4 billion Sm³ or 7.2 million tons of CO₂. The salt dome studied can accommodate the construction of 15 caverns, thus providing the confinement of approximately 108 million tons of gas. © 2018 Society of Chemical Industry and John Wiley & Sons, Ltd.

Keywords: salt caverns; gas storage; rock salt; creep strain rate; stability analysis

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Introduction

Governments, corporations, and other organizations increasingly recognize the importance of stabilizing the atmospheric concentration of greenhouse gases (GHG). Climatologists agree that increasing concentrations of GHG contribute to global climate change, and CO₂ is reported to dominate these emissions. The process of carbon capture and storage in geologic media has been identified as an important means of reducing anthropogenic GHG emissions.¹ Noting various ways to sequester CO₂ in geological media, including utilization in enhanced oil recovery (EOR)² operations, disposal in disused oil and gas reservoirs, replacement of methane in coalbeds, injection in deep saline aquifers, and storage in salt caverns, Bachu (2000) suggested several criteria to be considered when evaluating the potential of a sedimentary basin for CO₂ sequestration. These criteria include its tectonic setting and geology, the basin geothermal regime, the hydrodynamic regime of formation waters, the hydrocarbon potential and basin maturity, economic aspects relating to access and infrastructure, and socio-political conditions.²

Salt caverns have been identified as one of the best options for the underground storage of gas, the main advantages including the following:

- Low permeability. The permeability of salt rock is about 10^{-21} to 10^{-24} m² (1.01325×10^{-6} – 1.01325×10^{-9} mD); thus, it can provide excellent sealing for the salt cavern.
- Good mechanical properties. The damage self-recovery capability of salt rock can ensure the safety of a salt cavern even with frequent changes of gas pressure.
- Solution in water. Rock salt is easily dissolved in water, which facilitates the construction and shape control of the salt cavern.³
- Abundant resources. Salt rock is a very rich mineral resource with wide distribution and large reserves. Sites for the construction of salt caverns for gas storage near where they are required can therefore be found easily.⁴ Salt formations can be found in almost every part of the world. With some exceptions around the Pacific Rim, salt caverns can be constructed in naturally occurring thick salt domes, deep underground. Salt caverns can be formed through a leaching process by pumping fresh

water or sea water to dissolve the salt and removing the resulting brine via a single well, which then serves for both gas injection and withdrawal.⁵

Wellbore integrity has been identified in many studies as a key risk factor affecting the viability of geologic carbon capture and storage (CCS) projects.^{6,7} Wellbore integrity depends strongly upon the integrity of the interfacial bonding of the cement to the formation and the casing.^{6,8,9} Loss of well integrity leads to several consequences, including negative financial consequences, and significant environmental impacts, such as groundwater contamination, gas leakage, and fluid spills and seepage among others.^{7,10} Carbon capture and storage in geological media, also known as CO₂ geosequestration, has been identified as a viable climate-change mitigation technology.¹ For CCS to be successful, three fundamental conditions have to be met by the storage medium. First, the medium must have the capacity to store a large volume of CO₂; second, the injectibility of CO₂ into the medium must be ensured; and finally, the medium must be able to confine the stored CO₂ for a very long time without leakage.¹¹

In 2006, through the pioneer well TUPI (RJS-628), one of the largest oil provinces in the world for pre-salt reserves was discovered, stretching from the north to the south of the Brazilian continental front, between 200 km and 300 km away from the coast.¹² It consists of carbonate reservoirs underlying approximately 2000 m of stratified salt rock. Many provinces producing oil and natural gas globally are associated with reservoirs underlying thick layers of salt, consisting largely of halite (NaCl), whose mechanical resistance favors the drilling of wells. The peculiar geology of the Brazilian province is characterized by the stratification of thick layers of halite with intercalations of carnallite and tachyhydrite salt rock, whose creep strain rate is approximately two orders of magnitude higher than halite's rate for the same temperature and pressure state variables.¹² The stratified halite layers are perforated by giant pure halite diapirs coming from the bottom due to the creep movement of older geological layers of halite. These diapirs are the best choice for the construction of salt caverns.

The petroleum in the pre-salt reservoirs in Brazil usually possesses a very high gas–oil ratio (GOR) with a high carbon dioxide content. The high GOR of the pre-salt reservoir fluid with high CO₂ content poses another great challenge to be overcome for the

continuity of hydrocarbon production from these reservoirs.

Laboratory analyses have shown that CO₂ found in the pre-salt reservoirs in Brazil are of mantle origin, migrating to the pre-salt reservoirs through the intense presence of geological faults between crystalline carbonate reservoirs.¹² Due to the pressure and temperature state variables in the reservoirs, CO₂ is usually in a liquid or supercritical state and mixed with hydrocarbon inside the reservoirs. However, when it is produced at the surface, it gasifies and mixes with the other monetizable gases, such as methane (CH₄), and ethane (C₂H₆).¹²

Some of these gasses are treated and separated on the platform using membrane separation technology, reducing the CO₂ content to 3%, and it is possible to transfer a portion of the natural gas to shore through carbon steel pipelines. The remainder, consisting mainly of CO₂ gas, is reinjected into the reservoir. At the beginning of the life of the fields, the reinjection of this gas was used to support EOR in a process called water alternated gas (WAG). However, as the same molecule of CO₂ recycles several times within the drainage radius of the wells, the CO₂ content starts to increase significantly, making its treatment in the platform infeasible, which may force the closure of some production wells.

From the inception of production of pre-salt reservoirs, in the field development plan, there was a hypothesis that the CO₂ gas stream could be reinjected into nearby saline aquifers, which could only be selected below the pre-salt reservoirs. This was due to the small layer thickness of the sediment above the salt layer (between 500 m and 800 m) which cannot withstand the envisaged volume of CO₂ at the required injection pressure and temperature.¹²

However, a saline aquifer below the pre-salt is intercepted by a large number of geological faults, which naturally feed, without anthropogenic action, the pre-salt reservoirs with mantle CO₂. The injection of CO₂ under high pressures into these reservoirs possessing non-sealing geological faults or even with some sealing faults in place would only increase their permeability. This process would create a vicious cycle, increasing the level of reservoir contamination and adversely affecting field production over time.

Thus, the high concentration of CO₂ in the pre-salt reservoirs presents another adverse technological challenge requiring the separation and safe storage of CO₂. Even implementing conventional membrane

separation technology for a CO₂/ natural gas mixture would not be technically and economically feasible for reservoir fluids possessing high GOR with very high CO₂ content.

With these prevailing limiting conditions of pre-salt reservoirs, salt rock has become a strategic geomaterial for the process of confining gas streams with high CO₂ content because such gas can be injected and contained in salt caverns rather than being reinjected into reservoirs.

The innovative technology presented in this paper is peculiar to the environment in which the research project is being developed with very unique characteristics. It is unprecedented worldwide and requires a deep multidisciplinary knowledge of offshore oil engineering in ultra-deep waters, drilling of wells through very thick layers of salt rock, geomechanical design of salt caverns offshore in ultra-deep water, cavern construction technologies for offshore dissolution, fluid mechanics, gas thermodynamics, and CCS in salt caverns.

To date, there are no natural gas or other hydrocarbon storage systems in offshore salt caverns. Only the Gateway Project in the Irish Sea is upcoming; it involves a project to store natural gas in offshore salt caverns in shallow water 25 meters deep, but it has not been built yet.¹³

Simulation results obtained shows the technical feasibility of huge storage volumes of natural gas and CO₂ in giant salt caverns offshore. For the case study presented in this paper, using giant salt caverns, 450 m high by 150 m in diameter, one cavern can store 4 billion Sm³ or 7.2 million tons of CO₂. The salt dome studied can accommodate the construction of 15 caverns, thus providing the confinement of approximately 108 million tons of CO₂.

Technical feasibility study of salt cavern construction for carbon dioxide storage in ultra-deep water offshore

Basic simulation data

In the process of selecting potential areas for the development of salt caverns by solution mining, salt diapirs are selected to minimize the presence of interbedded non-soluble rock layers, such as shale and anhydrite, and to avoid the more soluble salts, namely carnallite and tachyhydrite, as much as possible. Some



- Specific weight of the sedimentary rock $\rightarrow 22.56 \text{ kN/m}^3$.

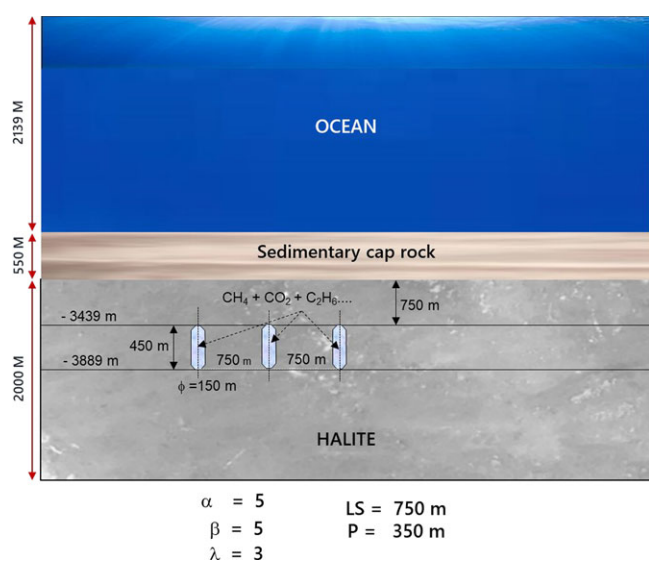


Figure 3. Basic geometric data used for building the structural geomechanical model.

- Specific weight of the rock salt $\rightarrow 21 \text{ kN/m}^3$.
- Slab protection of rock salt at the top of the cavern $\rightarrow \alpha \times \varnothing = 750 \text{ m}$.
- Minimum distance between caverns to avoid interference $\rightarrow \beta \times \varnothing$ (between axis) = 750 m.
- Temperature of the sea bed = 4°C .
- Geothermal gradient in sedimentary rock = $30^\circ\text{C} / 1000 \text{ m}$.
- Geothermal gradient in rock salt = $12^\circ\text{C} / 1000 \text{ m}$.

- Slenderness ratio of the cavern $\rightarrow \lambda = h/\phi = 3$.
- Shape $\rightarrow$ cylindrical.
- Height and diameter of cavern $\rightarrow$ 450 m and 150 m respectively.
- Initial state of stress at the top of the cavern $\rightarrow$

$$\sigma_0 = 2,139 * 10 + 550 * 22.56 + 750 * 21 = 21\,390 + 12\,408 + 15\,750 = 49\,548 \text{ kPa}.$$
- Maximum internal pressure of CO₂ inside the cavern $\rightarrow$ 90% of the effective initial stress.
- At the top of the cavern ($P_{\max}$) = 44 593.20 kPa.
- The geometric volume of the cavern is indicated in Fig. 4.

The creep strain rate of halite is activated by the temperature. The rock salt layer is divided into small layers to take into consideration the variation of the temperature with depth. Figure 4 shows the structural geomechanical model used in the simulation. COVES 2 is the simulator used for the structural geomechanical modeling.¹⁸

An explanation of the physical-mathematical formulation of the computer code, COVES 2, and the rock mechanics properties used in the geomechanical simulation has already been presented by one of the authors in another paper titled: ‘UGS in giant offshore salt caverns to substitute the actual Brazilian NG storage in LNG vessels.’¹⁵ As a recent development, the COVES solver was customized in GiD software.¹⁹ GiD is an adaptive pre- and post-processor for numerical

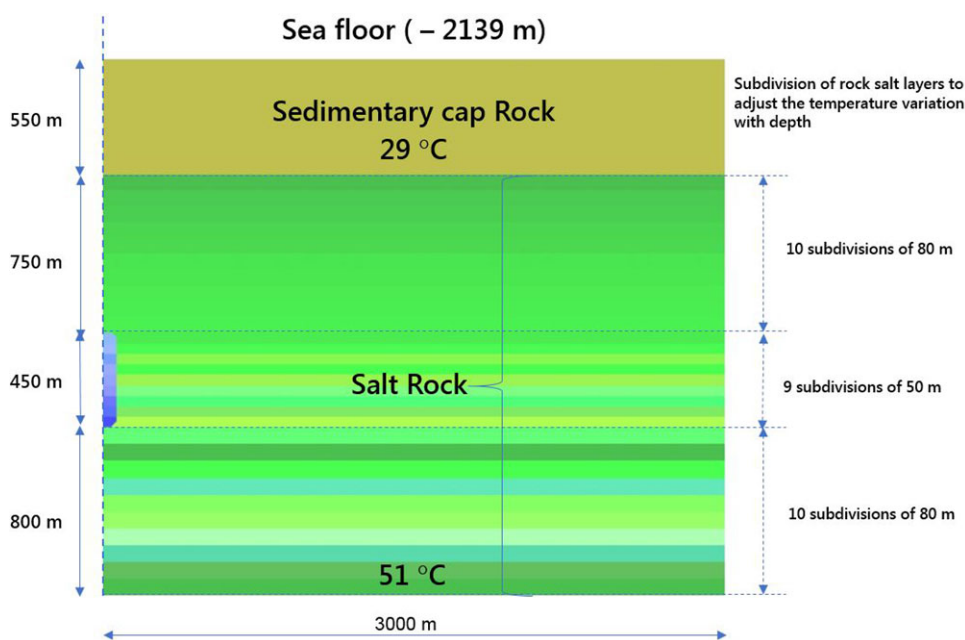


Figure 4. Geomechanical structural model of a CCS salt cavern in ultra-deep water.

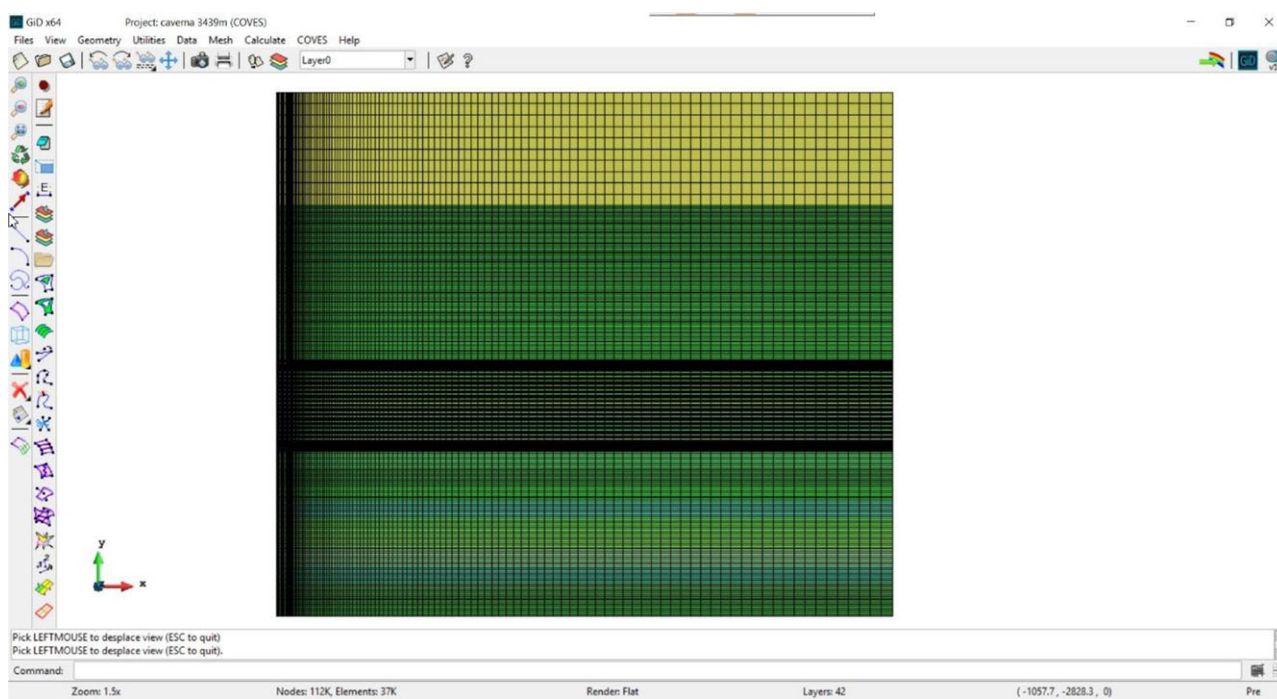


Figure 5. Finite-element mesh generated on the structural geomechanical model. (Solver COVES 2¹⁸ – pre-post-processor GiD.¹⁹)

simulations in science and engineering, where it is possible to implement conditions, material, and general data for specific purpose. All attributes of COVES were customized in menus, and the input data of COVES is written in its format using the tcl program language. The analysis results, which include many time steps and mesh modifications during analysis, are exported to GiD file format by an external developed software. Using this approach, COVES uses almost all the capability of the pre- and post-processor. For example, in the pre-processor, it is possible to create complex geometries with a CAD system, meshes, and import / export geometries. In the post-processor, it is possible to analyze results with fill and contour lines, vector plots, isosurfaces, animation, and so forth. In summary, the COVES solver is integrated in a complete environment of finite-element (FE) analysis.

An FE mesh is generated on the structural geomechanical model as depicted in Fig. 5. This is an FE axisymmetric model with 37 163 quadratic isoparametric elements of eight nodes and 112 352 nodal points.

Simulation results

COVES 2 was used to simulate the excavation of the cavern in nine steps of 50 m each, from the bottom to

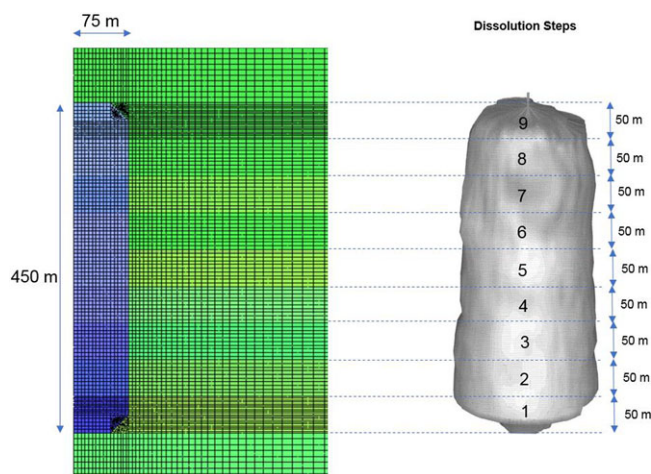


Figure 6. The 50 m excavation steps of the model.

the top of the cavern. This approach is like the technique employed in the stability analysis of salt caverns for brine production in Salgema Mineração.²⁰ Coincidentally, the steps of 50 m are directly related to the interval time between inspections by scan sonar 3D. Figure 6 shows the 9 excavation steps of 50 m each.

For efficient simulation of the stepwise cavern excavation, it is necessary to consider the dissolution time for each step of 50 m. From the alternatives studied for the construction of the cavern it was

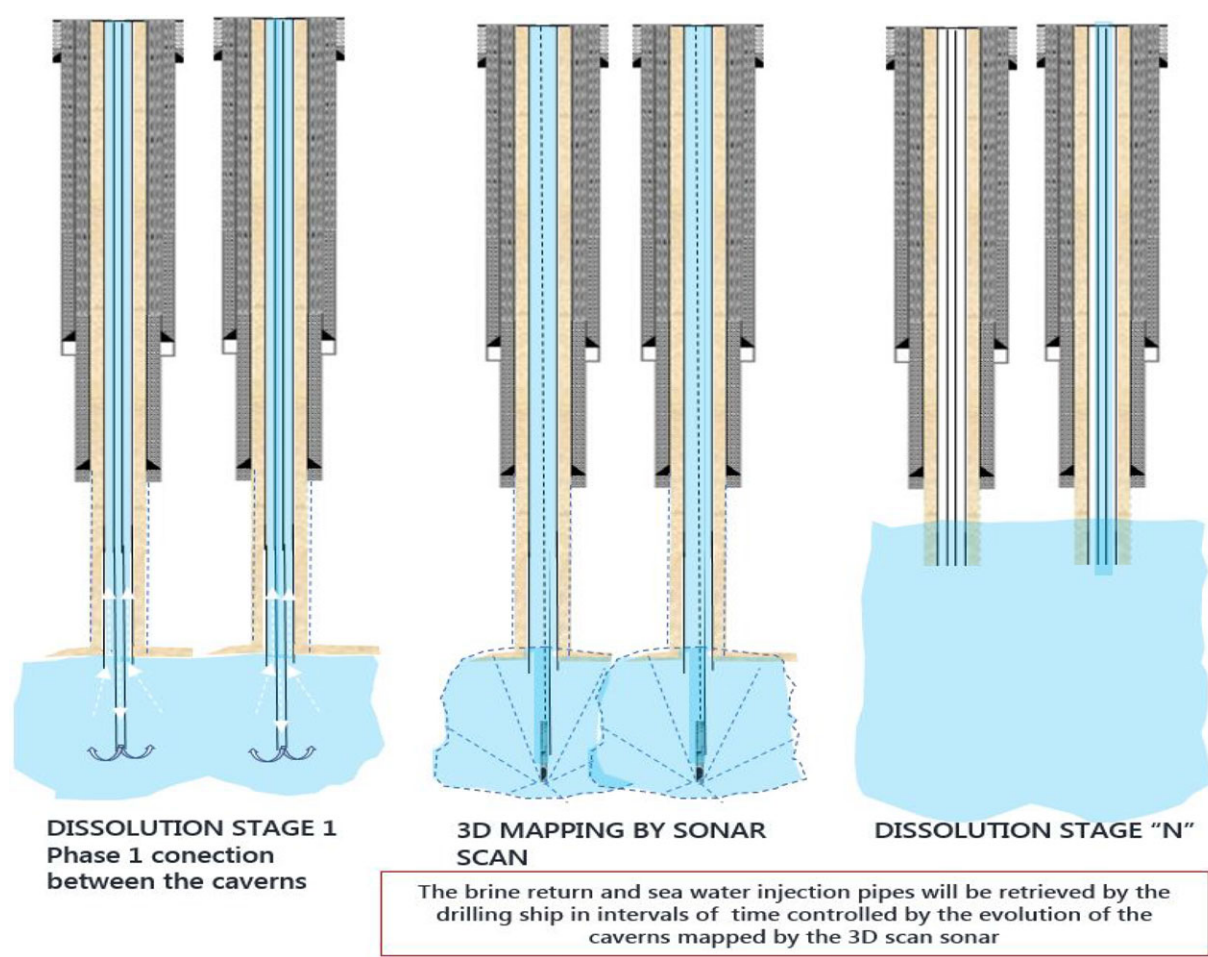


Figure 7. Dissolution process of the giant salt cavern.

assumed that two wells would be used for the dissolution with an injection flow rate of 140 000 barrels of sea water per day. The time to excavate each 50 m step would be approximately 15 weeks. Every 15 weeks the cavern would be monitored by scan sonar to control the diameter and shape. In case of any deviation from the predicted size and shape, in practice the stability of the cavern would be reanalyzed by structural geomechanical simulation. The SALGAS computer code was used to predict the dissolution time.²¹

Figure 7 shows how the dissolution of the giant cavern should be done. Two wells are drilled close to each other, and two caverns are constructed utilizing both wells. After some time the individual caverns intercept each other, becoming one giant cavern in the process of dissolution. During the dissolution process, the caverns are monitored by scan sonar 3D in each cavern.

COVES 2 simulates the excavation of the cavern step by step on a non-linear viscoelastic basis with time, excavating the elements included in each 50 m section, keeping the global structural equilibrium of the full model.

Figure 8 shows the positions of the nodal points with a continuous output from the simulation of the radial displacements to explain the effect of the excavation steps on the perimeter closure of the cavern. Figure 9 shows the evolution with time of the radial displacement of the cavern perimeter at the points shown in Fig. 8.

Each curve starts at a different time when the excavation reaches that position. The maximum closure of the cavern is about 1 m at the bottom of the cavern. The total time required for the dissolution of the cavern is 135 weeks, or approximately 2.5 years.

The effect of each excavation or dissolution step can be felt in the perimeter of the cavern, increasing the

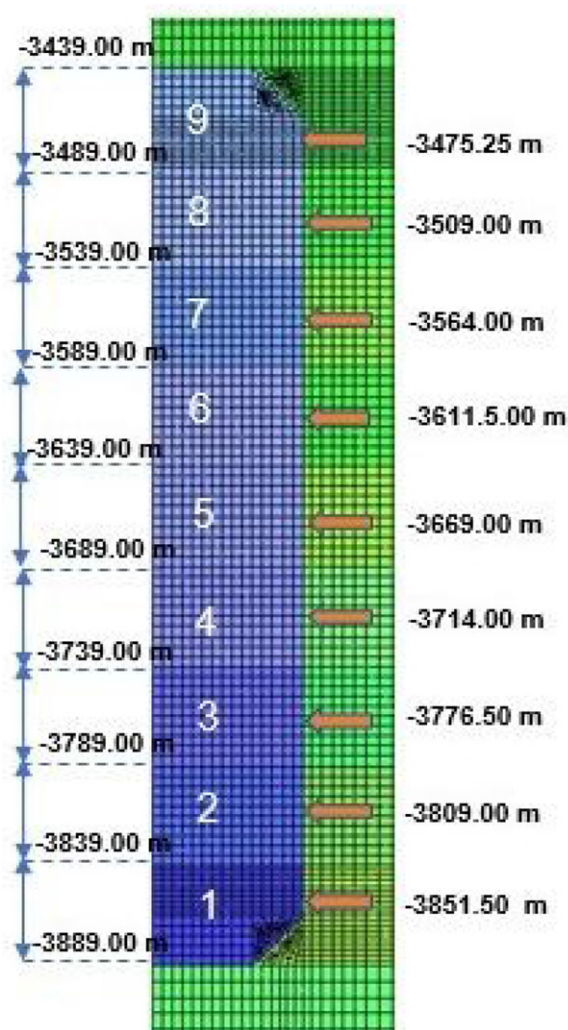


Figure 8. Dissolution stages and nodal points with continuous output of the radial displacements at the perimeter of the cavern.

closure velocity, as indicated in Fig. 9. The perimeter of the cavern is subjected to a differential stress of only 10% of the lithostatic stress (about 50 bar). At the same time the salt rock halite is quite cold at the position of the cavern, around 44°C, which justifies the small closure obtained by the simulation. A differential stress of 100 bar to approximately 150 bar is present at the openings of the underground potash mine at the state of Sergipe.^{4,18,22} A comparison between the closures measurements made in the mine, taking into consideration the volume of excavation and the differential stresses applied, validates the results obtained from the computer simulation of the cavern.

Figure 10 shows the isosurfaces of the radial displacements after 135 weeks, immediately after the

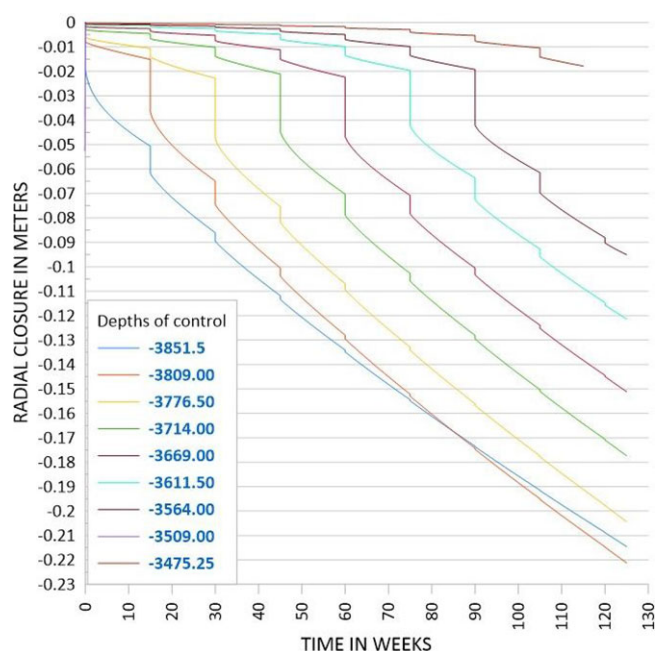


Figure 9. Radial displacement in various positions of the vertical cavern perimeter evolution with time. (Solver COVES 2¹⁸ – pre- and post-processor GiD.¹⁹)

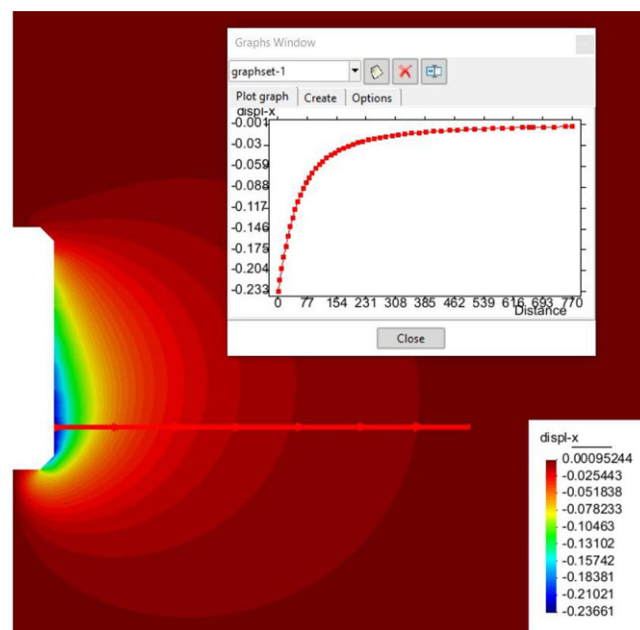


Figure 10. Isosurfaces of radial displacements at $t = 135$ weeks. (Solver COVES 2¹⁸ – pre and post-processor GiD.¹⁹)

dissolution of the cavern is completed. The isosurfaces and the decay of the displacement as a function of the point distance to the excavation perimeter is a perfect representation of the physical phenomenon of an underground excavation that is subject to creep.

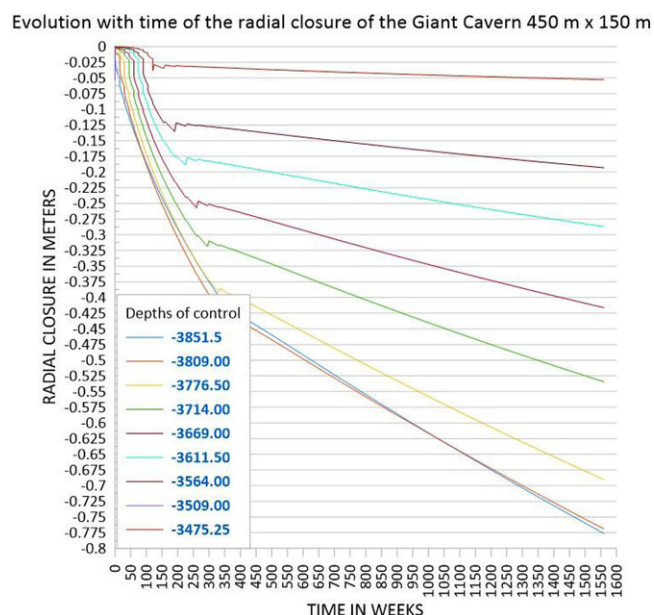


Figure 11. Radial displacement evolution with time showing the reaction of the cavern as gas is injected from the top. (Solver COVES 2¹⁸ – pre- and post-processor GiD.¹⁹)

Gas with high CO₂ content is injected after 125 days from the top to the bottom of the cavern, replacing the brine. Gas produced in the Floating Production Storage and Offloadings (FPSOs) that is not reinjected into the reservoir is separated by the membrane separation equipment at the process plant. The cleaned gas, with the maximum grade of 3% of CO₂, is compressed and transported through a gas pipeline to shore. The remaining gas, now with a high CO₂ content, will be reinjected into the reservoir for EOR, and part of it will be injected in the salt caverns. It is inferred that it will take 30 weeks to fill each 50 m section of the cavern from top to bottom. Figure 11 presents the reaction of the cavern as the gas is injected from the top downwards, replacing the brine.

For the geomechanical simulation, it was assumed that 270 weeks would be required to completely fill the cavern and replace the brine. After 395 weeks, the cavern is conditioned for abandonment. It was also assumed for the geomechanical simulation that 52 weeks would be required to close the cavern with all the monitoring instrumentation in place. At this point, the closure rate of the cavern is reduced with increasing injected gas pressure. At the critical state, the creep closure deformation of the cavern with time is stabilized. It is interesting to note that during the injection time, the nodal points used to monitor the

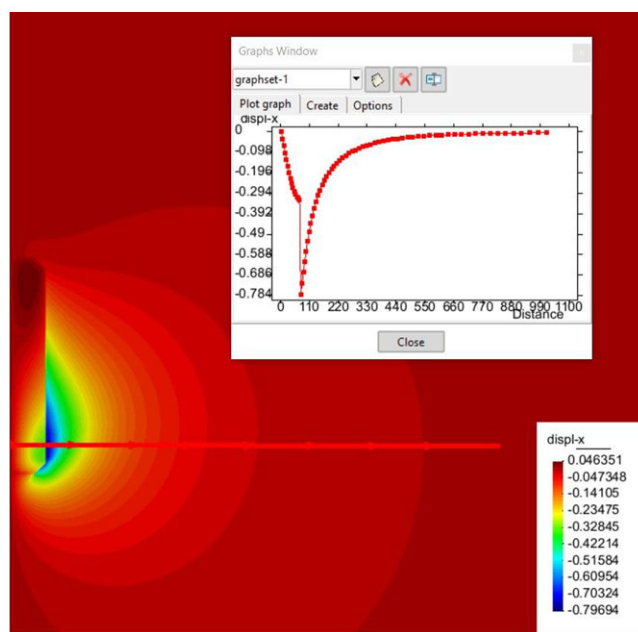


Figure 12. Isosurfaces of the radial displacements during the abandonment of the cavern at $t = 30$ years. (Solver COVES 2¹⁸ – pre- and post-processor GiD.¹⁹)

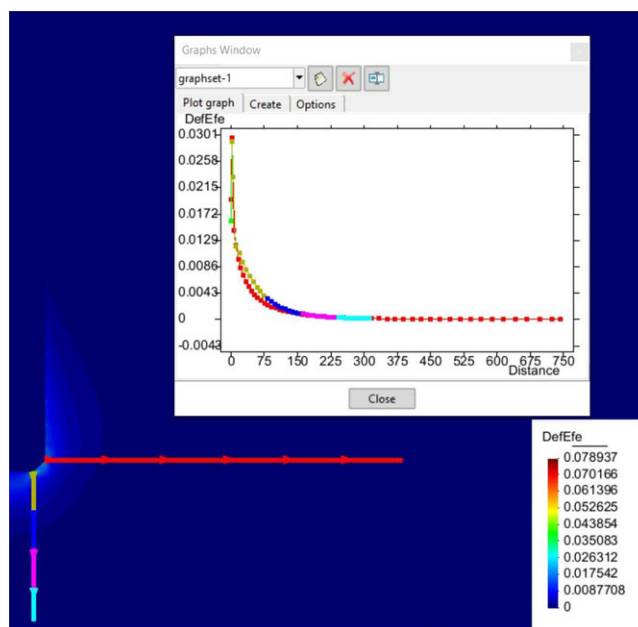


Figure 13. Isosurface of the effective creep strain in $t = 30$ years. (Solver COVES 2¹⁸ – pre and post-processor GiD.¹⁹)

simulated closure of the cavern continue to increase at the bottom with time due to the creep of halite. It stops only when the gas reaches that position.

COVES 2 simulates, in a single run, the excavation, replacement of the brine by a gas with high CO₂

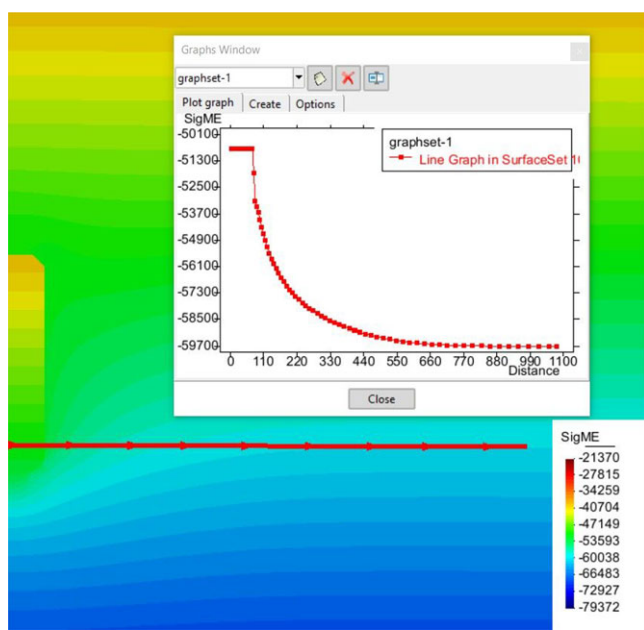


Figure 14. Isosurface of first invariant of stresses after abandonment for 30 years. (Solver COVES 2¹⁸ — pre and post-processor GiD.¹⁹)

content, and the final abandonment of the cavern. When the cavern is completely filled with gas, for the geomechanical simulation, it is necessary to have the bulk modulus of the gas/CO₂ at the critical state. To simulate the abandonment process, it was assumed that the cavern would be filled only with CO₂ due to lack of data for mixtures of gases with varying compositions. Currently, laboratory tests are focusing on obtaining these thermodynamic parameters of gas mixtures at the University of São Paulo in Brazil. For this simulation, a bulk modulus of $K = 80\,000$ kPa was used based on the data generated in the site: webmaster@peacesoftware.de.

Figure 12 shows the isosurface of the radial displacement to be expected in 30 years. One can easily observe the effect of the abandonment of the cavern. The isosurfaces of the displacements show the simulation results with the immediate smooth distribution of deformation between the rock and CO₂ at the critical state. Immediately after the commencement of abandonment of the cavern by closing the wet Christmas tree, the deformation goes to

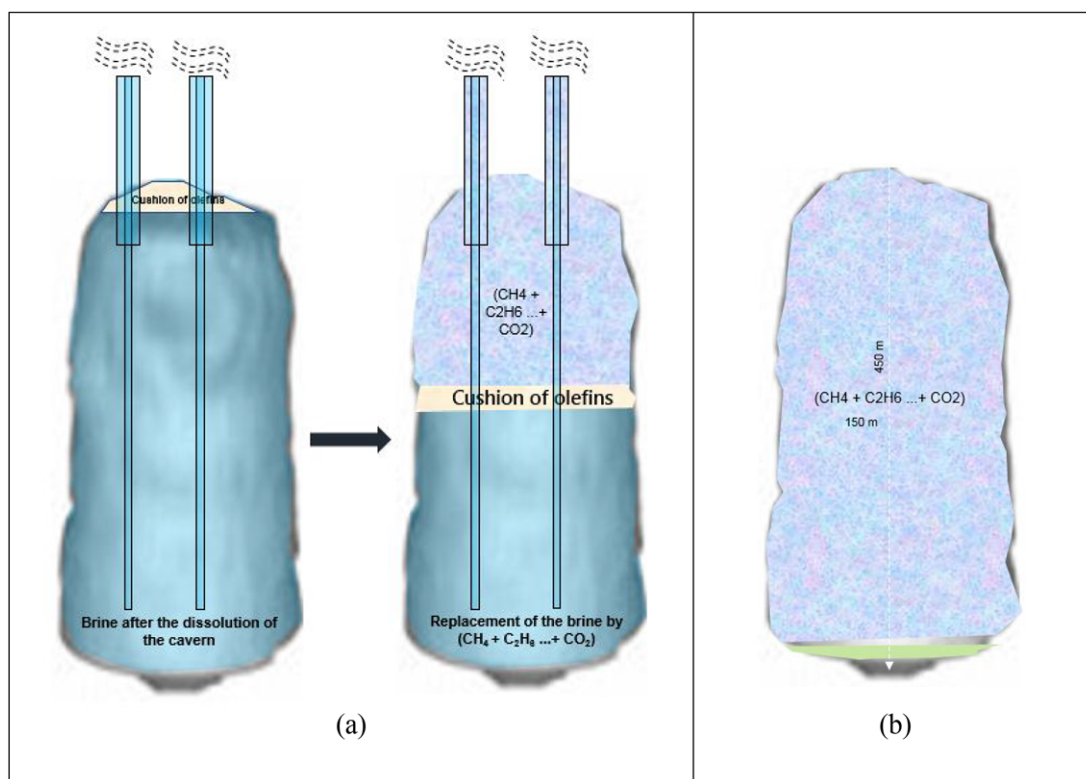


Figure 15. (a) Replacement of brine with gases at the critical state and (b) abandonment of the cavern filled with gases.

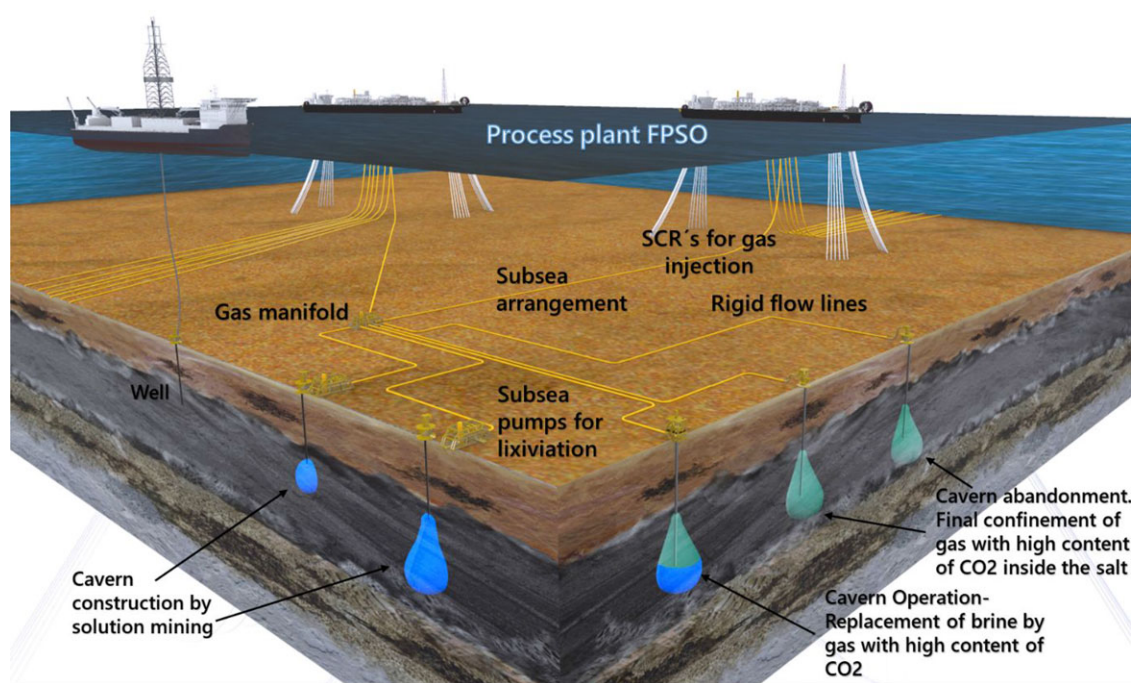
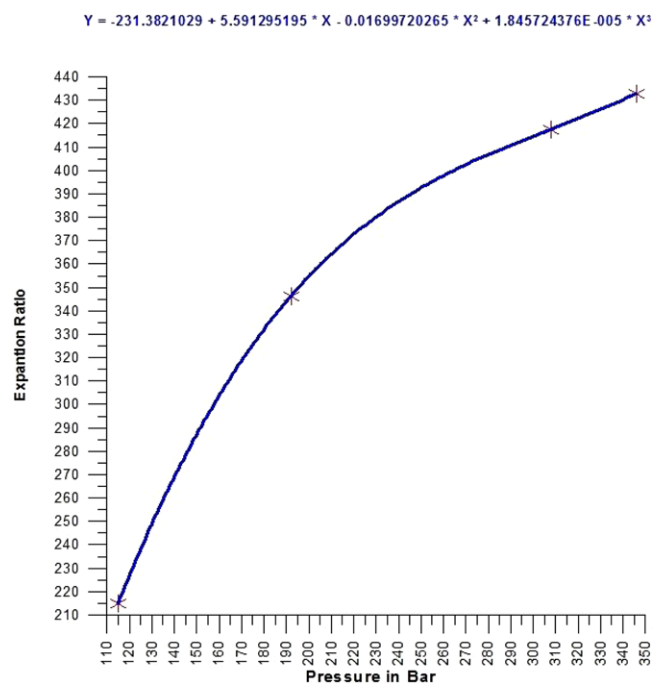


Figure 16. Logistics concept of the CCS system.

Figure 17. Expansion coefficient of a gas stream with 90% of CO₂ at the temperature of 44°C.

a steady-state condition with very small creep closure rate.

The process of scaling of the cavern wall starts when the creep strain reaches values ranging between 5% and

10% in conventional mining without counter pressure. For this case study, the maximum effective creep strain occurs only in the corner of the cavern wall, which will be smoothed over time. Figure 13 shows the isosurface of the effective creep strain accumulated in 30 years with the cavern already abandoned for 22 years.

The phenomenon of halite scaling at the cavern wall is negligible, which allows the use of this giant salt cavern for CCS application with an abandonment pressure of ~450 bar.

Another criterion that must be analyzed is the state of stresses of the cavern after abandonment. The cavern must always support itself against internal pressure limiting the closure tendency and thereby always creating a compressive stress distribution in its surroundings. This prevents the formation of micro-cracks or micro-fractures that may allow gases to permeate to another formation or even to the surface. Figure 14 shows the isosurface of the first invariant of stresses $\{\sigma_m = (\sigma_1 + \sigma_2 + \sigma_3)/3\}$ after 30 years of abandonment. The pressure inside the cavern is the gas injection pressure of about 450 bar, and in the boundary of the cavern wall, the pressure is about 600 bar at the bottom of the cavern. The cavern is protected by a compressive state of stresses, eliminating the possibility of gas leakage over time. Figure 15(a) illustrates brine replacement by gases after the dissolution of the cavern.

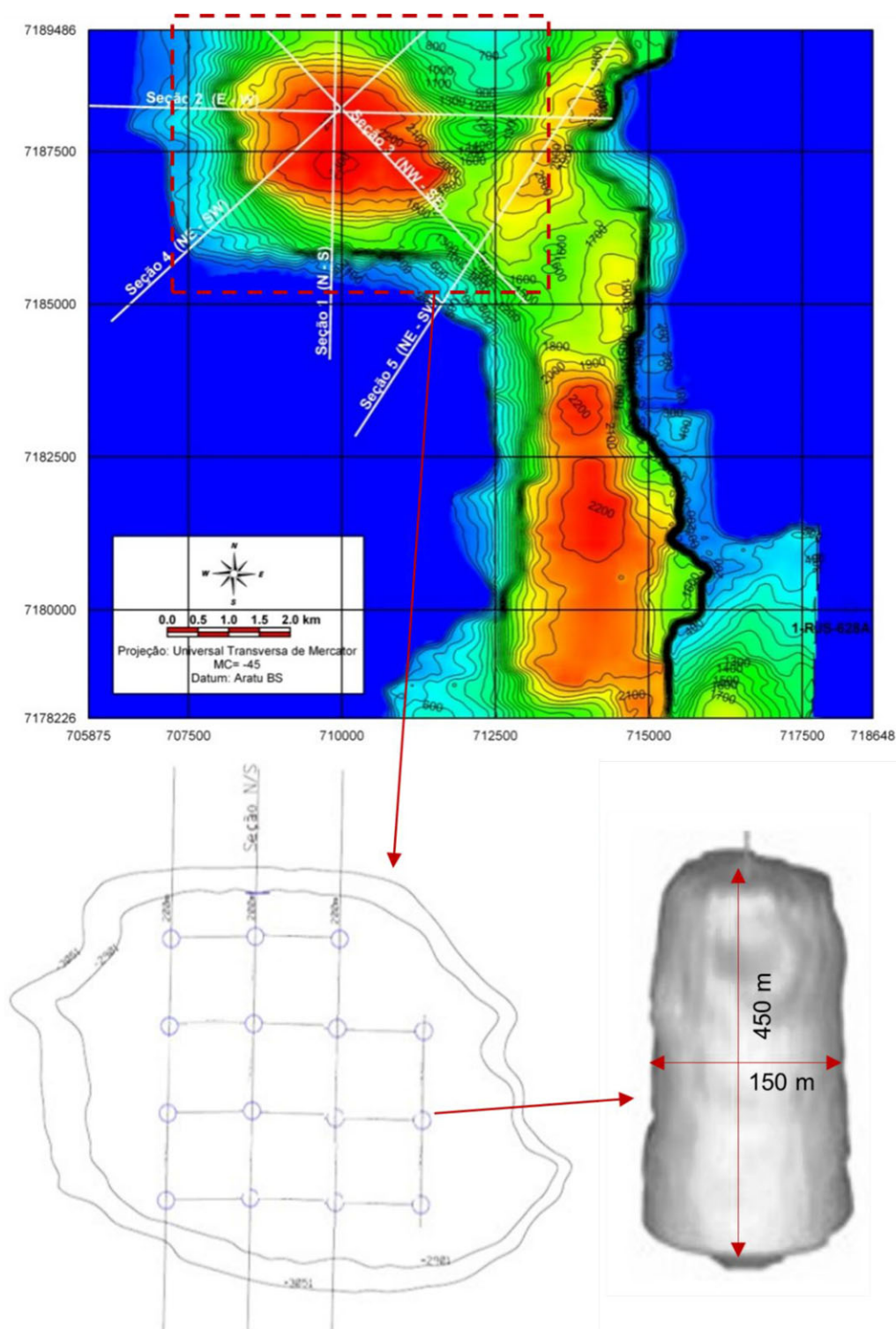


Figure 18. Cluster of salt caverns for CCS in ultra-deep water.
Source: Adapted from⁹.

Figure 15(b) shows the final stage of the cavern at abandonment filled with gases, and Figure 16 shows the logistics concept of the proposed system of CCS for ultra-deep water hydrocarbon production.

Figure 17 shows the curve of the expansion ratio of a gas stream with 90% of CO₂ for the temperature of 44°C as a function of pressure.

By interpolating the expansion curve using the gas pressure inside the cavern, it is possible to calculate the CCS volume of the cavern:

- Storage volume of gas stream with high CO₂ content (90% σ_0) $\rightarrow 518,78 \times 7.4 \times 10^6 = 3\,840\,000\,000\text{ Sm}^3$.
- $P = 446\text{ bar}$, $T = 42^\circ\text{C}$, $\rho = 966.9848\text{ kg/m}^3$.
- CO₂ mass storage $= 7.4 \times 10^6 \times 966.9843 \cong 7.2\text{ billion kg of CO}_2 = 7.2\text{ million tons of CO}_2$.

For the salt dome selected, it is possible to construct 15 caverns, and Figure 18 shows the proposed cluster configuration. Hence, the CCS storage station in ultra-deep water will have the capacity to confine 108 billion kg of CO₂ or 108 million tons of CO₂, as seen in this case study.

Conclusions

This article presents the potential of gas storage in salt caverns built in ultra-deep water in Brazil. Salt caverns are formed through a leaching process, which involves the pumping of hot water at regulated pressure to dissolve the rock salt and remove the resulting brine via a single well, which serves for both injection and withdrawal. The geology of the Brazilian province considered here is characterized by stratification of thick layers of halite with intercalations of carnallite and tachyhydrite salt rock, whose creep strain rate is almost two orders of magnitude higher than halite's creep rate under the same conditions of temperature and pressure. Addressing this challenge requires the adoption of similar technique based on experience gained from an operational sylvanite mine in Sergipe State, Brazil. The mining activity has been very successful with the mine operating for more than 30 years, producing 500 000 tons of potash per year.

The topology of the giant cavern considered in the design is summarized as follows:

- Thickness of the safety halite slab protection between the top of the cavern and the bottom of the sedimentary cap rock: 750 m.
- Depth of the top of cavern: -3440 m .
- Depth of the bottom of the cavern: -3890 m .

- Diameter: 150 m.
- Height: 450 m.
- Geometrical volume of the cavern: $7.4 \times 10^6\text{ m}^3$.

Computational FE simulations using COVES software provided the following findings and conclusions:

- The creep deformation of the experimental cavern is very small because of the support from the brine and gas pressure inside the cavern and due to the low temperature within the cavern. The interval of halite between -2700 m and -3898 m is at a temperature that varies within the following interval: $29^\circ\text{C} \leq T \leq 42^\circ\text{C}$.
- The cavern can be constructed by dissolution at a water depth of 2140 m through the injection of seawater flowing at $1000\text{ m}^3/\text{hour}$.
- For the brine return/gas operation tube, the casing should be 11 7/8" 71.80 lb/ft C-125 HC; for the seawater injection tube, the casing should be 7 5/8" 51.2 lb/ft SCr13/110 ksi.
- The storage volume of the gas stream with a high CO₂ content in one cavern: $3\,840\,000\,000\text{ Sm}^3$.
- Maximum gas pressure expected after abandonment: 465 bar.

Based on the salt dome studied for CCS application, one cavern can store 4 billion Sm³ or 7.2 million tons of CO₂. Considering the pillar size between caverns of 750 m, which is five times the diameter of one cavern, the salt dome can accommodate the construction of 15 caverns, thus providing space for the confinement of approximately 108 million tons of CO₂.

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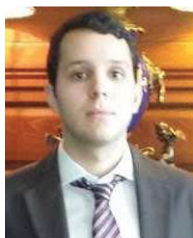
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