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# Structural and Spectroscopic Properties of Lanthanide Coordination Polymer (Ln-CPs) with N-phthaloylglycinates and N-(2-carboxybenzoyl)glycinate

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# **Highlights**

New structure of the  $Ln^{3+}$ -(N-phthaloylglycinate) coordination polymer and mixed Eu-Tb compounds with efficient ligand to lanthanide (Ligand  $\rightarrow Ln^{3+}$ ) and terbium  $\rightarrow$  europium energy transfers.

# Resumo/Abstract

Coordination polymers (CPs) designed by lanthanide (Ln3+) ions and organic ligands have attracted great interest for different applications in areas such as tunable luminescence, thermometry, chemical sensing, catalysis, and X-ray scintillator. Herein, structural and spectroscopic properties of the Ln3+ coordination polymers (Ln-CPs) have been synthesized with N-(2-carboxybenzoyl)glycinate (cbgly) and N-phthaloylglycinate (phtgly), {[Tb(phthgly)3(H2O)x]·xH2O}... and  $\{[Eu(phthgly)(cbgly)(H_2O)_2](H_2O)\}_{\infty}$  CPs as well as those mixed  $[Tb_{1-x}Eu_x(phthgly)(cbgly)(H_2O)_2](H_2O)\}_{\infty}$  (x = 0.25, 0.50 and 0.75) are discussed. Structural data revealed that the shortest distances between Ln3+ ions in these CPs are find between 4.0 and 5.3 Å (Figure 1a). Based on the PL spectra of the Ln-CPs, we can suggest that the ligand absorption via S<sub>0</sub>→S<sub>1</sub> transitions produces intense luminescence from the Tb<sup>3+</sup> (CP1 and CP2) and Eu<sup>3+</sup> (CP6) material respectively (Figure 1b). Additionally, in the mixed system  $Tb_{1-x}Eu_x - CPs$ , x = 0.25, 0.50 and 0.75 (CP3-CP5), as can be observed in the photoluminescence (Figure 1c) and X-ray excited optical luminescence (XEOL) (Figure 1d) spectra of the mixed Tb-Eu coordination polymer, the intensities of the bands arising to the  ${}^5D_0 \rightarrow {}^7F_J$  (Eu<sup>3+</sup>) are significantly higher than those ones from  ${}^5D_4 \rightarrow {}^7F_J$  (Tb<sup>3+</sup>) transitions even at low concentration of the "dopant" ion. This behavior can be explained by the high efficiency in the energy transfer from Tb³+ to Eu³+ ion. The Tb³+→Eu³+ energy transfer process was also investigated from the luminescence decay curves, the lifetime value for <sup>5</sup>D<sub>4</sub> level for Tb-CPs (τ = 1.177 ms) decreases deeply to 0.134 ms for x = 0.25 in (CP3), indicating a very efficient luminescence quenching of the Tb<sup>3+</sup> ion by the Eu<sup>3+</sup> ions close to emitting metal. In this way, the transfer efficiency Tb<sup>3+</sup> $\rightarrow$ Eu<sup>3+</sup> of  $\eta_{ET} \approx 90\%$  were determined.

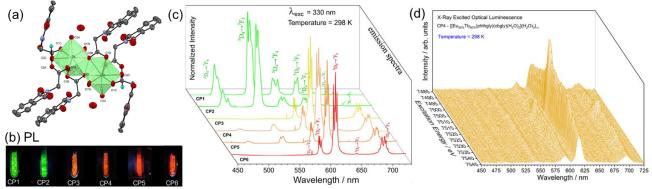


Figura 1. a) Structure of the  $Tb^{3+}$  coordination polymer CP-1. b) Photographs of the CP1 to CP6 (under 330 nm UV light) compounds, c) PL spectra of de CP1 to CP6 at 300 K, and XEOL spectra of the  $[Tb_{0.5}Eu_{0.5}(phthgly)(cbgly)(H2O)_2](H_2O)_{\odot}$  at 300 K.

#### Reference

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