

Persistent luminescence in doped $\text{Lu}_2\text{O}_2\text{S}$ materials: structural analysis and optical investigation

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Key words: Persistent luminescence, Rare-earths, Lutetium, Oxysulfides, Photonic applications.

Highlights

- Doped lutetium oxysulfides are versatile and highly efficient phosphors, with great thermal and chemical stability, and also of easy color tunability.
- Applications of $\text{Lu}_2\text{O}_2\text{S}$ -based materials include X-rays scintillator screens, field emission displays, and advances in medical and industrial areas.
- The effect of doping on the structure and optical behavior of $\text{Lu}_2\text{O}_2\text{S}$ materials was investigated through X-ray diffraction and spectroscopic techniques.

Resumo/Abstract

Rare earth oxysulfides ($\text{RE}_2\text{O}_2\text{S}$, RE^{3+} : Y, La, Gd, Lu) are suitable matrices for luminescent materials, which is due to their high thermal and chemical stability, low cost, and suitable sensitization of trivalent lanthanide ions, resulting in high luminescent efficiency [1]. Lutetium oxysulfide ($\text{Lu}_2\text{O}_2\text{S}$) has been widely investigated as the host material for three-dimensional plasma display panels, field emission displays and light-emitting diodes [2]. $\text{Lu}_2\text{O}_2\text{S}:\text{Eu}^{3+}$ is a persistent luminescent material in which Eu^{3+} efficiently emits red light, while $\text{Lu}_2\text{O}_2\text{S}:\text{Ti}^{3+/4+}$ presents a wide band in the orange region, related to titanium. The duration of these emissions after the irradiation removal can be increased by the insertion of Mg^{2+} , with the purpose of creating charge compensation defects and allowing energy storage in trap levels.

In this work, crystalline structure, optical absorptions, and persistent luminescence behavior of $\text{Lu}_2\text{O}_2\text{S}$, $\text{Lu}_2\text{O}_2\text{S}:\text{Eu}^{3+}$, $\text{Lu}_2\text{O}_2\text{S}:\text{Ti}^{3+/4+}$ and $\text{Lu}_2\text{O}_2\text{S}:\text{Mg}^{2+}$ were investigated. The effect of co-doping was also studied in $\text{Lu}_2\text{O}_2\text{S}:\text{Eu}^{3+},\text{Ti}^{3+/4+}$, $\text{Lu}_2\text{O}_2\text{S}:\text{Eu}^{3+},\text{Mg}^{2+}$, $\text{Lu}_2\text{O}_2\text{S}:\text{Ti}^{3+/4+},\text{Mg}^{2+}$ and $\text{Lu}_2\text{O}_2\text{S}:\text{Eu}^{3+},\text{Ti}^{3+/4+},\text{Mg}^{2+}$ materials.

$\text{Lu}_2\text{O}_2\text{S}$ materials were obtained by a rapid and energy-saving microwave-assisted solid-state synthesis. Phase purity and crystal parameters were obtained through X-ray diffraction (XRD) with Rietveld refinements. The incorporation of Eu^{3+} , $\text{Ti}^{3+/4+}$ and Mg^{2+} ions in $\text{Lu}_2\text{O}_2\text{S}$ host was evaluated, along with structural changes.

Optical absorptions were measured through diffuse reflectance spectroscopy (DRS), which also enabled the calculation of $\text{Lu}_2\text{O}_2\text{S}$ band gap energy with Kubelka-Munk theory. Eu^{3+} 4f-4f transitions and ligand-to-metal charge transfer (LMCT) transitions were observed in the 200-700 nm region of the spectra. The presence of $\text{O}^{2-}/\text{S}^{2-} \rightarrow \text{Eu}^{3+}$ and/or $\text{O}^{2-}/\text{S}^{2-} \rightarrow \text{Ti}^{3+/4+}$ bands proved to be correlated with luminescence mechanism in these materials.

Photoluminescence spectroscopy studies and persistent luminescence measurements demonstrated the effect of the incorporation of activator ions (Eu^{3+} and/or $\text{Ti}^{3+/4+}$), as well as an optically inactive co-dopant (Mg^{2+}), into $\text{Lu}_2\text{O}_2\text{S}$ structure. Emissions in the visible region, excited by UV or X-ray irradiation, and the high storage capability of these materials provide several applications in dosimetry, bioimaging and optoelectronic devices [1,2].

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Agradecimentos/Acknowledgments

The authors thank the financial support of São Paulo Research Foundation, Brazil (FAPESP, grants #2019/26689-1 (KTF), #2021/05603-1 (LCVR), and #2021/08111-2 (HFB)).