



Article

Phytocompounds Recovered from the Waste of Cabernet Sauvignon (*Vitis vinifera* L.) Vinification: Cytotoxicity (in Normal and Stressful Conditions) and In Vitro Photoprotection Efficacy in a Sunscreen System

Alexandra Almeida Hübner ¹, Daniel Pecoraro Demarque ¹, Felipe Rebello Lourenço ¹, Catarina Rosado ², André Rolim Baby ^{1,*}, Irene Satiko Kikuchi ^{1,†} and Elfriede Marianne Bacchi ^{1,†},

- Department of Pharmacy, Faculty of Pharmaceutical Sciences, University of São Paulo, São Paulo 05508-000, Brazil
- CBIOS, Universidade Lusófona's Research Center for Biosciences & Health Technologies, 1749-024 Lisbon, Portugal
- * Correspondence: andrerb@usp.br
- † These authors contributed equally to this work.
- [‡] In memoriam.

Abstract: We investigated plausible reuse for the dermocosmetic industry of byproducts from the winemaking process of red grapes (Vitis vinifera L. cv. C. Sauvignon) through the evaluation of one extract (grape pomace extract, GPE) and two fractions (one chloroform, GPE-CHF; one ethyl acetate, GPE-EAF). The samples were characterized analytically by liquid chromatography (HPLC) using a NIH 3T3 fibroblast cell culture to verify a cytosafety profile in normal and stressful environment (presence of H₂O₂), and by using it in a sunscreen system to observe improvements in the in vitro efficacy by diffuse reflectance spectrophotometry with an integrating sphere. The HPLC results for GPE-EAF and GPE-CHF samples with the best profile of syringic and p-coumaric acids, quercetin, and trans-resveratrol were used in the further assays. GPE-EAF and GPE-CHF, both at 30.00 µg/mL, maintained the cell viability in the absence of H₂O₂ (normal condition). In the sequence, GPE-EAF and GPE-CHF were evaluated against the oxidative stressor H₂O₂ in NIH 3T3 cells. A sharp drop in viability was only observed for GPE-CHF, and cytotoxicity of GPE-EAF was considered absent even in a hostile environment. Since GPE-EAF previously developed the best results, its potential performance was investigated in a sunscreen system. The in vitro sun protection factor of the phytoderivative-free formulation was 9.0 + 2.5; by adding GPE-EAF at 10.0%, its efficacy was elevated to 15.0 + 2.5. Both samples suffered a negative effect after artificial ultraviolet exposition (500 W/m²); however, the presence of GPE-EAF improved the photostability of the sunscreen system.

Keywords: fibroblast culture; sun protection factor; sunscreen; Vitis vinifera; winemaking waste



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1. Introduction

Grapes were domesticated thousands of years ago and the *Vitis vinifera* L. grapevine is the most cultivated grape species in the world, being about 50–70% of the entire production used in the elaboration of fine wines. The Cabernet Sauvignon variety (*V. vinifera* L. cv. C. Sauvignon) is one of the most cultivated and widely distributed in almost all continents. In winemaking, 20–30% of solid waste is formed from skins, seeds, residual pulp, stalks, and fermentation yeasts, and the grape pomace (GP) is the main residue of the wine industry, which could be representative in terms of a byproduct volume. On average, 1 ton of wine produces 200 kg of GP, generally used for energy, fertilizer production, animal food, distilled beverages, or building construction. The inappropriate waste disposal in the environment has an impact on crop, living beings, and water and soil quality due to antinutritional substances, oxygen depletion, and greater resistance to microorganisms [1,2].

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Thus, this scenario highlights the necessity to create mechanisms to improve the reuse and to reduce this kind of industry residue.

GP contains a variety of substances that include unsaturated fatty acids, monosaccharides, polysaccharides, proteins, fibers, minerals, and polyphenols. Compared to other agrofood residues, the red GP (peels and seeds) shows high amount of phenolics, possibly due to the presence of flavanols in the seeds [3,4]. The waste of polyphenols remaining in this pomace and the greater demand for natural products, mainly with antioxidant, antibacterial, and antiviral activities, represents a sustainable and ecofriendly opportunity to obtain these bioactive compounds in several biotechnological applications [2,4].

Nowadays, consumers are more aware of and interested in natural topical products with less harm to health and to the environment instead of synthetic substances [5]. Natural antioxidants provide biological and therapeutic activities, high tolerability, and, additionally, biodegradability. Owing to several properties of polyphenols from *V. vinifera*, the development of dermocosmetics is an interesting strategy to vehiculate such substances, thus providing better health skin conditions [5–7]. For instance, specialized literature reported that GP extract was able to improve the in vivo sun protection factor (SPF) of a sunscreen system by about 20% [8].

Aiming at to investigate a plausible reuse of byproducts from the winemaking process in the dermocosmetic industry, we obtained one extract and two fractions from the pomace of red grapes (*V. vinifera* cv. C. Sauvignon). The natural samples prepared from the winemaking waste were characterized analytically by liquid chromatography, evaluated in a cell culture to design a cytosafety profile in a normal and in a stressful environment, and the most successful sample indicated from these tests had its potential application challenged in a sunscreen system by diffuse reflectance spectrophotometry with an integrating sphere.

2. Materials and Methods

2.1. Materials

Dulbecco MEM (DMEM) supplemented with L-glutamine, sodium bicarbonate, antibiotic, antimycotic, fetal bovine serum (FBS), tripsin, and MTT (3-(4,5-dimethylthiazolyl-2)-2,5-diphenyltetrazolium bromide) were from Cultilab® (Campinas, Brazil). Dimethyl sulfoxide (DMSO) was from Amresco® (São Paulo, Brazil) and ethyl alcohol was bought in Ciclo Farma Fabbe® (Serrana, Brazil). Phosphate-buffered saline was from Synth® (Diadema, São Paulo). Potassium chloride P.A., anhydrous monobasic potassium phosphate, and anhydrous bibasic sodium phosphate were purchased from Inlab® (São Paulo, Brazil). Hydrogen peroxide 35% was from Scharlab® (São Paulo, Brazil). Acetic acid, quercetin (\geq 98% purity), p-coumaric acid (\geq 98% purity), and syringic acid (\geq 95% purity) were from Sigma-Aldrich® (Cotia, Brazil) and trans-resveratrol (\geq 98% purity) was from Active Farmacêutica® (Palhoça, Brazil). Acetonitrile was from Honeywell® (Itajubá, Brazil). NIH 3T3 fibroblast cells (P6, 18th January 2010) were kindly donated by the Clinical Cytopathology Laboratory, Faculty of Pharmaceutical Sciences, University of São Paulo (São Paulo, Brazil). Ultrapure water was used in all experiments.

2.2. Agroindustrial Vinification Byproduct

Grape pomace byproduct (skins, seeds, stalks) from the winemaking industry was kindly provided by Vinícola Beraldo di Cale located in Jundiaí, state of São Paulo, Brazil. Red grapes (*V. vinifera* cv. C. Sauvignon) were collected during the end of summer in Caxias do Sul vineyard, state of Rio Grande do Sul, Brazil (29°13′29″ S, 51°9′31″ W and altitude 470 m). A voucher specimen (Hübner, A.1) was deposited in the herbarium (SPF) at the Institute of Bioscience, IB/University of São Paulo.

2.2.1. Dry Whole Grape Pomace Extract (GPE)

The vinification residue from the manufacture of wine (grape pomace byproduct) was dried in an air circulation oven (Fabbe Center®, São Paulo, Brazil) at 40 °C for five days until constant mass. A sample of the grape pomace weighing 4.65 kg was ground in a knife

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and hammer mill (Thomas Scientific[®], Swedesboro, NJ, USA), passing through a 1 mm sieve. The fine dried powder (20 mesh) was extracted with ethanol:water (7:3 v/v) at 25 °C using the percolation method. The leachate was concentrated at reduced pressure around 65 °C in an ascending film evaporator, homogenized, and the aqueous solution freeze dried (lyophilizer model Liotop L202, Liobras[®], São Carlos, Brazil). The GPE was stored in tight, light-resistant container.

2.2.2. Sample Fractions

GPE (20 g) was dissolved in 200 mL of ultrapure water and placed on a magnetic stirrer for 30 min. The formed aqueous solution was submitted to consecutive liquid–liquid extraction using chloroform or ethyl acetate [9]. First, the aqueous solution was extracted with 60 mL of chloroform by shaking (5 min) it at room temperature. The procedure was repeated three times until reaching a total volume of 200 mL, combined and evaporated to dryness under reduced pressure at 40 $^{\circ}$ C. The solid residue was denominated GPE-CHF. Forthwith, the same procedure was performed using ethyl acetate. The solid residue obtained after ethyl acetate evaporation was called GPE-EAF.

2.3. HPLC Analysis of Polyphenols

The quantification of polyphenols in the GPE, GPE-CHF, and GPE-EAF fractions were based on the comparison of retention times and maximum absorbance values of the peak areas detected in the samples with those obtained with the markers (p-coumaric acid, syringic acid, quercetin, and resveratrol) [10,11]. High performance liquid chromatography (HPLC) quantifications were performed with a Shimadzu[®] chromatograph and C18 Shim-pack VP-ODS column (4.3 \times 250 mm, 5 μ m), oven temperature at 35 °C, flow of 1.0 mL/min. The solvent system was (A) purified water (Milli-Q) and 0.1% acetic acid, and (B) acetonitrile. Gradient conditions were 0-10 min (98A/2B); 10-80 min (60A/40B); 80-85 min (2A/98B). GPE and GPE-EAF/GPE-CHF fractions were at 14.70 and 5.00 mg/mL, respectively, and the injection volume was 30 µL. The samples were evaluated using the linear regression equation for the analytical curve of the markers (equivalent in µmol/L extract). The result of phenolics was expressed as mean \pm standard deviation (n = 9). Linear ranges of p-coumaric acid (347 nm), syringic acid (275 nm), and quercetin (309 nm) were between 1.00 and 100.00, and trans-resveratrol (309 nm), from 0.10 to 10.00 µg/mL. The coefficients of determination (R²) were all at least 0.996. Phenolic contents (mg/mL) were calculated according to Equation (1).

Phenolic [mg/g] = concentration (μ g/mL) × 1 g/sample concentration (g/mL)/1000 (1)

2.4. Cytotoxicity Establishment

Solvents were sterilized by using sterile membrane with 0.20 μ m pore size (Minisart®, Sartorius Stedim Biotech, Goettingen, Germany). A 10 μ L aliquot of each sample solution derived from *V. vinifera* was added to a single well containing cells in 90 μ L of culture medium (DMEM + 10% FBS) to reach the final concentration of 30.00 μ g/mL. The mixture of DMSO (0.25%) and EtOH (0.75%) was used in this assay. Water was the negative control. Hydrogen peroxide 1:50.000 (H₂O₂, 206 μ M) [12,13] was tested as the oxidative-process inducer to provide a hostile environment to further determine cytosafety profile.

2.4.1. NIH 3T3 Fibroblast Cell Viability

NIH 3T3 cells (3rd passage) were cultivated in monolayers in 25 cm² flasks containing DMEM and 10% FBS, at 37 °C in a humidified atmosphere with 5% CO₂ (Thermo Fisher Scientific®, Waltham, MA, USA) until reaching, approximately, 80% confluence. The fibroblasts were trypsinized and cell concentration was adjusted to 2.38×10^4 cells/well in culture medium. A 90 μ L volume of this cell suspension was distributed in each well and cultured for 24 h before interaction with the samples at 37 °C for 48 h [14]. Immediately afterward, an aliquot of 100 μ L of the 0.5 mg/mL MTT aqueous solution was added to each

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well and incubated for 3 h. Then, 100 μ L of DMSO was added to each well and incubated for 1 h. The absorbance reading was performed in a multimode microplate reader (Infinite® M200 PRO, Tecan, Männedorf, Switzerland) at 570 nm. Cell viability of the samples (n = 8) was calculated by the ratio of the mean absorbance (A_m) of the treated and control group, according to Equation (2).

Cell viability (%) =
$$A_{\text{m treated}}/A_{\text{m control}} \times 100$$
 (2)

2.4.2. Hostile Environment by H₂O₂ Treatment

NIH 3T3 cells (2.38 \times 10^4 cells/well) were cultured in a 96-well microplate (Kasvi $^{\circledR}$, São José dos Pinhais, PR, Brazil) for 24 h and followed by interaction with samples at 30 $\mu g/mL$, 37 $^{\circ}C$ for 48 h. After this period, 10 μL of H_2O_2 (206 μM) was added to each well and incubated at 37 $^{\circ}C$ and 5% CO_2 for 3 h. Afterward, 10 μL of MTT (5.0 mg/mL) was added to each well and incubated for 3 h. The supernatants were removed from the wells individually and then 100 μL of DMSO was added to solubilize the blue formazan crystals from living cells, shaken gently, and incubated at 37 $^{\circ}C$ for 1 h [12]. The absorbance reading followed the same as described earlier.

2.5. In Vitro Sun Protection Factor (SPF) and Photostability Establishment

In vitro SPF of the sample with the best performance from the previous assays was determined using a diffuse reflectance spectrophotometer with an integrating sphere (Labsphere UV-2000S® Transmittance Analyzer) [15–17], illustrated in Figure S1 (Supplementary Materials) as "diffuse reflectance spectrophotometer with integrating sphere from de Laboratory of Cosmetology, Faculty of Pharmaceutical Sciences, University of São Paulo [18]". For the photostability test, we used a chamber with artificial UV ray emission (Suntest® CPS Atlas, Linsengericht, Germany). A sample amount of 0.75 mg/cm² was applied uniformly over polymethylmethacrylate (PMMA) plates (HelioScreen® Helioplate HD 6, North Sutton, NH, USA) (n = 3). Right after sample application, plates were kept at in the dark at room temperature for 15 min. After 0, 1, and 2 h of exposure to UV radiation at the dose of 500 W/m², the in vitro SPF was obtained through the UV2000 software. The sunscreen system was investigated earlier by our Research Group with its composition described by Hübner and coworkers, 2020 [8]; however, Table 1 describes the abovementioned system in terms of active ingredients. The best $V.\ vinifera$ -derived sample from previous tests was used at 10.0% (w/w).

Table 1. Qualitative and o	luantitative of the sunscreen-s	ystem active composition.

To any diameter	Concentration (% w/w)		
Ingredients	SS	SS + GPE-EAF	
GPE-EAF	-	10.0	
Butylmethoxydibenzoyl methane		2.5	
Ethylhexyl methoxycinnamate		5.0	
Ethylhexyl dimethyl PABA		4.0	

 $\overline{GPE\text{-}EAF}$ = ethyl acetate fraction of grape pomace extract; SS = sunscreen system; SS + GPE-EAF = sunscreen system + ethyl acetate fraction.

2.6. Statistical Analysis

Data were expressed as mean \pm standard deviation. Analysis of variance test (ANOVA) followed by Tukey's test were performed to statistical analyze and verify differences between groups. Minitab 21 (Minitab Inc., LLC, State College, PA, USA), Statistic Program 13.5 (TIBCO Software Inc., Palo Alto, CA, USA) and MS Excel (Microsoft Corp., Redmond, WA, USA) were used; p-values < 0.05 were considered significant.

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3. Results

3.1. Analytical Characterization of the Natural Samples

The aspect of extract and fractions is illustrated in Figure 1. We noticed differences amongst GPE, GPE-CHF, and GPE-EAF (p < 0.05) regarding the presence and quantity of some natural compounds. Syringic acid concentration was the highest one (p < 0.0001). Syringic and p-coumaric acids and quercetin were distinct in all samples. Resveratrol concentration in GPE and GPE-CHF was considered similar (p > 0.05). The abovementioned results are shown in Figure 2. The GPE-EAF and GPE-CHF were the samples with the best profile of phytocompounds, which were further investigated in this research work.

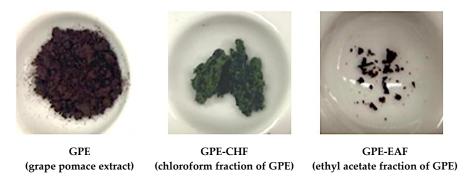


Figure 1. Aspect of the samples from the winemaking process using *V. vinifera* cv. C. Sauvignon.

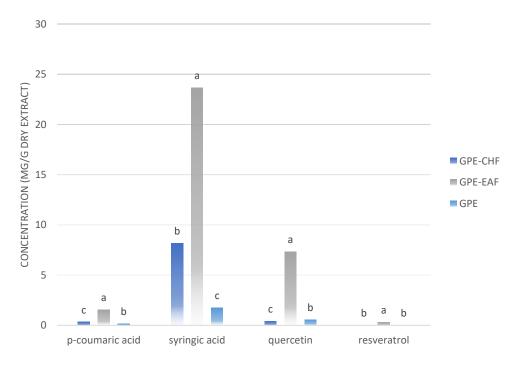


Figure 2. Polyphenols in the grape pomace extract (GPE) and in the ethyl acetate (GPE-EAF) and chloroform (GPE-CHF) fractions. Results are expressed as mean \pm standard deviation. In each group, different letters (a, b, and c) indicate significant differences (p < 0.05).

3.2. Cytotoxicity Establishment

GPE-EAF and GPE-CHF (in EtOH/DMSO solution) were tested using NIH 3T3 fibroblasts. The assay demonstrated that the concentration of the samples at 30.00 μ g/mL maintained cell viability in the absence of H₂O₂, being *V. vinifera* derivatives cytobiocompatible with the fibroblast cells. There was no difference between the negative control and culture medium (p > 0.05), and we also observed that solvents (EtOH/DMSO) did not interfere with cell viability [12,19].

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In the sequence, GPE-EAF and GPE-CHF were evaluated in vitro against the oxidative stressor H_2O_2 in NIH 3T3 cells, as shown in Table 2. A sharp drop in viability was only observed in the presence of GPE-CHF. GPE-EAF maintained the cell viability, being considered absent of cytotoxicity even in a hostile environment (presence of H_2O_2).

Table 2. Effect of ethyl acetate (GPE-EAF) and the chloroform (GPE-CHF) fractions on viability of NIH 3T3 cells in the absence (normal condition) and presence (hostile conditions) of H_2O_2 .

Sample	NIH 3T3 Viability (%)	
	H ₂ O ₂ -Free	H ₂ O ₂ (206 μM)
Water	$100.0 \pm 2.0 ^{\mathrm{a,b}}$	76.7 ± 1.5 ^c
GPE-EAF	102.9 ± 2.1 a	87.5 ± 1.8 b
GPE-CHF	102.7 ± 2.1 a	56.9 ± 1.1 ^c

Cell viability (%) was assessed by MTT. Results are expressed as the mean \pm standard deviation (n = 8). Analysis of variance test (ANOVA) followed by Tukey's test showed significantly differences between groups (Control, GPE-EAF, and GPE-CHF) (p < 0.001) and interaction between the factors "treatment" and "groups" (p < 0.001). Different letters indicate significant differences.

3.3. In Vitro SPF and Photostability

GPE-EAF developed the best results from the previous assays; therefore, its potential performance was investigated in a sunscreen system. The phytoderivative-free formulation developed in vitro SPF equal to 9.0 \pm 2.5. By adding GPE-EAF at 10.0%, the sunscreen system had its efficacy parameter elevated to 15.0 \pm 2.5. The photostability profiles are illustrated in Figure 3. Both samples suffered negative impact after 1 and 2 h of artificial UV exposition; however, the presence of GPE-EAF improved the photostability of the sunscreen system used, at least, for 1 h.

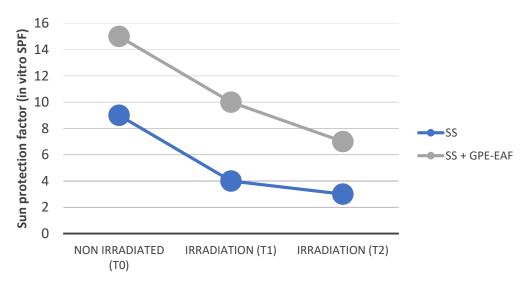


Figure 3. Photostability profile of the sunscreen systems. SS = sunscreen system; SS + GPE-EAF = sunscreen system + 10.0% ethyl acetate fraction of grape pomace extract; T1 = 1 h of irradiation; T2 = 2 h of irradiation. According Dunnett's multiple comparisons test, SS + GPE-EAF and SS had differences in the in vitro SPF in T0 and T1 (p < 0.001; p < 0.005, respectively).

4. Discussion

Valorization of natural compounds for use in the pharmaceutical, food, and cosmetic industries is a conscious measure of the entire production chain of agroindustrial resources, promoting sustainability and circular economy development. Circular economy aims to prolong the time of use of materials, products, and substances from waste, minimizing disposal and its consequences on the environment. Therefore, recovering an agroindustrial

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byproduct through its reuse and recycling make them new sources of raw materials for several sectors of the economy. During red wine vinification, there is production of organic residues such as stalks, pomace, and lees, but mainly grape pomace, composed of 25% seeds, 25% stalks, and 50% skins, and corresponding to approximately 20–25% of the weight of the processed grapes [20,21].

The recovery of polyphenols from grape pomace depends on the plant material, extractive methods, suitable solvents, and harvest season [12]. Compared to other V. vinifera grape pomace extracts, the extractive condition adapted herein was efficient in obtaining active compounds, such as p-coumaric and syringic acids and quercetin. We found in the specialized literature similar values for syringic acid (1.74 \pm 156.3 mg/g GPE DW) and quercetin (0.557 \pm 83.9 mg/g GPE DW); a lower value for p-coumaric acid (63.6 \pm 5.3 mg/g GPE DW); and a higher value for trans-resveratrol (36.0 \pm 4.9 mg/g GPE DW) [2,11,22]. The GPE-EAF sample had the highest concentrations of the markers, i.e., syringic acid and p-coumaric acids, quercetin, and resveratrol (23.64 \pm 4.63; 1.54 \pm 0.29; 7.33 \pm 3.13; and 0.29 ± 0.16 mg/g dry extract, respectively). C. Sauvignon pomace can present variation in its qualitative and quantitative phenolic profiles depending on the cultivation region and extraction procedure. V. vinifera is a species rich in polyphenols such as flavonoids in the skins, seeds, and stalks, and nonflavonoids in the fruit pulp. Biologically, phenolic compounds can act by sequestering and inactivating reactive oxygen species (ROS), for instance, thus protecting DNA. Additional to the antioxidant action, the oral and topical use of flavonoids are associated with anti-inflammatory, antimutagenic, anticarcinogenic, and enzyme-modulating properties [20,22-24]. Flavonoids may prevent oxidative stress and, consequently, cellular damage. In grapes and red wine, the main groups capable of exerting antioxidant behavior are flavonols (quercetin, kaempferol, myricetin, epicatechin, and rutin), flavones (luteolin, apigenin, and tangeritin), flavanones or dihydroflavones (hesperitin, naringenin, and eriodictyol), flavanols or flavan-3-ols (catechins), and anthocyanins (cyanidin, delphinidin, malvidin, pelargonidin, and peonidin). The beneficial health effects of polyphenols are given according to the grape variety, place of cultivation, extractive process, and antioxidant content [25–27].

Considering that GPE samples presented the lowest content of the markers, we continued the investigation with GPE-EAF and GPE-CHF. Regarding the H₂O₂-free group, both fractions were biocompatible with the cell line in the tested concentration. This cytosafety profile allowed us to stress the fibroblasts with H_2O_2 to observe how the fractions would interact with the cells in such unfavored condition and suggest a possible mechanism of action to use in dermocosmetics. Specialized literature reported that ROS formation after exposure to H₂O₂ could be suppressed by pretreatment with 30.00 µg/mL of some extracts rich in polyphenols [12]. H_2O_2 , is not as reactive as superoxide $(O_2^{\bullet-})$ and hydroxyl (${}^{\bullet}OH$) radicals; however, in high concentrations in the biological systems it results in cellular and molecular damage. All cells, mitochondria, and certain endogenous enzymes produce H₂O₂. The relationship between the effect of redox and the human aging process has aroused the interest of the scientific community to the use of antioxidants [28]. Studies have evaluated the antioxidant activity of grape pomace in preventing premature aging in skin cell cultures; however, to the best of our knowledge, little is known about the interactions and biological effects of the winemaking byproduct fractions, more specifically their use in sunscreen systems [29,30]. The extract richest in phytochemicals (GPE-EAF) showed better safety profile compared to GPE-CHF in cells damaged by H₂O₂. The most nonpolar sample, GPE-CHF, showed the strongest interaction with H_2O_2 stress, maximizing the oxidative effect (synergistic effect of H₂O₂ and GPE-CHF fraction induced loss of cell viability), which justified the use only of GPE-EAF in the in vitro establishment of photoprotection efficacy.

Since the GPE-EAF sample developed the best results from the aforementioned assays, i.e., more content of some phytocompounds and safety in the fibroblast culture in the absence and in the presence of H_2O_2 , we further investigated its potential as an adjuvant for use in sunscreens, as a performance booster and/or photostabilizer agent. The GPE-CHF negative behavior in the fibroblasts treated with H_2O_2 must be considered in the

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context of the cutaneous photoprotection, since the skin in contact with the UV radiation can suffer augmentation of reactive species [31] that could be once more amplified by the presence of chloroform fraction.

The addition of GPE-EAF at 10.0% improved the sunscreen-system efficacy in more than 60% (in vitro SPF from 9 to 15). Since our natural sample was recovered from the waste of the winemaking industry, such favorable response highlights a real source of potential adjuvants and/or active ingredients to be considered in the cosmetic industry. We have been investigating the behavior of bioactive sunscreens since 2008 and several natural compounds or substances, inspired by their nature, were studied with distinct performance profiles [32–34]. Herein, GPE-EAF, rich in syringic acid and quercetin (greater content) but also presenting, in low concentration, p-coumarinic acid and trans-resveratrol, was able to elevate the sample's action against UVB radiation by improving the SPF in vitro. Additionally, we noticed an interesting behavior of our fraction as a photostabilizer for at least 1 h of artificial irradiation. Even though the in vitro SPF decayed in function of the irradiation time exposition, GPE-EAF maintained the best results of this efficacy parameter (for 1 h), suggesting that the phytochemical interaction [35] with the sunscreen system favored its functional stability.

Considering some limitations from this study, although our investigation has real potential use in the dermocosmetic industry, possible impacts on regional, agronomic (soil, grape quality, strain, and others), and/or climatic variation (radiation, precipitation, wind speed, and others) may affect the grape composition. For example, phenolic compounds are sensitive to variations in soil and climate [12,25]. Another important point is the clonal variability, observed in wine grape *V. vinifera* cv. Kalecik Karasi. Results showed approximately 40% phenotypic variation in grape berry clones and significant differences in the number of phenolic compounds (protocatechuic acid and gallic acid) [36]. Therefore, further studies with other grape strains and with *V. vinifera* from different regions of the world may better elucidate the chemical variables involved in the cytosafe and cytotoxic effects.

5. Conclusions

From the $V.\ vinifera$ waste of the winemaking process, we obtained three natural derivatives (one extract and two fractions). GPE-CHF and GPE-EAF fractions were the ones with the highest content of syringic and p-coumaric acids, quercetin, and trans-resveratrol. Both fractions were cytobiocompatible with fibroblasts in normal condition culture; however, GPE-CHF developed cytotoxicity when the cells were treated with H_2O_2 (stressful environment). Since GPE-EAF had more content of the evaluated phytocompounds and developed a protective profile in the fibroblast culture in the H_2O_2 presence, its potential as an adjuvant for sunscreens was investigated. GPE-EAF elevated the in vitro SPF, along with its photostability for at least 1 h, highlighting that winemaking waste could be considered as a plausible source for dermocosmetic adjuvant and/or active ingredients.

Supplementary Materials: The following supporting information can be downloaded at: https://www.mdpi.com/article/10.3390/cosmetics10010002/s1, Figure S1: Diffuse reflectance spectrophotometer with integrating sphere from de Laboratory of Cosmetology, Faculty of Pharmaceutical Sciences, University of São Paulo [18].

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Conflicts of Interest: The authors declare no conflict of interest.

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