


# Mechanism of imidazole inhibition of a GH1 $\beta$ -glucosidase

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## Keywords

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Imidazole is largely employed in recombinant protein purification, including GH1  $\beta$ -glucosidases, but its effect on the enzyme activity is rarely taken into consideration. Computational docking suggested that imidazole interacts with residues forming the active site of the GH1  $\beta$ -glucosidase from *Spodoptera frugiperda* (Sf $\beta$ gly). We confirmed this interaction by showing that imidazole reduces the activity of Sf $\beta$ gly, which does not result from enzyme covalent modification or promotion of transglycosylation reactions. Instead, this inhibition occurs through a partial competitive mechanism. Imidazole binds to the Sf $\beta$ gly active site, reducing the substrate affinity by about threefold, whereas the rate constant of product formation remains unchanged. The binding of imidazole within the active site was further confirmed by enzyme kinetic experiments in which imidazole and cellobiose competed to inhibit the hydrolysis of *p*-nitrophenyl  $\beta$ -glucoside. Finally, imidazole interaction in the active site was also demonstrated by showing that it hinders access of carbodiimide to the Sf $\beta$ gly catalytic residues, protecting them from chemical inactivation. In conclusion, imidazole binds in the Sf $\beta$ gly active site, generating a partial competitive inhibition. Considering that GH1  $\beta$ -glucosidases share conserved active sites, this inhibition phenomenon is probably widespread among these enzymes, and this should be taken into account when considering the characterization of their recombinant forms.

Glycoside hydrolases (GH), based on their sequence and structural similarities, are grouped into more than 170 families in the CAZY databank (<http://www.cazy.org>) [1,2]. Among these families, the GH1 family comprehends  $\beta$ -glucosidases, enzymes that hydrolyze the O- or S-glycosidic linkages of  $\beta$ -glycosides, and fold as TIM barrels ( $[\beta/\alpha]_8$  barrel) [3], a repetition of a  $(\beta/\alpha)$  motif [4].  $\beta$ -glucosidases enzymes (3.2.1.21) are over-spread among all forms of life, from bacteria to eukaryotes [5], and are related to a series of biological functions such as digestion, defense, and plant cell wall metabolism [6–8].

$\beta$ -glucosidases catalyze glycoside hydrolysis via a double-displacement mechanism involving two

carboxylic acid-containing side chains in the active site. One of these groups, a carboxylate, functions as a nucleophile, leading to a glucosyl–enzyme intermediate. The other carboxylate acts as a general acid catalyst in the formation of this intermediate, and as a general base catalyst in its breakdown promoted by water [9]. Inhibitors of these enzymes, responsible for the disruption of their activity, have played a vital role in revealing their functions in the living system by modifying or blocking specific metabolic processes [10], leading to several applications of these chemicals in agriculture and medicine [11]. For instance, the finding that  $\beta$ -glucosidases are linked to Gaucher's syndrome [12] prompted detailed investigations on

## Abbreviations

EDC, 1-ethyl-3-(dimethylamino-propyl) carbodiimide; GEE, glycine ethyl ester; GH, glycoside hydrolases; Sf $\beta$ gly, GH1  $\beta$ -glucosidase from *Spodoptera frugiperda*.

compounds able to inhibit these enzymes. Therapeutic use of these inhibitors can be found in the control of this disease, which is related to disturbed lysosomal storage [12]. Besides the therapeutic applications,  $\beta$ -glucosidase inhibitors are also useful for probing the binding properties of these enzymes and may also be used as a tool for investigating and elucidating their mechanism of catalysis [13–16].

Here, we investigated the effect of imidazole, a largely employed reagent in the recombinant protein purification [17,18], as an inhibitor of the recombinant GH1  $\beta$ -glucosidase from the fall armyworm *Spodoptera frugiperda* (hereafter called S $\beta$ gly). S $\beta$ gly is a secreted digestive enzyme that is associated with the glycocalyx of midgut epithelial cells [6], and it was previously studied regarding substrate recognition, catalysis, thermal stability, and oligomerization [6,19–22]. The S $\beta$ gly crystallographic structure (PDB ID: 5CGO) [20] shows an active site divided into subsites, among which the subsite –1, which binds the monosaccharide of the substrate nonreducing end, is shaped by a set of residues that form hydrogen bonds with the ligand [23], substrate glycones, or inhibitors, which indicate a potential interaction site with imidazole. Imidazole derivatives have been shown before as a  $\beta$ -glucosidase inhibitor [13–16,24]; however, those experiments were conducted with nonhomogeneous sweet almond  $\beta$ -glucosidase samples. In addition, only low concentrations of the derivatives were tested [16]. Here we performed experiments with a purified GH1  $\beta$ -glucosidase (S $\beta$ gly), which crystallographic structure is already known. In addition, we used higher concentrations of imidazole, the same reagent and conditions employed in the recombinant protein purification. That led to the characterization of the mechanism of inhibition of this GH1  $\beta$ -glucosidase by imidazole. Finally, considering the conservation of the residues forming their active sites, this inhibition mechanism is probably widespread among GH1  $\beta$ -glucosidases, information that should be taken into account in the characterization of these recombinant enzymes, reinforcing the relevance of the imidazole removal after the protein purification steps.

## Results and Discussion

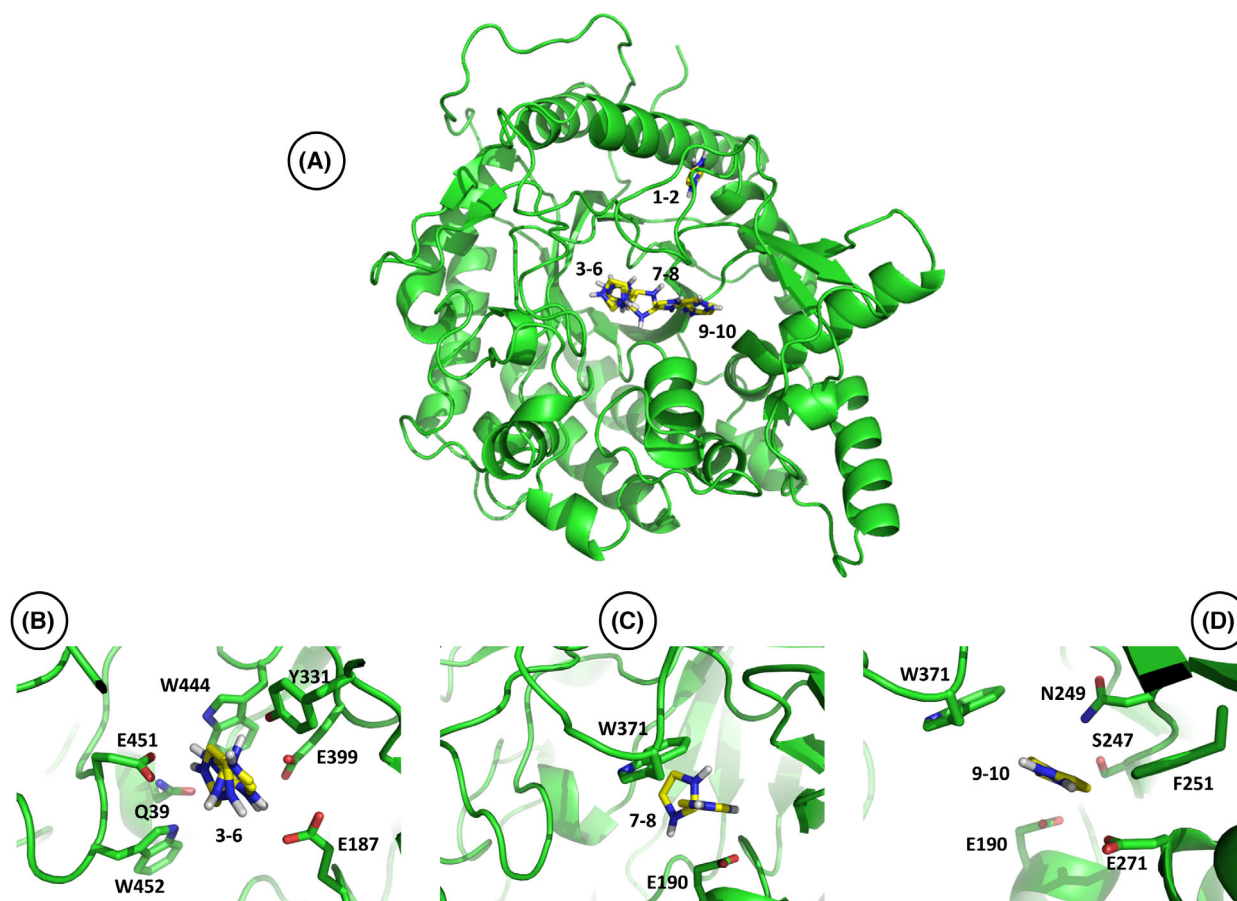
The potential interaction between imidazole and  $\beta$ -glucosidase S $\beta$ gly was initially evaluated by computational docking. The 10 best interaction solutions identified in the Autodock search included eight different orientations of the imidazole ring within the –1 and +1 subsites of the active site (solutions 3–10), and finally, two binding spots between residues S378 and

S424 (solutions 1 and 2), which are part of a contact network that indirectly affects the S $\beta$ gly substrate specificity [19] (Table 1; Fig. 1A). The structural analysis of the solutions 3–10, within the active site, showed residues Q39, E187, E190, S247, N249, F251, E271, Y331, W371, E399, W444, E451, and W452 at 3.5 Å from N<sub>1</sub> or N<sub>3</sub> of the imidazole ring, suggesting that they may form noncovalent interactions (Table 1). Residues E187 and E399 are the S $\beta$ gly catalytic residues, Q39 and E451 form hydrogen bonds with the glycone OH4 of the substrate and W444 is the basal platform of the –1 subsite, which also forms contacts with the substrate glycone [6] (Fig. 1B), whereas residues E190 and W371 are part of the subsite +1 that interacts with the substrate aglycone [25] (Fig. 1C). Finally, residues S247, N249, F251, and E271, seen in the solutions 9 and 10, form a pocket very close to the subsite +1 within the active site opening (Fig. 1D). These residues are also central in the S $\beta$ gly noncovalent interaction network [21].

In short, the docking solutions confirm the imidazole interaction within and close to the  $\beta$ -glucosidase S $\beta$ gly active site is a reasonable possibility, which then could have effects on the enzyme activity.

**Table 1.** Docking solutions for imidazole binding in the GH1  $\beta$ -glucosidase S $\beta$ gly.

| Autodock solution | Binding score | S $\beta$ gly residue within 3.5 Å of the imidazole nitrogen atoms | S $\beta$ gly region               |
|-------------------|---------------|--|------------------------------------|
| 1                 | –4.0          | S378, I379, and S424   | Surface                            |
| 2                 | –3.8          | S378, I379, and S424   | Surface                            |
| 3                 | –3.6          | Q39, E187, E451, and W452  | Active site; –1 subsite            |
| 4                 | –3.6          | E187, Y331, W371, E399, and E451                                   | Active site; –1 and +1 subsites    |
| 5                 | –3.6          | E187, W371, E399, W444, and E451                                   | Active site; –1 and +1 subsites    |
| 6                 | –3.4          | E187, Y331, W371, E399, W444, and E451                             | Active site; –1 and +1 subsites    |
| 7                 | –3.2          | E187, E190, S247, and N249   | Active site; +1 subsite and pocket |
| 8                 | –3.1          | W143, E187, E190, and Y331   | Active site; –1 and +1 subsite     |
| 9                 | –3.0          | R189, E190, S247, N249, and F251                                   | Active site; +1 subsite and pocket |
| 10                | –3.0          | R189, E190, S247, N249, F251, and E271                             | Active site; +1 subsite and pocket |



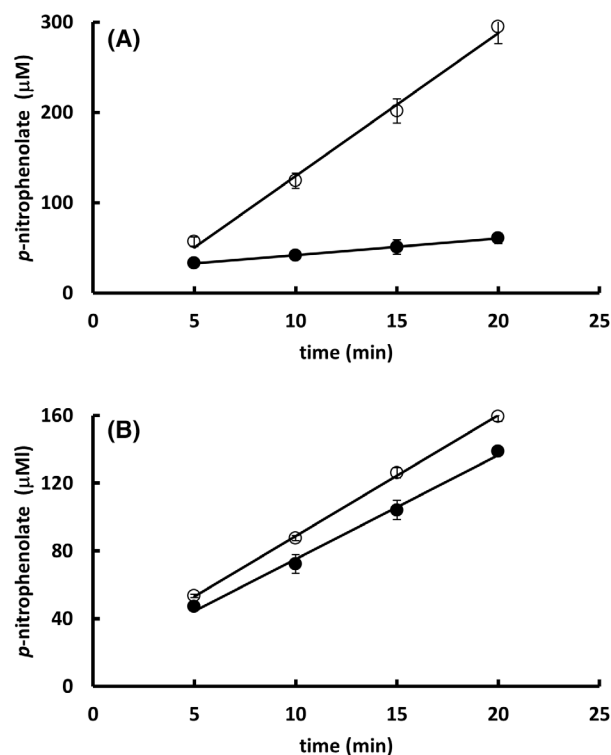
**Fig. 1.** Structural mapping of the docking solutions for imidazole binding in the GH1  $\beta$ -glucosidase Sf $\beta$ gly. (A) Superposition of the 10 best docking solutions. Solutions are numbered as presented in Table 1. (B) Different orientations of the imidazole within the Sf $\beta$ gly active site (subsite -1) observed in the docking solutions 3–6. (C) Imidazole binding in the Sf $\beta$ gly active site (subsite +1) observed in the docking solutions 7 and 8. (D) Imidazole interaction in the pocket close to the active site opening presented in the docking solutions 9 and 10. Residues Q39, E187, E190, S247, N249, F251, E271, Y331, W371, E399, W444, E451, and W452 are at 3.5 Å of the imidazole N<sub>1</sub> or N<sub>3</sub>.

Indeed, we observed that 150 mM imidazole, when present in the reaction mix, reduced the initial rate of substrate hydrolysis catalyzed by the Sf $\beta$ gly to 12% (Fig. 2A). That inhibitory effect does not arise from enzyme inactivation since the previous incubation of Sf $\beta$ gly at 30 °C with 150 mM imidazole for 18 h followed by an activity assay in ‘imidazole-free’ conditions did not significantly change the Sf $\beta$ gly activity (Fig. 2B).

The possibility that the inhibitory effect actually resulted from the participation of imidazole as an acceptor in transglycosylation reactions catalyzed by Sf $\beta$ gly was evaluated by following the formation of the products glucose and *p*-nitrophenolate from the substrate *p*-nitrophenyl  $\beta$ -glucoside. It was observed that those products are formed in a 1 : 1 ratio (4.4 and 4.5  $\mu\text{M}\cdot\text{min}^{-1}$ ; Fig. 3) in reactions performed without imidazole. This ratio also holds even when 150 mM

imidazole was present (2.7 and 2.9  $\mu\text{M}\cdot\text{min}^{-1}$ ; Fig. 3). Considering that the occurrence of transglycosylation reaction should reduce the rate of glucose production, which would be incorporated in the transglycosylation product, the observed 1 : 1 ratio indicates the absence of transglycosylation reactions in both conditions (Fig. 3).

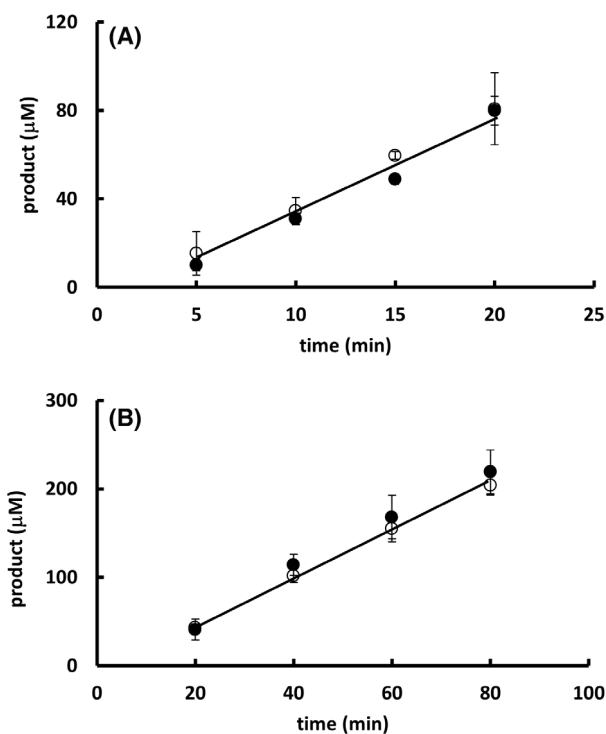
Hence, an enzyme kinetics approach was used to determine the mechanism of imidazole inhibition upon Sf $\beta$ gly. The initial rate of substrate hydrolysis was determined in different *p*-nitrophenyl  $\beta$ -glucoside concentrations in the presence of several fixed imidazole concentrations. Lineweaver–Burk plots were used to analyze the data revealing a hyperbolic relation between the apparent  $K_s/k_3$  (calculated from the line slope) and imidazole concentration, whereas the apparent  $1/k_3$  (calculated from the line intercept) remained constant (Fig. 4A–D). Those are features of a partial



**Fig. 2.** Effect of imidazole on the activity of the  $\beta$ -glucosidase SfBgl. (A) Determination of the initial rate of the 15 mM *p*-nitrophenyl  $\beta$ -glucoside hydrolysis catalyzed by SfBgl in the absence (O) and presence (●) of 150 mM imidazole. Initial rates are 16 and 2  $\mu$ M $\cdot$ min $^{-1}$  in the absence and presence imidazole, respectively. SfBgl concentration was 0.1  $\mu$ M. (B) Determination of the initial rate of the 15 mM *p*-nitrophenyl  $\beta$ -glucoside hydrolysis catalyzed by SfBgl previously incubated at 30 °C for 18 h with 150 mM imidazole (●) and without imidazole (O). Rates are 6 and 7  $\mu$ M $\cdot$ min $^{-1}$ , respectively. SfBgl concentration was 0.05  $\mu$ M. Data are mean and standard deviation of three determinations of the product formed in each incubation time using three separate assays with the same enzyme sample. The substrate was prepared in 100 mM phosphate buffer pH 6.0. Activity assays were performed at 30 °C.

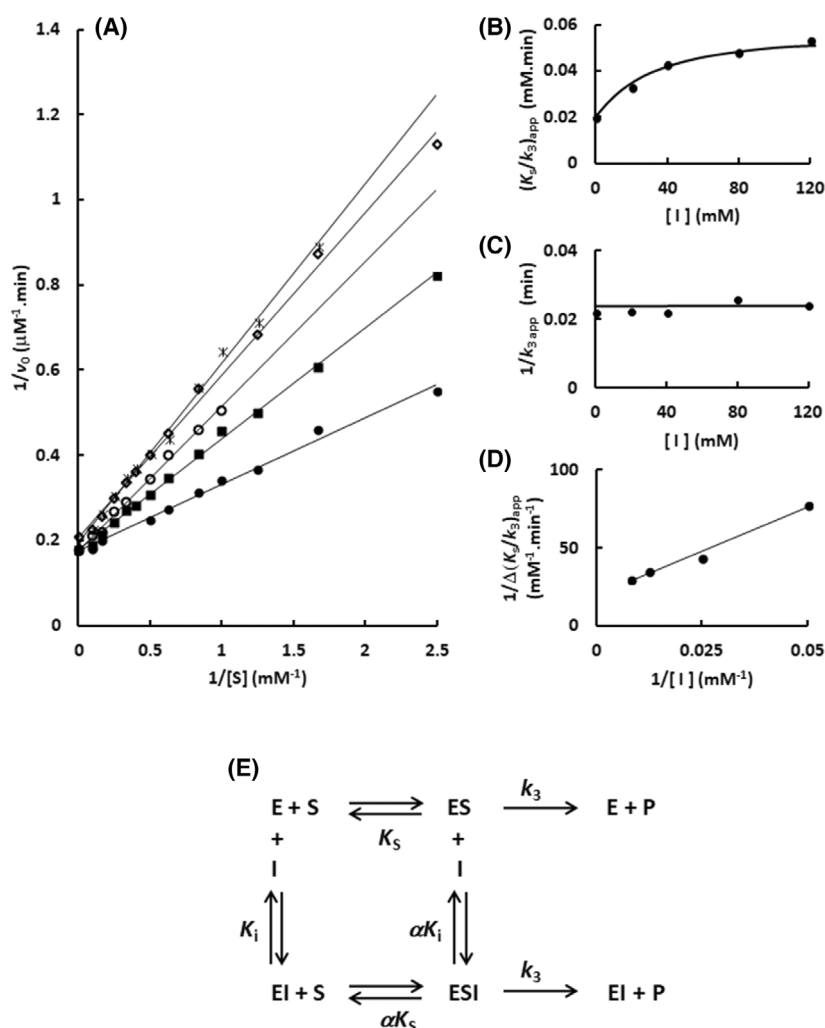
competitive inhibition mechanism (simple intersecting hyperbolic competitive; [26]), which is depicted in Fig. 4E.

In this partial competitive inhibition mechanism, the constant line intercept in the  $1/v_0 \times 1/[S]$  plot (which is proportional to  $1/k_3$ ) indicates that the *p*-nitrophenyl  $\beta$ -glucoside and imidazole may bind simultaneously in the enzyme forming a ternary complex ESI that is as productive as the ES complex, that is, exhibits the same  $k_3$ . However, the imidazole interaction with the enzyme reduces the affinity for the substrate, that is, increases the  $K_S$  by a factor termed  $\alpha$  ( $\alpha > 1$ ). The reverse is also true. The substrate binding reduces the affinity between the enzyme and imidazole, that is, also increases the  $K_i$



**Fig. 3.** Evaluation of transglycosylation reactions catalyzed by the  $\beta$ -glucosidase SfBgl. (A) Determination of the activity of the SfBgl followed by the production of *p*-nitrophenolate (O) and glucose (●). (B) Determination of the activity of the SfBgl followed by the production of *p*-nitrophenolate (O) and glucose (●) in the presence of 150 mM imidazole. The substrate was 15 mM *p*-nitrophenyl  $\beta$ -glucoside prepared in 100 mM phosphate buffer pH 6.0. SfBgl concentration was 0.05  $\mu$ M. Data are mean and standard deviation of three determinations of the product formed in each incubation time using the same enzyme sample. Activity assays were performed at 30 °C.

by the same  $\alpha$  factor. Moreover, as the imidazole concentration increases, the complete enzyme population is driven to the EI and the ESI complexes, which bring about the same  $V_{\max}$  but a higher  $K_S$ . Therefore, the imidazole inhibitory effect results from the apparent lower substrate affinity. Thus, as the enzyme population shifts from E and ES to the EI and ESI states, the slope (which is proportional to  $K_S/k_3$ ) of the lines in the Lineweaver–Burk plot increases and reaches a maximum ( $\alpha K_S/k_3$ ), producing the already mentioned hyperbolic curve in the  $K_S/k_3$  app vs [imidazole] plot, whereas the  $V_{\max}$  remains constant as shown in the  $1/k_3$  app vs [imidazole] plot (Fig. 4). Based on the expression describing the line slope in the Lineweaver–Burk plot [26], a secondary plot ( $1/\Delta_{K_S/k_3}$  app vs  $1/[\text{imidazole}]$ ) was produced, from which the  $K_i$  for imidazole,  $15.3 \pm 0.3$  mM, and its effect on the  $K_S$ ,  $\alpha = 3 \pm 1$ , were evaluated (Fig. 4; Figs S2–S4).



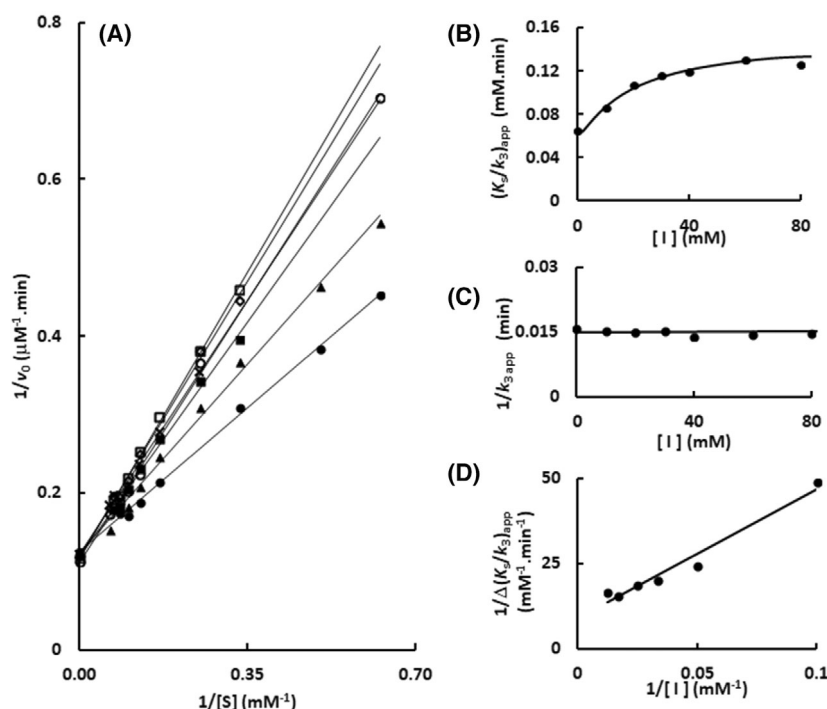
**Fig. 4.** Determination of the mechanism of Sf $\beta$ gly inhibition by imidazole using the substrate *p*-nitrophenyl  $\beta$ -glucoside. (A) Lineweaver–Burk plots representing the effect of the *p*-nitrophenyl  $\beta$ -glucoside concentration on the initial rate of substrate hydrolysis in the presence of different imidazole concentrations: (●) 0; (■) 20 mM; (○) 40 mM; (◇) 80 mM; (\*) 120 mM. (B) Effect of the imidazole concentration on the apparent  $K_s/k_3$  (calculated from the line slope in the Lineweaver–Burk plot). (C) Effect of the imidazole concentration on the  $1/k_3$  apparent (calculated from the line intercept in the Lineweaver–Burk plot). (D) Secondary plot ( $1/\Delta(K_s/k_3)_{app}$  versus  $1/[I]$ ) representing the effect of the imidazole concentration on the apparent  $K_s/k_3$ .  $\Delta(K_s/k_3)_{app} = K_s/k_{3i} - K_s/k_{30}$ , in which  $K_s/k_{3i}$  corresponds to the line in the presence of imidazole and  $K_s/k_{30}$  to the line in the absence of imidazole. The substrate and inhibitor were prepared in 100 mM phosphate buffer pH 6.0. Activity assays were performed at 30 °C. The results presented above correspond to the average of experiments repeated in the same conditions with three different enzyme samples (0.125  $\mu\text{M}$ ; Figs S1–S4). (E) Schematic representation of the partial competitive inhibition mechanism observed for imidazole. I corresponds to imidazole; S represents the substrate. E is the GH1  $\beta$ -glucosidase Sf $\beta$ gly. P is the product.  $K_s$  is the dissociation constant for the enzyme–substrate (ES) complex.  $K_i$  is the dissociation constant for the enzyme–inhibitor (EI) complex.  $k_3$  is the rate constant for the product formation from the ES complex. The  $\alpha$  factor represents the hindering effect between the substrate and inhibitor ( $\alpha > 1$ ), based on [26].

The same approach was used to characterize the imidazole inhibitory mechanism upon different substrates. These substrates range from cellobiose (C2; Fig. 5; Figs S5–S7), which occupies the subsites –1 and +1 similarly to the *p*-nitrophenyl  $\beta$ -glucoside, to a larger oligocellodextrin, cellotetraose (C4; Fig. 6; Figs S8–S10), which fill subsites beyond +1, even

reaching the active site opening. For comparative purpose, the kinetic parameters for hydrolysis of those substrates are reported in Table S1.

The same imidazole inhibitory mechanism,  $K_i$  and  $\alpha$  were observed for these oligocellodextrins (Table 2; Figs 4–6; Figs S2–S10). Indeed, the extension of the substrate, that is, the subsites filled, did not change the





**Fig. 5.** Determination of the mechanism of Sfbgly inhibition by imidazole using the substrate cellobiose. (A) Lineweaver–Burk plots representing the effect of the cellobiose concentration on the initial rate of substrate hydrolysis in the presence of different imidazole concentrations: (●) 0; (▲) 10 mM; (■) 20 mM; (×) 30 mM; (○) 40 mM; (□) 60 mM; (◇) 80 mM. (B) Effect of the imidazole concentration on the apparent  $K_s/k_3$  (calculated from the line slope in the Lineweaver–Burk plot). (C) Effect of the imidazole concentration on the apparent  $1/k_3$  (calculated from the line intercept in the Lineweaver–Burk plot). (D) Secondary plot ( $1/\Delta(K_s/k_3)_{\text{app}}$  versus  $1/[I]$ ) representing the effect of the imidazole concentration on the apparent  $K_s/k_3$ .  $\Delta(K_s/k_3)_{\text{app}} = K_s/k_{3i} - K_s/k_{30}$ , in which  $K_s/k_{3i}$  corresponds to the line in the presence of imidazole and  $K_s/k_{30}$  to the line in the absence of imidazole. The substrate and inhibitor were prepared in 100 mM phosphate buffer pH 6.0. Activity assays were performed at 30 °C. The results presented above correspond to the average of experiments repeated in the same conditions with three different enzyme samples (0.125  $\mu\text{M}$ ; Figs S1, S5 and S7).

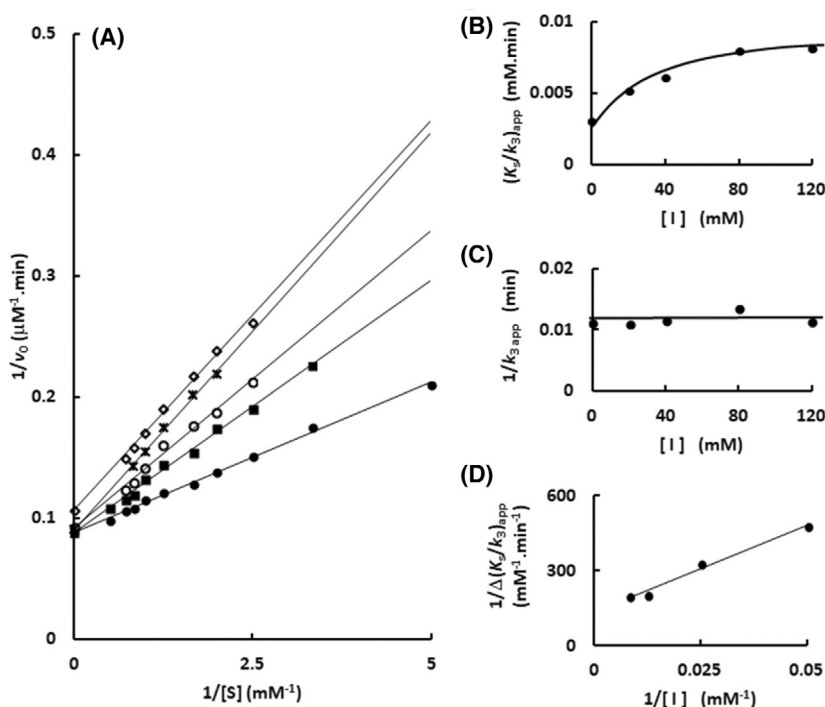
mutual hindrance between imidazole and substrate. Therefore, imidazole binds at the same region of the Sfbgly active site in the presence of these different substrates.

Next, the interaction of the imidazole with the Sfbgly was investigated using a different approach, which is depicted in Fig. 7. The enzyme and substrate were simultaneously combined with imidazole and a competitive inhibitor. Assuming that both bind in the active site, the presence of the competitive inhibitor should hinder the imidazole binding, hence altering its inhibitory effect upon the substrate hydrolysis. On the contrary, if imidazole does not bind within the active site, the presence of a competitive inhibitor would not influence the imidazole inhibitory effect. Hence, in this approach, we seek to determine the effect that cellobiose, as a competitive inhibitor, exerts upon the imidazole inhibitory effect.

Cellobiose was used as this second competitive inhibitor because, as a natural substrate, it would

surely bind in the Sfbgly active site. Moreover, when the *p*-nitrophenyl  $\beta$ -glucoside is the substrate and the initial rate of reaction is determined by following the formation of *p*-nitrophenolate, cellobiose behaves as a competitive inhibitor. Indeed, that was demonstrated and the  $K_i$  for cellobiose is 10 mM (Fig. S11). Finally, it should be noted that the  $k_3/K_s$  for the cellobiose hydrolysis is about 3 times lower than the *p*-nitrophenyl  $\beta$ -glucoside (Table S1). Hence, the initial rate condition applies to both, indicating that cellobiose concentration is also approximately constant during the experiment time.

The initial rates of the hydrolysis of several *p*-nitrophenyl  $\beta$ -glucoside concentrations were determined in the presence of increasing fixed imidazole concentration, similarly to the experiments reported above (Fig. 4). However, now these experiments were performed in the presence of constant cellobiose concentration, 5, 10, 20 and 30 mM that correspond to 0.5, 1, 2, and 3  $K_i$  (Figs S12–S15). So, these cellobiose



**Fig. 6.** Determination of the mechanism of Sfbgly inhibition by imidazole using the substrate cellotetraose. (A) Lineweaver–Burk plots representing the effect of the cellotetraose concentration on the initial rate of substrate hydrolysis in the presence of different imidazole concentrations: (●) 0; (■) 20 mM; (○) 40 mM; (◇) 80 mM; (×) 120 mM. (B) Effect of the imidazole concentration on the apparent  $K_s/k_3$  (calculated from the line slope in the Lineweaver–Burk plot). (C) Effect of the imidazole concentration on the apparent  $1/k_3$  (calculated from the line intercept in the Lineweaver–Burk plot). (D) Secondary plot ( $1/\Delta(K_s/k_3)_{app}$  versus  $1/[I]$ ) representing the effect of the imidazole concentration on the apparent  $K_s/k_3$ .  $\Delta(K_s/k_3)_{app} = K_s/k_{3i} - K_s/k_{30}$ , where  $K_s/k_{3i}$  corresponds to the line in the presence of imidazole and  $K_s/k_{30}$  to the line in the absence of imidazole. The substrate and inhibitor were prepared in 100 mM phosphate buffer pH 6.0. Activity assays were performed at 30 °C. The results presented above correspond to the average of experiments repeated in the same conditions with three different enzyme samples (0.125  $\mu$ M; Figs S1, S8 and S10).

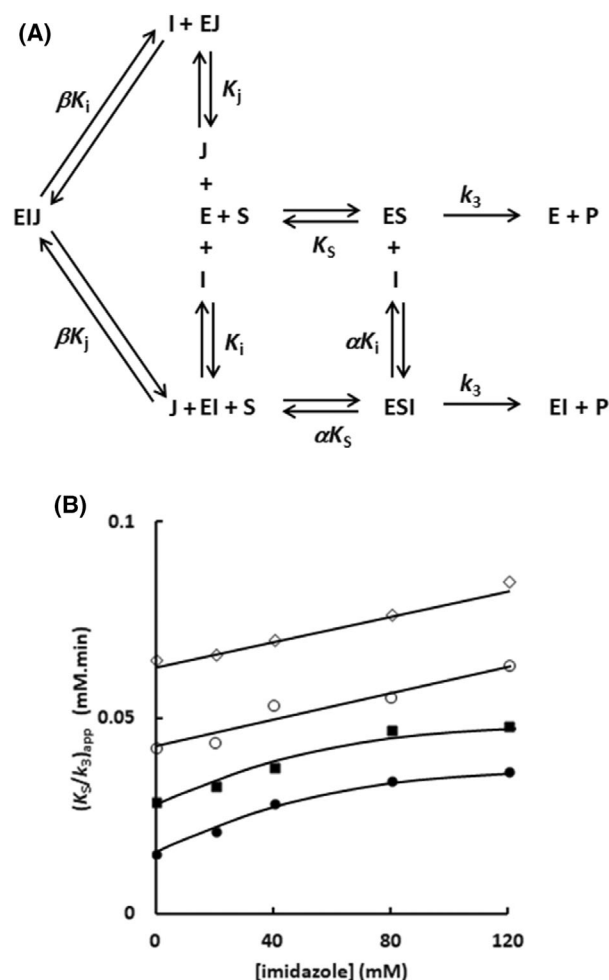
**Table 2.** Parameters of the imidazole inhibitory mechanism upon the hydrolysis of different substrates by the Sfbgly. The inhibition mechanism is partial competitive as depicted in the Fig. 4E. The  $\alpha$  factor represents the mutual hindrance between imidazole and substrate.  $K_i$  is the dissociation constant for the enzyme–imidazole complex. Data are mean and standard deviation based on three independent experiments (Figs S2–S10). NP $\beta$ glc, *p*-nitrophenyl  $\beta$ -glucoside; C2, cellobiose; C4, cellotetraose. Kinetic parameters for the substrate hydrolysis are reported in the Table S1.

| Substrate      | $K_i$ (mM)     | $\alpha$      |
|----------------|----------------|---------------|
| NP $\beta$ glc | $15.3 \pm 0.3$ | $3 \pm 1$     |
| C2             | $14 \pm 3$     | $2.3 \pm 0.1$ |
| C4             | $15 \pm 5$     | $2.8 \pm 0.7$ |

concentrations should produce a gradient of increasing concentration of the Sfbgly–cellobiose complex, that is, the EJ complex in Fig. 7A, diverting the enzyme away from the complexes with imidazole (EI and ESI). Again, we analyzed the imidazole inhibitory effect by following the changes in the apparent  $K_s/k_3$  calculated

from the line slopes in the Lineweaver–Burk plots (Fig. 7B and Figs S12–S15).

At 5 mM cellobiose (0.5  $K_i$ ), the imidazole inhibitory effect is observed again as the hyperbolic behavior of the apparent  $K_s/k_3$  vs [imidazole] plot. However, at 30 mM cellobiose ( $3K_i$ ) the apparent  $K_s/k_3$  increases linearly with imidazole concentration (Fig. 7B). Hence the increment of the cellobiose concentration is shifting the inhibitory effect away from the partial competitive mechanism. Indeed, that is expected if both inhibitors, cellobiose and imidazole, were competing for the same spot in the active site. The increment of cellobiose concentration would saturate the enzyme, hampering the imidazole binding (Fig. 7A). Moreover, in a saturating cellobiose concentration, the lines in the Lineweaver–Burk plot (Figs S12–S15) would tend to the same apparent  $K_s/k_3$ , that is, slope, because the dominant inhibitory effect would result from the same cellobiose concentration. In agreement, the ratio between the apparent  $K_s/k_3$  in the absence and in the presence of the higher imidazole concentration is about 2.5 at



**Fig. 7.** Simultaneous effect of imidazole and cellobiose on the Sfbglg hydrolysis of the substrate. (A) Schematic representation of the simultaneous effect of imidazole (*I*; a partial competitive inhibitor) and cellobiose (*J*; a competitive inhibitor) on the substrate (*S*; *p*-nitrophenyl  $\beta$ -glucoside) hydrolysis. E, GH1  $\beta$ -glucosidase Sfbglg; P—product.  $K_s$  is the dissociation constant for the enzyme–substrate complex (ES).  $K_i$  is the dissociation constant for the enzyme–imidazole complex (EI).  $K_j$  is the dissociation constant for the enzyme–cellobiose complex (EJ). The  $\alpha$  factor represents the hindering effect between the substrate and imidazole (Table 2). The  $\beta$  factor represents the hindering effect between imidazole and cellobiose. (B) Effect of the imidazole concentration on the apparent  $K_s/k_3$  (calculated from the line slope in the Lineweaver–Burk plot) in the presence of constant cellobiose concentration. The initial rates of the hydrolysis of several *p*-nitrophenyl  $\beta$ -glucoside concentrations were determined in the presence of different imidazole concentrations. The constant cellobiose concentrations were (●) 5, (■) 10, (○) 20, and (◇) 30 mM, which correspond to 0.5, 1, 2, and 3  $K_j$ . Lineweaver–Burk plots resulting from these experiments are presented in Figs S12–S15. The substrate and inhibitors were prepared in 100 mM phosphate buffer pH 6.0. Activity assays were performed at 30 °C.

5 mM cellobiose, but it decreases to 1.3 at 30 mM cellobiose (Fig. 7B). In short, the presence of cellobiose is abolishing the imidazole inhibitory effect.

Next, we moved to quantify the mutual hindering between cellobiose and imidazole when both act as inhibitors in the experiment reported above (Fig. 7; Figs S12–S15). The value of such parameter, which was termed  $\beta$  in Fig. 7A, should be comparable to  $\alpha$ , previously evaluated when using the imidazole as an inhibitor of the substrate hydrolysis (Table 2). Based on the scheme describing the simultaneous interaction of imidazole and cellobiose with Sfbgl (Fig. 7A), we deduced the rate equation below (Fig. S16) describing the lines in the Lineweaver–Burk plot presented in Figs S12–S15.

$$\frac{1}{v_o} = \frac{K_s}{V_{\max}} \frac{\left( \alpha K_i + \alpha [I] + \frac{\alpha K_i [J]}{K_j} + \frac{\alpha [I] [J]}{\beta K_j} \right)}{\alpha K_i + [I]} \frac{1}{[S]} + 1/V_{\max}. \quad (1)$$

$J$  is cellobiose,  $I$  is imidazole,  $\alpha$  represents the mutual impediment between  $S$  and imidazole (Table 2) and  $\beta$  corresponds to the mutual impediment between cellobiose and imidazole. The substrate  $S$  is  $p$ -nitrophenyl  $\beta$ -glucoside.  $K_s$  is the dissociation constant for the enzyme–substrate complex.  $K_i$  is the dissociation constant for the enzyme–imidazole complex (Table 2).  $K_j$  is the dissociation constant for the enzyme–cellobiose complex (Fig. S11).

As seen, the slope of this linear equation (i.e., apparent  $K_s/V_{\max}$ ) describes both inhibitor effects on the enzyme, whereas the intercept ( $1/V_{\max}$ ) does not depend on their concentration. Hence, the slope was isolated and expressed in two limiting situations: infinite concentration of imidazole (slope $_{\infty}$ ) and absence of imidazole (slope $_0$ ), both in the presence of the same cellobiose concentration.

$$\text{slope}_\infty = \frac{K_s}{V_{\max}} \alpha \left( 1 + \frac{[J]}{\beta K_j} \right) \quad (2)$$

$$\text{slope}_0 = \frac{K_s}{V_{\max}} \left( 1 + \frac{[J]}{K_i} \right) \quad (3)$$

The ratio  $\text{slope}_\infty/\text{slope}_0$  results in the expression below, in which the  $\beta$  describes the mutual impediment between cellobiose and imidazole ( $J$  and  $I$ , respectively; Fig. 7A).

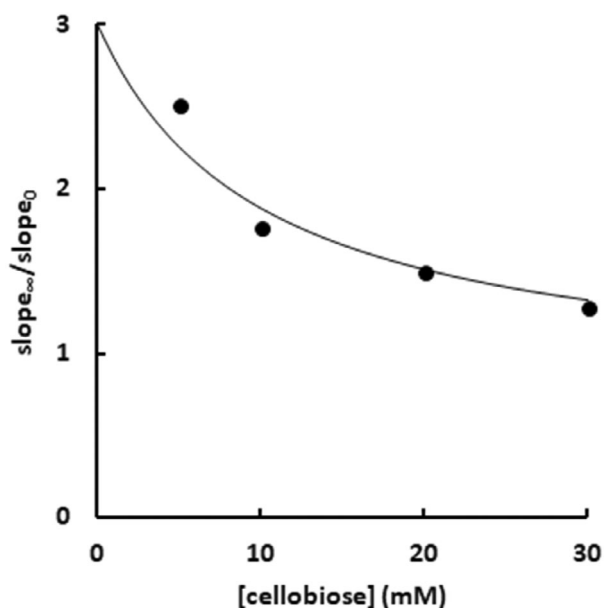
$$\frac{\text{slope}_\infty}{\text{slope}_0} = \frac{\alpha}{\beta} \frac{(\beta K_j + [J])}{(K_j + [J])} \quad (4)$$



Considering that parameters  $\alpha$  and  $K_i$  were previously determined (Table 2 and Fig. S11), Eqn (4) above was fitted into the  $\text{slope}_\infty/\text{slope}_0$  and [cellobiose] ( $[J]$ ) data (extracted from Fig. 7B; see [Materials and methods](#) for further details) to estimate  $\beta$  as  $3.9 \pm 0.6$  (Fig. 8).

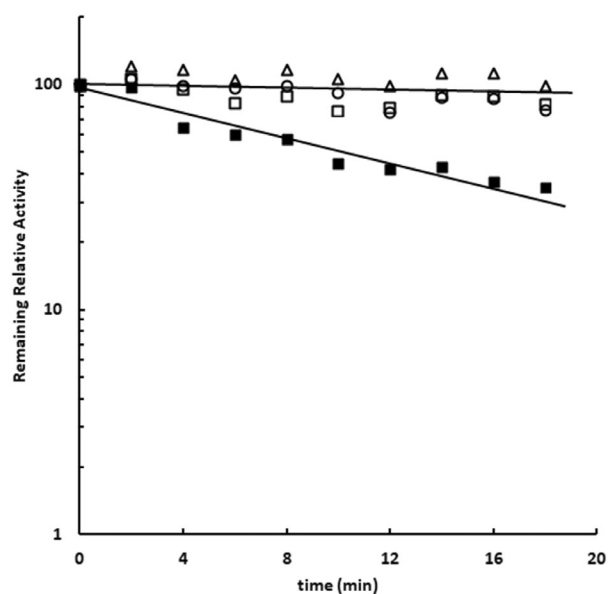
This  $\beta$  estimate, higher than 1, but lower than infinite, indicates a mutual hindrance effect on the cellobiose and imidazole binding, but they are not mutually exclusive ligands. Moreover,  $\beta$  is similar to the  $\alpha$ , previously determined from the imidazole inhibitory effect upon the substrate hydrolysis (Table 2). Thus, both approaches, the inhibition of the substrate hydrolysis with imidazole and the competition between imidazole and a second inhibitor (Figs 4–8), are reporting the same hindering effect between cellobiose and imidazole. Therefore, recalling that cellobiose definitely interacts with the active site, both experiments showed unequivocally that imidazole also binds in the S $\beta$ gly active site.

Finally, a third approach was devised to test the imidazole binding in the S $\beta$ gly active site. Carboxylic groups from the side chain of glutamic acid residues can react with carbodiimides forming an ester that promptly reacts with primary amines at acidic pH [27]. The catalytic activity of the GH1  $\beta$ -glucosidases depends on two glutamic acid residues within their active site. For S $\beta$ gly these residues are E187 and



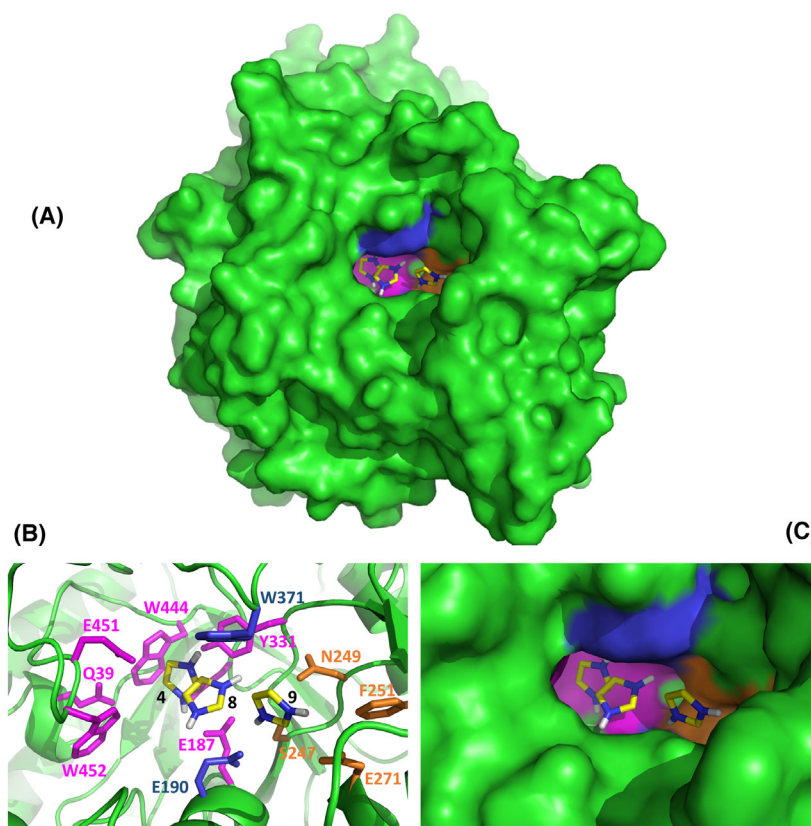
**Fig. 8.** The ratio  $\text{slope}_\infty/\text{slope}_0$  for different cellobiose concentrations. The  $\text{slope}_\infty$  and  $\text{slope}_0$  data were extracted from Fig. 7B (see [Materials and methods](#) for further details). The line represents the best fit of the data in Eqn (4) ( $R^2 = 0.96$ ).

E399 [6]. Hence, 1-ethyl-3-(dimethylamino-propyl) carbodiimide (EDC) can prompt the reaction of these two E residues with glycine ethyl ester (GEE), converting their side chains in an amide and resulting in the S $\beta$ gly inactivation [27]. The kinetics of the S $\beta$ gly inactivation in the presence of 12 mM EDC and 40 mM GEE was determined (Fig. 9) showing an observable rate constant ( $k_{\text{obs}}$ ) of  $0.024 \text{ min}^{-1}$ . A control reaction containing only the enzyme showed that S $\beta$ gly is stable in those conditions. Interestingly, the addition of 40 mM cellobiose ( $4K_i$ ) to the reaction mix abolished the enzyme inactivation promoted by EDC, suggesting that the cellobiose binding hinders the EDC and GEE accesses to the active site, protecting the enzyme from inactivation. Similarly, the addition of 60 mM imidazole ( $4K_i$ ) to that modification reaction also protected S $\beta$ gly from inactivation (Fig. 9). Hence, the saturating concentration of both, cellobiose and imidazole, hampers the access of EDC and GEE to the S $\beta$ gly catalytic residues, confirming that imidazole binding spot is within the S $\beta$ gly active site.



**Fig. 9.** Inactivation of S $\beta$ gly with carbodiimide and glycine ethyl ester (GEE). The inactivation was performed in the presence of 40 mM cellobiose ( $\Delta$ ), 60 mM imidazole ( $\circ$ ), and without any ligand ( $\blacksquare$ ). A control experiment, containing only the enzyme prepared in 100 mM phosphate buffer pH 6.0, was also performed ( $\square$ ). The 12 mM EDC and 40 mM GEE were prepared in 100 mM phosphate buffer pH 6.0. Reactions were performed at 30 °C. The remaining activity was determined by using 15 mM *p*-nitrophenyl  $\beta$ -glucoside as substrate. Substrate and ligands were also prepared in 100 mM phosphate buffer pH 6.0. Activity assays were performed at 30 °C. The  $k_{\text{obs}}$  of the inactivation reaction was calculated based on the line slope.

**Fig. 10.** Imidazole binding within different regions of the Sfbgly active site. (A) General perspective of the Sfbgly molecular surface showing the active site niche containing the imidazole docking solutions 4, 8, and 9 (Table 1). These solutions illustrate the binding in the subsite  $-1$  (pink), subsite  $+1$  (blue), and the pocket in the active site entrance (orange). (B) Detailed structural features of the imidazole docking solutions 4, 8, and 9 within the subsite  $-1$  (pink), subsite  $+1$  (blue), and the pocket in the active site entrance (orange), respectively. Residues shown as sticks are within 3.5 Å from the N<sub>1</sub> or N<sub>3</sub> of the imidazole. (C) Molecular surface of the active site niche showing the imidazole docking solutions 4, 8, and 9 within the subsite  $-1$  (pink), subsite  $+1$  (blue), and the pocket in the active site entrance (orange), respectively. Panels B and C are different representations of the same view.



Therefore, the different approaches presented above converged showing that imidazole binds to the Sfbgly active site and reduces the substrate affinity. Taking into consideration the indications of the molecular docking initially presented (Fig. 1; Table 1), the imidazole could be occupying the inner portions of the active site or even its entrance (Figs 1B–D and 10A). These three binding spots are properly represented in the docking solutions 4, 8, and 9 (Fig. 10B). Nevertheless, the imidazole docking in the  $-1$  subsite seems too restrictive for the simultaneous imidazole and substrate interactions (Figs 1B and 10B,C). Conversely, the imidazole binding in the entrance of the active site, a wider region involving the  $+1$  subsite and the lateral pocket (Figs 1C,D and 10B,C), could weaken the substrate binding but keep the enzyme equally active once the substrate–enzyme complex were formed, that is, it would result in a productive ESI complex (Fig. 4). Indeed, the structural superimposition of the active sites of the Sfbgly with the imidazole docking solutions 4, 8, and 9 and GH1  $\beta$ -glucosidases complexed with a substrate (PDB 3A10, 2O9R, and 2Z1S; Fig. S17), suggests that the imidazole binding within the lateral pocket in the entrance of the active site (Fig. 10B,C) is compatible with an ESI complex. Hence, this binding

spot, the pocket nearby the active site opening (docking solutions 9 and 10 in Table 1; Figs 1D and 10B, C), is an attractive hypothesis to explain our results.

Finally, considering that residues forming the active site are conserved among the GH1  $\beta$ -glucosidases, this imidazole inhibition may be widespread within this enzyme family. Hence, this information should be taken into account particularly in the characterization of the catalytic activity and enzyme kinetic parameters of recombinant GH1  $\beta$ -glucosidases purified using imidazole.

## Materials and methods

### Production and purification of the recombinant Sfbgly

The recombinant Sfbgly was produced in BL21(DE3) bacteria using the expression vector pET46 as previously described in [23]. Following that, recombinant Sfbgly was purified by affinity to the Ni-NTA agarose resin (Qiagen) as also previously described in [23]. The homogeneity of the recombinant Sfbgly was checked in SDS/PAGE [28] (Fig. S1). Protein concentration was determined by the bicinchoninic acid method [29]. The purified Sfbgly sample was submitted to

buffer exchange by using PD Minitrap G-25 columns (Cytiva, Marlborough, MA, USA). After that, this final sample in 100 mM phosphate buffer pH 6 was stored at 4 °C.

### Determination of the S $\beta$ gly catalytic activity

The S $\beta$ gly hydrolysis activity upon *p*-nitrophenyl  $\beta$ -glucoside was determined in reactions (100  $\mu$ L) performed at 30 °C and interrupted by the addition of 0.5 M Na<sub>2</sub>CO<sub>3</sub> (100  $\mu$ L). The product (*p*-nitrophenolate) formation was followed by absorbance at 415 nm. Absorbance was converted into mol by using standard curves prepared in the same conditions as the enzymatic reactions. Briefly, different amounts (nmols) of *p*-nitrophenol prepared in 100 mM phosphate buffer pH 6 (final volume of 100  $\mu$ L) were combined with 0.5 M Na<sub>2</sub>CO<sub>3</sub> (100  $\mu$ L). Their absorbance at 415 nm and amounts (nmols) of *p*-nitrophenol were plotted and submitted to linear regression. The line equation was used to convert the absorbance into mol of *p*-nitrophenol (product) formed along the enzyme assays.

The S $\beta$ gly hydrolysis activity upon cellobiose and cellotetraose was determined by following the formation of glucose. Reactions (100  $\mu$ L) were performed at 30 °C and interrupted by boiling (3 min). Glucose formation was detected with the Trinder kit (Analisa, Belo Horizonte, Brazil). This kit is based on the oxidation of glucose catalyzed by the enzyme glucose oxidase, which also produces hydrogen peroxide. That last product reacts with 4-aminoantipyrine and phenol resulting in a dye that absorbs at 490 nm. Standard curves for glucose quantification were prepared by combining different amounts (nmols) of glucose (final volume of 100  $\mu$ L) prepared in 100 mM phosphate buffer pH 6 with the reaction solution (150  $\mu$ L) of Trinder kit. After 10 min at 30 °C their absorbance at 490 nm was determined. The amounts (nmols) of glucose and absorbance data were plotted and submitted to linear regression. The line equation was used to convert the absorbance at 490 nm into mol of glucose (product) formed along the enzyme assays.

The occurrence of transglycosylation reactions was evaluated by following the ratio of the two products, *p*-nitrophenolate and glucose, generated from 15 mM *p*-nitrophenyl  $\beta$ -glucoside. Reactions (100  $\mu$ L) were performed at 30 °C and interrupted by boiling (3 min). The reaction mix was divided into two equal aliquots, which were used for the product detection. The *p*-nitrophenolate and glucose were determined as indicated above.

Initial rates were determined based on the line slope in the [product] vs time plots. Linear regression was used to evaluate the data. Correlation coefficients higher than 0.95 were accepted.

The S $\beta$ gly hydrolysis activity upon 15 mM *p*-nitrophenyl  $\beta$ -glucoside was determined in the absence and presence of 150 mM imidazole. Assays were performed as described above. Substrate and imidazole were prepared in 100 mM phosphate buffer pH 6.

The effect of the imidazole on the S $\beta$ gly stability was evaluated by incubating the enzyme in the presence and absence of 150 mM imidazole for 18 h at 30 °C. After that, the imidazole was washed away with 50 volumes of 100 mM phosphate buffer pH 6 by using centrifugal filter devices (Amicon Ultracel-10K, Millipore, Burlington, MA, USA). Next, the enzyme activity of these two samples (after removal of imidazole) was determined by using 15 mM *p*-nitrophenyl  $\beta$ -glucoside as previously described. Substrate and imidazole were prepared in 100 mM phosphate buffer pH 6.0.

In order to prevent changes in the buffer pH, buffering reagents were previously combined with imidazole and/or NaCl in the appropriate proportion to attain the concentration and pH of interest. Next, these reagents were dissolved, and after that, the buffer pH was set to 6.0. NaCl was added to the buffer in reactions without imidazole to normalize the ionic strength.

### Inhibition of the S $\beta$ gly activity with imidazole

The initial rate of hydrolysis of at least 10 different substrate concentrations was determined in the presence of at least five different imidazole concentrations (0–120 mM). Substrates were *p*-nitrophenyl  $\beta$ -glucoside (0.4–10 mM), cellobiose (0.6–16 mM), and cellotetraose (0.2–2 mM). Reactions were performed at 30 °C. Substrates and inhibitors were prepared in 100 mM phosphate buffer pH 6.0. The product formation (*p*-nitrophenolate and glucose) was detected as described above. NaCl was added to the inhibitor solution to normalize the ionic strengths among reactions, taking 120 mM imidazole as a referential. Initial rates were determined based on three product quantifications. Each complete experiment was performed with three different enzyme samples. Initial rates and substrate concentration data were analyzed using Lineweaver–Burk plots. Line slopes and intercepts correlation with inhibitor concentration was used to identify the inhibition mechanism [26]. Data were submitted to linear regression analysis. Fittings were accepted when showing linear correlation coefficients higher than 0.9. Kinetic parameters ( $K_i$  and  $\alpha$ ) were expressed as median and standard deviation resulting from three independent experiments.

### Inhibition of the S $\beta$ gly activity with cellobiose

The initial rate of hydrolysis of at least 10 different *p*-nitrophenyl  $\beta$ -glucoside (0.4–10 mM) concentrations was determined in the presence of five different cellobiose concentrations (0–45 mM). Reactions were performed at 30 °C. *p*-nitrophenyl  $\beta$ -glucoside and cellobiose were prepared in 100 mM phosphate buffer pH 6.0. The product formation (*p*-nitrophenolate) was detected as described above. Initial rates were determined based on three product quantifications using the same enzyme sample. Initial rates and substrate concentration data were analyzed using Lineweaver–

Burk plots. Data were submitted to linear regression analysis. Fittings were accepted when showing linear correlation coefficients higher than 0.9. Line slopes and intercepts correlations with cellobiose concentration were used to identify the inhibition mechanism [26].

### Simultaneous inhibition of Sfbgly activity with imidazole and a competitive inhibitor

The initial rate of hydrolysis of at least 10 different *p*-nitrophenyl  $\beta$ -glucoside concentrations (0.4–10 mM) was determined in the presence of five different imidazole concentrations (0–120 mM). Each of these experiments was performed in the presence of fixed cellobiose concentrations (5–30 mM), which was used as a competitive inhibitor. Reactions were performed at 30 °C. The product formation (*p*-nitrophenolate) was detected as described above. NaCl was added to the inhibitor solution to normalize the ionic strengths among reactions, taking 120 mM imidazole as a referential. Initial rates of the *p*-nitrophenyl  $\beta$ -glucoside hydrolysis were determined based on three product quantifications using the same enzyme sample. Initial rate condition also applies to the cellobiose hydrolysis. So, both *p*-nitrophenyl  $\beta$ -glucoside and cellobiose concentrations are considered constant in this experiment. Initial rates and substrate concentration data were analyzed using Lineweaver–Burk plots. The correlations between line slopes and imidazole concentration were used to evaluate the cellobiose effect on the inhibition mechanism. The slope in the absence of imidazole (slope<sub>0</sub>) was estimated from the intercept of those plots. On the contrary, the slope at infinite imidazole concentration (slope<sub>∞</sub>) was estimated from the intercept of the 1/line slope *vs* [imidazole]. Equation (4) was fitted into the slope<sub>∞</sub>/slope<sub>0</sub> and [cellobiose] by using nonlinear regression analysis in the Origin 2019 software.

### Inactivation of Sfbgly with carbodiimide

Inactivation reactions were performed by incubating Sfbgly in the presence of 12 mM EDC and 40 mM GEE at 30 °C [27]. Samples of the reaction mix were removed after different incubation times, combined with 100 mM citrate–phosphate buffer pH 6 to react with and remove the EDC excess. Next, these samples were employed to determine the remaining enzyme activity upon 15 mM *p*-nitrophenyl  $\beta$ -glucoside as described above. A sample of Sfbgly was incubated in the same conditions, but in the absence of EDC and GEE, to evaluate the enzyme stability. 40 mM cellobiose and 60 mM imidazole were separately added to the inactivation reaction to evaluate their enzyme protection effect. The enzyme inactivation was analyzed as a pseudo-first-order reaction by plotting the log(relative remaining activity) *vs* time. Hence, the observable rate constant of the enzyme inactivation (*k*<sub>obs</sub>) was estimated from the slope of that line. Substrate, EDC and GEE, cellobiose, and

imidazole were prepared in 100 mM phosphate buffer pH 6.0. Data were submitted to linear regression analysis. Fittings were accepted when showing linear correlation coefficients higher than 0.9.

### Computational docking between imidazole and Sfbgly

Sfbgly initial coordinates were taken from the PDB entry 5CG0. The imidazole 3D coordinates were generated in Gabedit v. 2.5.1 [30]. The addition of hydrogen atoms, and deletion of water molecules, ions, and other molecules from the protein initial structure were carried out using Chimera v. 1.14 [31], as well as the grid box coordinates covering all protein atoms, as previously described [32]. Docking of imidazole and Sfbgly was performed using AutoDock Vina v. 1.1.2 [33] launched from Chimera searching the 10 best models. The imidazole was docked in the mono-protonated (+1 charge) state. The Sfbgly–imidazole complexes representing the different docking solutions were visualized by using the PYMOL v 0.99 software (Schrödinger LLC, New York, NY, USA).

The crystallographic structures of the  $\beta$ -glucosidases from *Neotermes koshunensis* and *Paenibacillus polymyxa* (PDB ID 3A10, 2O9R, 2Z1S, respectively) containing the substrates NPbglc, thiocellobiose and C4 were structurally superposed to the Sfbgly (PDB ID 5CG0). Next, the structure of the  $\beta$ -glucosidases from *N. koshunensis* and *P. polymyxa* were removed leaving only the substrate overlaid on the Sfbgly active site niche. The superimpositions were prepared and visualized using the PYMOL v 0.99 software.

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### Conflict of interest

The authors declare no conflict of interest.

### Author contributions

RSC, RKS, and SRM planned the experiments. RSC and MARP performed the experiments. RSC, FAMO,



MARP, RKS, and SRM analyzed the data. RKS and SRM contributed to the resources. RSC, FAMO, MARP, RKS, and SRM involved in writing—original draft, and review and editing.

## Data accessibility

The data that support the findings of this study are available in the figures, tables, and the supplementary material of this article.

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## Supporting information

Additional supporting information may be found online in the Supporting Information section at the end of the article.

**Fig. S1.** SDS/PAGE of purified recombinant Sfbgly.

**Fig. S2.** Determination of the mechanism of Sfbgly inhibition by imidazole. The substrate is *p*-nitrophenyl  $\beta$ -glucoside. Enzyme sample number: 1.

**Fig. S3.** Determination of the mechanism of Sfbgly inhibition by imidazole. The substrate is *p*-nitrophenyl  $\beta$ -glucoside. Enzyme sample number: 2.

**Fig. S4.** Determination of the mechanism of Sfbgly inhibition by imidazole. The substrate is *p*-nitrophenyl  $\beta$ -glucoside. Enzyme sample number: 3.

**Fig. S5.** Determination of the mechanism of Sfbgly inhibition by imidazole. The substrate is cellobiose. Enzyme sample number: 1.

**Fig. S6.** Determination of the mechanism of Sfbgly inhibition by imidazole. The substrate is cellobiose. Enzyme sample number: 2.

**Fig. S7.** Determination of the mechanism of Sfbgly inhibition by imidazole. The substrate is cellobiose. Enzyme sample number: 3.

**Fig. S8.** Determination of the mechanism of Sfbgly inhibition by imidazole. The substrate is cellotetraose. Enzyme sample number: 1.

**Fig. S9.** Determination of the mechanism of Sfbgly inhibition by imidazole. The substrate is cellotetraose. Enzyme sample number: 3.

**Fig. S10.** Determination of the mechanism of Sfbgly inhibition by imidazole. The substrate is cellotetraose. Enzyme sample number: 2.

**Fig. S11.** Determination of the mechanism of Sfbgly inhibition by cellobiose. The substrate is *p*-nitrophenyl  $\beta$ -glucoside.

**Fig. S12.** Determination of the mechanism of Sfbgly inhibition by imidazole in the presence of 5 mM cellobiose. The substrate is *p*-nitrophenyl  $\beta$ -glucoside.

**Fig. S13.** Determination of the mechanism of Sfbgly inhibition by imidazole in the presence of 10 mM cellobiose. The substrate is *p*-nitrophenyl  $\beta$ -glucoside.

**Fig. S14.** Determination of the mechanism of Sfbgly inhibition by imidazole in the presence of 20 mM cellobiose. The substrate is *p*-nitrophenyl  $\beta$ -glucoside.

**Fig. S15.** Determination of the mechanism of Sfbgly inhibition by imidazole in the presence of 30 mM cellobiose. The substrate is *p*-nitrophenyl  $\beta$ -glucoside.

**Fig. S16.** Deduction of the rate equation describing the simultaneous effect of a partial competitive and a competitive inhibitor on the enzymatic hydrolysis of the substrate.

**Fig. S17.** Imidazole and substrate binding within the Sfbgly active site.

**Table S1.** Kinetic parameters for the hydrolysis of different substrates by Sfbgly.