

## New miniaturized electrode based on platinum black and poly-eugenol for NO sensing

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Nitric oxide (NO) is a fundamental mediator in regulating the dilation and contraction of blood vessels and being involved in many biological functions [1]. The direct detection of NO presents significant analytical challenges due to its low concentrations (pico to nanomolar range) and the presence of electroactive interferents in the cellular environment, such as dopamine, ascorbic acid, uric acid, and H<sub>2</sub>O<sub>2</sub> [2]. Electrochemistry combined with microelectrodes and electrode surface modification can meet the requirements for real-time NO detection without the addition of any tag or label in the medium, minimizing the perturbation of the microenvironment of the measurement. In this work, we present the development of a new miniaturized electrode for the detection of NO in media containing H<sub>2</sub>O<sub>2</sub>. The sensor was fabricated by modifying a platinum disc ultramicroelectrode (25 µm diameter) in two steps: (1) deposition of Platinum Black (PtB) and (2) electrodeposition of a poly-eugenol. Scanning electron microscopy (SEM) analysis confirmed the morphological changes, including increased surface roughness after PtB deposition and membrane formation due to the poly-eugenol layer. Electrochemical characterization demonstrated that the PtB/poly-eugenol modification enhanced the current response for nitric oxide oxidation by 3-fold compared to a platinum/poly-eugenol electrode. The modified sensor showed reduced interference from nitrite, ascorbic acid, and uric acid. Calibration plots for NO were obtained in the range of 0.48 to 9 µmol L<sup>-1</sup> in the presence of three different concentrations of H<sub>2</sub>O<sub>2</sub>. The sensor also demonstrated good reproducibility, repeatability, and stability over one week when stored in phosphate-buffered saline (PBS).

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### References:

- [1] Ignarro, L. J.; Buga, G. M.; Wood, K. S.; Byrns, R. E.; Chaudhuri, G., Proc. Natl. Acad. Sci. 84., 9265-9269 (1987)
- [2] Hall C. N., Garthwaite J., Nitric Oxide – Biol. Chem. 21., 92-103 (2009)