

Cyclodextrins-based sorbents for sustainable sample preparation focusing on food analysis

Edvaldo Vasconcelos Soares Maciel^{a,b}, Natalia Gabrielly Pereira dos Santos^a,
 Deyber Arley Vargas Medina^a, Fernando Mauro Lanças^{a,*}

^a Institute of Chemistry of São Carlos, University of São Paulo, São Carlos, SP, Brazil

^b Department of Chemistry, Clemens Schöpf Institute, Technical University of Darmstadt, Darmstadt 64287, Germany

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ABSTRACT

Food analysis is critical for ensuring human health and safety, as it involves the detection of contaminants such as pesticides, heavy metals, mycotoxins, allergens, and microbial contaminants. Effective extraction and pre-concentration of these analytes are crucial to achieve accurate analysis and comply with regulatory standards. However, food matrices are complex, including solid, semi-solid, and liquid components, fats, proteins, carbohydrates, vitamins, minerals, and other organic and inorganic compounds. Therefore, isolation of target analytes is a challenging task. Developing efficient sample preparation techniques is essential to overcome such a downside. Exploring new and improved sorbent materials is one of the advanced strategies for developing selective and greener sample preparation approaches. Cyclodextrins (CDs) play a significant role in this topic. CDs can interact with various compounds with their external hydrophilic structure and central hydrophobic cavity. The functionalization of CDs' external hydroxyl groups allows for modifications in solubility, cavity opening, and bonding with other sorbent materials. This review discusses the current status of CDs as a robust sorbent material in food- and nutraceutical analysis. It covers the principles of CD-based sorbent preparation. It explores combining them with other materials, creating customized sorbents for solid-phase extraction techniques, including dispersive, packed devices-based, and coated devices-based methods. Moreover, we offer insights into relevant applications focusing on food surveillance and nutraceutical analysis.

Introduction

The significant advancements achieved in analytical instrumentation during the last decades have pushed forward the capabilities of the existing methods for food analysis. Monitoring and ensuring regulatory compliance in nutraceutical products is of great concern. However, this is always challenging, mainly because sample preparation of such matrices is still a bottleneck in many cases. This is primarily attributed to this procedure's labor-intensive and time-consuming nature, which makes the analytical methods susceptible to errors [1]. Sample

preparation can account for approximately 70% of the total analysis time, including the entire analytical workflow [2]. It is often the most expensive and environmentally impactful step [3]. Traditional methods such as liquid-liquid extraction (LLE) and solid-phase extraction (SPE) remain prevalent in the industry sector [3]. However, analytical chemists are actively developing greener and high-efficient approaches consistent with the principles of Green Analytical Chemistry (GAC) [4, 5].

Green sample preparation involves using safe solvents, reagents, and renewable, recycled, and reusable materials. This approach minimizes

Abbreviations: ATP, attapulgit; CD, cyclodextrin; CNTS, carbon nanotubes; COF, covalent organic framework; DAN, danofloxacin; DLLME, dispersive liquid-liquid microextraction; EPI, epichlorohydrin; HF-LPME, hollow fiber liquid phase microextraction; IL, ionic liquid; GAC, green analytical chemistry; GO, graphene oxide; LLE, liquid-liquid extraction; MEPS, microextraction by packed sorbent; MIM, molecularly imprinted monolith; MIP, molecularly imprinted polymer; MNPs, magnetic nanoparticles; MOF, metal-organic framework; MSPD, matrix solid-phase dispersion; MWCNTs, multiple-walled carbon nanotubes; NC, nanocellulose; NP, nanoparticle; NS, nanosponge; PAHs, polycyclic aromatic hydrocarbons; PANI, polyaniline; PISA, polymerization-induced self-assembly; RAM, restricted access media; SBSE, stir bar sorptive extraction; SDME, single-drop microextraction; SMs, supramolecular macrocycles; SPME, solid-phase microextraction; SPE, solid-phase extraction; TFME, thin film microextraction.

* Corresponding author.

E-mail address: flancas@iqsc.usp.br (F.M. Lanças).

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waste generation and reduces energy demand [6]. While a rigorous evaluation is necessary to assess the actual environmental impact of new sample preparation developments, specific approaches are generally recognized for contributing positively towards more sustainable methods. Among these approaches, miniaturization, simplification/automation of techniques, and developing new efficient and sustainable extraction phases are key factors [7,8].

The miniaturization of sorbent-based techniques initiated by introducing solid-phase microextraction (SPME) in the early 1990s revolutionized sample preparation in analytical chemistry [9]. In short, SPME is based on the same principles as its predecessor SPE but focuses on reducing the use of toxic organic solvents and the sample amount [10]. The advent of SPME has encouraged the development of other miniaturized sorbent-based techniques over the past three decades, including μ -SPE, microextraction by packed sorbent (MEPS), stir bar sorptive extraction (SBSE), thin film microextraction (TFME), and more [11]. In addition, sorbent-based sample preparation microtechniques currently include sorbent-coated and packed extraction devices and dispersive-based approaches [10].

While this review primarily focuses on sorbent-based techniques, mentioning the modernization of solvent-based techniques derived from LLE is essential. Modern examples include dispersive liquid-liquid microextraction (DLLME), single-drop microextraction (SDME), and hollow fiber liquid phase microextraction (HF-LPME), among others [11]. All these miniaturized techniques have found extensive applications in food and nutraceutical analysis, contributing significantly to developing more sustainable methods. Furthermore, these techniques are aligned with green and sustainable principles, always focusing on less consumption of chemicals and reduced generation of toxic waste. Similarly, the automation of the analytical workflow has become crucial for developing modern sample preparation procedures [8,12]. Automating this procedure offers several advantages, including improved precision, enhanced reproducibility, increased analytical frequency, and faster analysis. Automated sample preparation techniques can run with significantly reduced human intervention, producing results less prone to analytical errors. Also, the analyst is less frequently exposed to hazardous chemicals, promoting a safer working environment [13]. Automating and integrating it with other analytical techniques is pivotal in the progress of analytical chemistry, including the food sector.

Although both miniaturization and automation have been keystones in this field, the possibility of obtaining even more sensitive methods depends highly on new, more selective, and efficient extraction materials [14]. Developing new sorbents is an attractive research area, and different classes are constantly springing up around the literature [15, 16]. Widely known examples of them include molecularly imprinted polymers and monoliths (MIPs and MIMs), carbon-based compounds, ionic liquids (ILs), metal and covalent organic frameworks (MOFs and COFs), restricted access media (RAM), aptamer-based materials, immunosorbents, and supramolecular macrocycles (SMs) [7,17]. For more details about them, a series of excellent reviews are cited herein [18–25].

Among them, cyclodextrins (CDs) have a particular interest because of their selectivity, capability to interact with polar and non-polar

compounds, as well as the possibility of being derivatized, supported, or chemically functionalized to obtain sorbents with customized properties [26]. Furthermore, CDs are supramolecular compounds (Fig. 1), i. e., they can be used to obtain nanoscopic solids without covalent bonds between them [24]. In this case, the main interactions emerge from host-guest principles (inclusion complexes). In addition, CDs have large recognition cavities performing an excellent role as host molecules [26]. Due to their uniqueness, CDs can extract and pre-concentrate target analytes from matrices with several endogenous and exogenous compounds.

This review will primarily focus on samples related to food analysis, as they hold significant relevance due to their direct impact on human health. Developing and implementing analytical methods in this field is a complex task that requires careful consideration [28]. Usually, food-derived products are susceptible to contamination with external compounds (e.g., pharmaceutical drugs, pathogens, mycotoxins), mostly during production, transportation, or storage [29]. For this reason, international regulatory agencies and governments have strict laws to keep those residues and contaminants below harmful levels to humans and other animals [30]. However, these toxic compounds are often in lower concentrations than matrix substances, requiring very efficient sample preparation methods capable of extracting target analytes selectively and pre-concentrating them [31]. Here, cyclodextrin-based sorbents and their application in modern sample preparation methods for food analysis will be discussed, highlighting the most critical features. Relevant selected applications will be reviewed, mainly covering the last ten years.

Cyclodextrin-based sorbents

CDs are non-reducing oligosaccharides formed by linking glucopyranose units in a truncated cone shape [32]. These molecules possess a hydrophobic cavity that enables the formation of inclusion complexes with various analytes of interest in food analysis [33], including veterinary drugs, polycyclic aromatic hydrocarbons, dyes, aromatic derivatives, and volatile organic compounds [24]. From a purely analytical viewpoint, the above characteristics should be enough to give CDs the title of a "good" sorbent. What is not so obvious is that cyclodextrins are naturally occurring members of the cyclic oligosaccharides family of nontoxic compounds, usually possessing a -OCCO- binding motif. CDs can easily interact with metals from IA and IIA groups, forming customized materials even in an aqueous solution [34]. For example, metal-organic frameworks can be prepared under ambient conditions using just aqueous media [35]; or members of the apatite family can easily interact with CDs to form cross-linked nanocomposites [36]. Moreover, when it is necessary to produce CDs synthetically, especially in low amounts, some processes can further use the waste as raw material for producing other active substances, for example, alcohol [37]. In short, employing CD-based sorbents is often an environmentally friendly idea.

Among macrocyclic molecules, cyclodextrins offer notable advantages due to their relatively simple synthesis methodology and the ability to customize their properties for specific applications [38].

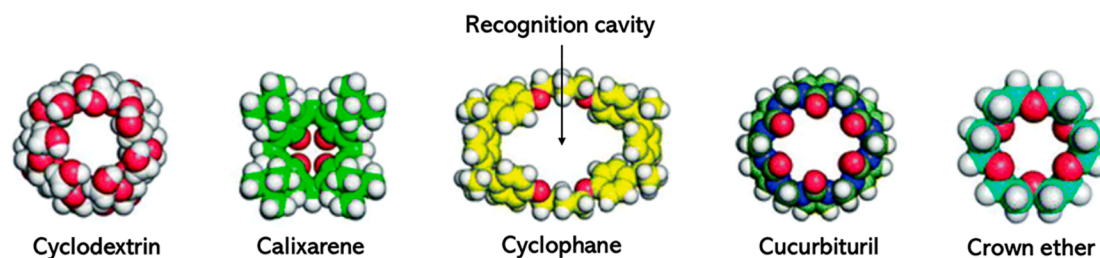


Fig. 1. Representative structures of common supramolecular macrocycles with potential for sample preparation use. Adapted from [27] with permission of Royal Chemistry Society, 2012.

Cyclodextrins are enzymatically derived from starch degradation, and the most commonly used homologs are α -CD, β -CD, and γ -CD, consisting of six, seven, and eight glucopyranose units, respectively [32]. However, β -CD is widely used in sample preparation due to its unique characteristics, including an intermediate-sized molecular cavity compared to α - and γ -CD, reduced solubility in water based-solutions resulting from favorable hydrogen bond localization, excellent complexation ability, and inherent crystal stability [39].

CDs have limited solubility in water, which restricts their use as standalone sorbents in sample preparation [40]. Aqueous CD solutions have been previously used to remove interfering compounds and extract target analytes. For example, Zhu et al. employed an aqueous β -CD solution to extract and quantify eight phenolic compounds from jujube fruit tissues [41]. The procedure involved adding dry pulp, seeds, and peel powders to a β -CD solution, followed by agitation, centrifugation, and filtration. The resulting clear liquid was then subjected to chromatographic analysis.

CDs are primarily used as sorbents in solid-phase extractions. However, their grade of solubility in water demands their covalent attachment to solid supports to enhance stability and enable their application in sorbent-based extraction methods [42]. CDs-based sorbents are typically achieved through three main approaches: (i) immobilization onto inert supports, (ii) combination with various nanomaterials (such as magnetic nanoparticles, carbon nanomaterials, carbonaceous nanomaterials, and polymer-based sorbents), and (iii) formation of nanosponges (NSs) by polymerizing CDs (monomeric units) with a suitable cross-linking agent [43].

Among these approaches, immobilizing CDs onto inert materials or combining them with nanomaterials often leads to sorbents with a low proportion of CDs. In contrast, NSs are regarded as auspicious materials. This approach allows the production of materials with diverse physical textures, mechanical properties, and shapes while maintaining a balanced presence of hydrophilic and hydrophobic groups. Furthermore, the swelling properties and amorphous nature of NSs facilitate analyte diffusion within the polymer network and simplify the sorption process. In recent developments, the focus has been on producing even more efficient materials. Efforts are underway to prepare CDs-based sorbents functionalized with diverse ionic and neutral groups, enabling simultaneous organic compounds, metals, and anions adsorption. Additionally, there is interest in creating materials capable of operating over a wide pH range [43].

However, CDs are primarily utilized as sorbents in solid-phase extractions. Due to their solubility in water, the covalent attachment of CDs to solid supports is essential to enhance their stability and make them suitable for sorbent-based extraction methods. These hybrid CD-based sorbents involve the incorporation of organic, inorganic, or polymeric supports into the hydroxyl groups of cyclodextrins [42].

Preparation of cyclodextrin-based sorbents

Cyclodextrins anchored to inert supports

Attapulgite (ATP) and silica are the more commonly used inert materials for immobilizing CDs [43]. Attapulgite is a hydrated magnesium aluminum silicate, $[(\text{OH}_2)_4(\text{Mg}, \text{Al}, \text{Fe})_5(\text{OH})_2\text{Si}_8\text{O}_{20}] \cdot 4\text{H}_2\text{O}$, and it is typically coupled with CDs using KH-560 ((3-Glycidyloxypropyl)trimethoxysilane) as a coupling agent. The coupling agent reacts with the OH groups of ATP through its methoxy silane groups, allowing the subsequent anchoring of CDs by reaction with the oxirane moiety in the presence of NaH and using dimethylformamide as the solvent (Fig. 2) [44]. For instance, Cui and coworkers prepared a β -CD/ATP sorbent for DSPE of fluoroquinolones in honey samples, achieving recoveries between 74% and 103% with relative standard deviations (RSD) of less than 7.4% [44].

Silica is widely employed as a material for functionalizing CDs due to its controlled composition, morphology, and porosity. CD-silica

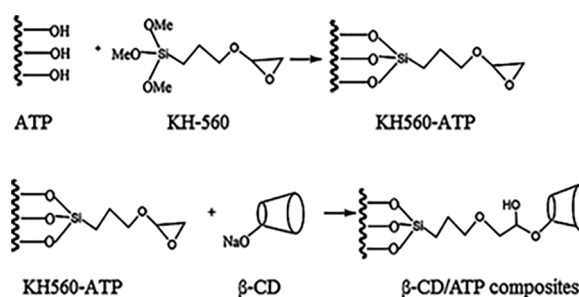


Fig. 2. Schematic representation of the synthetic route for the obtention of β -CD/ATP Composites [44].

materials can be categorized into two main classes: cyclodextrin-functionalized silicas prepared through grafting or coating reactions and cyclodextrin-silica hybrid systems prepared through sol-gel or self-assembly methods [45]. The first approach yields grafted matrices (Fig. 3a) and coated materials (Fig. 3b). These materials offer improved access to binding sites and enhanced thermal, mechanical, and chemical stability. However, they may suffer from uneven distribution of cyclodextrin molecules and low cyclodextrin loading, which can limit their adsorption capacities. In the second approach, nanoporous frameworks are obtained (Fig. 3c). These hybrid systems exhibit high cyclodextrin group loadings and robust structures, improving their adsorption capacity.

The literature comprehensively covers cyclodextrin-based silica materials' preparation, characterization, properties, and applications [43,45]. In this section, we have highlighted a few notable examples. However, it is relevant to note that numerous other silica-CD sorbents will be discussed in subsequent sections, particularly concerning different modalities of sorbent-based extractions for the treatment of food and beverage samples. For example, Zhang et al. [46] developed a sorbent from silica core-shell particles containing a magnetic core, followed by functionalization with mesoporous silica and cyclodextrin ($\text{Fe}_3\text{O}_4@/\text{SiO}_2@m\text{SiO}_2\text{-}\beta\text{-CD}$), which was used for the extraction of doxycycline, an antibiotic of the tetracycline class. Langmuir isotherm models demonstrated a maximum adsorption capacity of 78 mg/g compared to 34 mg/g for the reference sorbent without cyclodextrin ($\text{Fe}_3\text{O}_4@/\text{SiO}_2@m\text{SiO}_2$). In another study [47], they functionalized mesoporous silica with β -CD ($\text{SiO}_2\text{-}\beta\text{-CD}$) using the sol-gel method to remove methylene blue dye. The Toth isotherm model was used in this case, revealing a maximum adsorption capacity of 60.55 mg/g. In this case, De Carvalho et al. [48] also dedicated efforts to removing methylene blue dye; however, the sorbent was developed from silica gel. The results showed a maximum adsorption capacity of 212 mg/g using the Langmuir isoform method. CDs immobilized onto silica particles also have demonstrated significant relevance in food analysis. In a recent study, a large pore hybrid mesoporous silica functionalized with β -cyclodextrins (LP-MS- β -CD) was utilized for the extraction of polyphenols from strawberry tree fruits, including arbutin and catechins. Both SPE and DSPE were employed. Remarkably, the DSPE technique using CD sorbents exhibited the most favorable extraction outcomes, with recoveries ranging from 104 to 113%. [49].

Cyclodextrin nanocomposites

CDs can also be attached to other nanomaterials to obtain hybrid sorbents with enhanced properties. For instance, carbon allotropes such as graphene oxide (GO) and carbon nanotubes (CNTs) functionalized with cyclodextrins have shown excellent extraction performance in various applications. [33]. These carbon allotropes stand out due to their high surface area, which in association with the CD adsorption capability, results in sorbent material with enhanced extraction characteristics. In the case of GO, the high aggregation of nanosheets hinders

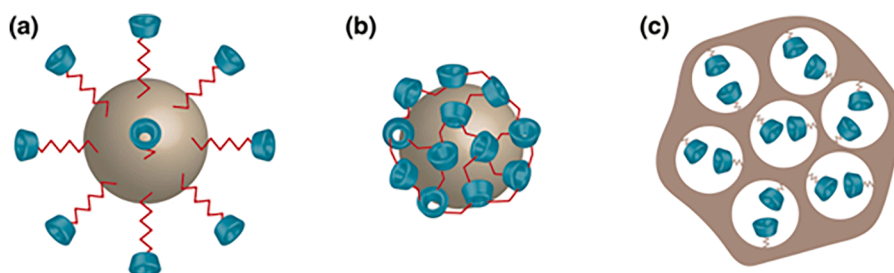


Fig. 3. Schematic representations of cyclodextrin-containing silica networks: (a) Grafted matrices, (b) Coated materials, and (c) Nanoporous frameworks [45].

its dispersion in an aqueous solution, making challenging applications in this medium. However, such limitations can be overcome through CD-GO anchoring. Song et al. [50] prepared β -CD-GO via in situ polymerization by sorption of Co(II) from aqueous solutions. The results showed that the ions' complexation occurs due to many functional groups containing oxygen on the surface of β -CD-GO, obtaining sorption of 72.73 mg/g under pH 6.0 \pm 0.1 and $T = 303$ K.

Focusing on food analysis, Lança's research group recently introduced a β -CD-GO@SiO₂ composite for the MEPS of isoflavones from juice samples [51]. The extractions were performed using only 1.0 mg of the sorbent material, resulting in excellent selectivity and enrichment capability while minimizing matrix effects. Additionally, this same material was utilized as an extractant in the automated online coupling of SPME with liquid chromatography [52]. We developed innovative coated needle sleeve devices attached to the endpoint of a syringe needle on a lab-made robot. This automated system performed the entire extraction procedure, including injecting the enriched extract into the chromatographic system.

Like GO, CNTs also possess low dispersivity in an aqueous solution. However, this material has excellent characteristics such as hollow and porous cylindrical structure and low mass density. In this sense, cyclodextrin can be used to increase the dispersion of this material [33]. As an example, He et al. [53] developed a sorbent based on cyclodextrin grafted on multiple-walled carbon nanotubes (MWCNTs/ β -CD) to remove Cr(IV) and phenol. The material's adsorption capacity was determined by the Langmuir isotherm method and compared with carbon nanotubes (MWCNTs-COOH). The data demonstrated that MWCNTs/ β -CD showed adsorption of 9.74 mg/g for Cr(VI) due to the complexation of ions with hydroxyl groups on the outer surface of beta-CD, compared to 6.76 mg/g of MWCNTs-COOH. This adsorption was 100.27 mg/g for phenol and 77.66 mg/g, respectively. In an illustrative application of this type of material in food analysis, Chen and colleagues devised a carbon nanotube (CNT) nanocomposite incorporated with β -cyclodextrins (CNT- β -CD), which was subsequently immobilized within the pores of a hollow fiber using sol-gel technology. This composite material was employed for the trace analysis of tomato pesticides, yielding satisfactory recoveries ranging from 84.2 to 108.9% [54].

As illustrated in subsequent sections, combining CDs with magnetic nanoparticles (MNPs) is one of the more notable current trends in developing new CD-based sorbents for sample preparation methods. Among nanoparticles, iron oxide (Fe₃O₄) stands out as the primary magnetic carrier for CD functionalization, primarily due to its easy preparation and its excellent magnetic properties that facilitate efficient sorbent uptake using an external magnetic field. Some recent examples of this kind of sorbent involve the functionalization of β -cyclodextrin with ionic liquids and immobilized on magnetic nanoparticles (Fe₃O₄@ β -CD-Vinyl-TDT) to analyze polycyclic aromatic hydrocarbons (PAHs) in rice samples [55]. In addition, Ansari et al. [56] described the synthesis of hydroxyapatite- β -cyclodextrin (Fe₃O₄@HA- β -CD) used to remove heavy metals. More recently, Liu and coworkers [57] developed an N, N'-carbonyldiimidazole- β -cyclodextrin sorbent (Fe₃O₄@ β -CD-CDI) for sorption and degradation of bisphenol A.

Cyclodextrin nanoporous frameworks (Nanosponges)

Cyclodextrin-based nanosponges are a term used to define a family of insoluble organic and inorganic polymers which possess voids, internal cavities, and a porous network of nanoscale dimensions [58]. Usually, they formed colloidal dispersions in a solution that could encapsulate small molecules inside its irregular lattice [59]. Therefore, CD-based nanosponges have been found applicable as sorbent for sample preparation. So far, there are applications of them covering: decontamination of drinking water [60], bisphenols from water and juices [61], PAHs from rice [55], and pesticides from cow milk [62], among other water-related applications [63]. These polymers are obtained by polymerizing a cyclodextrin derivative as a monomer or by polycondensing cyclodextrin monomers with cross-linking agents (bi- or multifunctional molecules) [33]. The latter approach is most commonly used for synthesizing CD-based sorbents. For instance, epichlorohydrin (EPI) is a widely used cross-linker due to its high efficiency but is not environmentally friendly [64]. Other examples of cross-linkers are diisocyanates (e.g., hexamethylene diisocyanate, toluene diisocyanate), which require organic solvents in the reaction, or more eco-friendly alternatives such as carbonyls (e.g., diphenyl carbonate, N, N'-carbonyldiimidazole), polycarboxylic acids, halides, and anhydride derivatives [65].

Another example is ethylenediaminetetraacetic acid (EDTA) [66], which synthesized the EDTA- β -CD polymer with a high sorption performance in removing metals. It is worth remembering that nanosponges are distinct from other CD-based sorbents, especially regarding the adsorption mechanism [43]. The characteristic cone shape-like of cyclodextrins is just partially available because they are primarily blocked throughout the nanoporous 3D structure of the solid material [43]. This means that in addition to host-guest interactions, CD-based nanosponges can trap analytes by other mechanisms that are still not fully understood [43,67]. This would explain the less popularity of this material compared to the other CD-based sorbents and the necessity of developing more greener nanosponges without using toxic cross-linkers and solvents.

Applications of cyclodextrins-based sorbents in food and nutraceutical analysis

CDs have become pivotal in advancing sorbent-based extractions from food and beverage samples. This encompasses diverse extraction techniques involving various CD-based sorbents, devices, and modalities. However, it is essential to note that food constitutes a collection of distinct matrices rather than a singular entity. The origin of the food, whether animal or plant-based, introduces a vast spectrum of samples characterized by substantial variations in physical attributes (solid, semi-solid, liquid, and viscous materials) and chemical composition. Such diversity poses challenges and necessitates food analysts to employ cutting-edge science and technology.

The physical characteristics of food samples contribute to significant variation due to uneven analyte distribution and interfering matrix components. Consequently, sample pretreatment is necessary to ensure

homogenization, particularly for solid and semi-solid food samples. Techniques such as dry grinding, wet grinding-slurry mixing, or cryogenic grinding are commonly employed to achieve uniformity in solid samples [68]. While Matrix Solid-Phase Dispersion (MSPD) and QuEChERS enable direct treatment of solid samples, most cases involve suspension, centrifugation, and filtration to transfer analytes into a liquid medium suitable for CD-based sorbent microextraction. After these operations, the homogenized solid sample can be extracted with a suitable solvent, resulting in an extract that can be further processed as a liquid sample [69]. The choice of suspension solvent and specific sample pretreatment details vary depending on the application and sample characteristics.

Even after sample pretreatment, food extracts remain complex matrices due to the presence of diverse compounds, including proteins, fats, sugars, salts, vitamins, additives, and more, each possessing distinct polarity, size, and chemical properties. These matrix interferences adversely affect critical analytical parameters such as limits of detection and quantification, accuracy, and precision, leading to reduced sensitivity and selectivity during analysis [70]. Therefore, direct analysis of food samples is generally impractical, and an appropriate extraction procedure must precede instrumental analysis. In this context, CD-based sorbents play a pivotal role by offering numerous opportunities for combining with other materials to create sorbents with enhanced selectivity. As discussed further in the subsequent section, these sorbents have demonstrated outstanding performance in a wide range of applications, encompassing diverse embodiments of sorbent-based extraction and addressing various analytes and matrices.

CDs in packed devices-based micro extractions

The efficacy of cyclodextrin-functionalized composites as extraction media has also been substantiated in various packed-device-based micro extractions, with offline approaches especially prevalent in recent years. Compared with SPME or SBSE, sorbent-packed-based techniques can offer some advantages, such as easier preparation of the extractant devices and a more significant number of accessible active sites for analyte/sorbent interaction. The packed-devices micro techniques rely on extraction microcolumns through which samples and solvents are percolated. With the appropriate sequence of column conditioning, sampling, washing, and elution, analytes can be selectively enriched, and the matrix interferences eliminated.

In recent years, cyclodextrin-functionalized sorbents have gained

considerable attention in the development of packed devices for various techniques, including conventional solid-phase extraction (SPE) and its miniaturized counterparts such as μ -SPE, Disposable pipette tip micro-extraction (DPX), and MEPS. Table 1 provides a concise summary of selected examples highlighting the application of cyclodextrin-functionalized sorbents in analyzing food and beverage samples using packed devices-based techniques. The table showcases the versatility of these techniques, in conjunction with a diverse range of cyclodextrin-based sorbents, for the extraction of a wide array of volatile and non-volatile organic compounds, including pesticides, mycotoxins, flavonoids, and hormones, among many others. These compounds have been successfully extracted from various matrices, including vegetables, meat, feeds, juices, and wine. Additionally, it includes details regarding the quantity of sorbent utilized in each case, the analytical system employed, and the limits of detection (LODs) obtained. These comprehensive insights provide the reader with a clear understanding of the efficacy of cyclodextrin-based sorbents in these applications.

Conventional solid-phase extraction (SPE) is a simple approach for testing functionalized sorbents. It has been used to assess materials such as cyclodextrins@silica nanocomposites [80], β -cyclodextrin polymers [61], β -cyclodextrin based MIPs [81], polymers grafted with allotropic forms of carbon [82], β -cyclodextrin-nanosponges [73], and Cyclodextrin-metal-organic framework (CD-MOF) composites [35], among many others. In most cases, between 50 and 100 mg of sorbent are packed into a reusable SPE device (or a polypropylene syringe) and retained between two porous frits. Although incredibly convenient for testing particulate cyclodextrins functionalized materials, SPE has also been employed with other sorbents, such as cyclodextrins functionalized sponges. For example, Hou et al. prepared a covalently β -Cyclodextrin modified three-dimensional graphene oxide-wrapped melamine foam (MF) [72]. In this case, once prepared, the functionalized MF was packed into a 3 mL propylene SPE cartridge, compressed by two sieve plates, washed, and conditioned for further SPE of flavonoids from a *Lycium barbarum* extract (Fig. 4a). The β -CD/GO-wrapped MF exhibited an interconnected framework with high porosity and some lamellar structure, providing excellent extraction efficiency through π - π , hydrogen bonding, and host-guest interactions. Afterward, the same research group introduced a similar approach for the SPE of green malachite from seafood samples [76]. After preparation, a cube of novel β -CD-GO/MF was transferred into a 2.5 mL syringe and stuck between 0.22 μ m filters. This approach proved to be a fast and efficient sample preparation strategy, with satisfactory recoveries and lower

Table 1

Selected applications of cyclodextrins-functionalized sorbents in packed-devices-based techniques for the treatment of food and beverage samples.

Analytes	Matrix	technique	Sorbent	Sorbent amount (mg)	analysis	LOD	Refs.
<i>Food and beverage samples</i>							
Methyl parathion and fenthion	Lettuce	DPX	Acryloyl β -CD-silica hybrid monolithic column	–	HPLC-DAD	4.5–6.0 μ g kg ⁻¹	[71]
Sulfonamides	Chicken, pork, liver, and fish	SPE	β -CD/MOF	40	HPLC-UV	0.32–1.7 μ g L ⁻¹	[35]
Flavonoids	Chinese herbal medicine	SPE	β -CD/GO/MF	–	HPLC-UV	0.5–2 μ g L ⁻¹	[72]
Ochratoxin A	Grape juice and Wine	SPE	polyurethane/ β CD composite nanosponge	300	HPLC-FLD	0.2 μ g L ⁻¹	[73]
Carbendazim and carbaryl	Leafy vegetables	DPX	Acryloyl β -CD-silica hybrid monolithic column	–	HPLC-DAD	1.0 and 1.5 μ g kg ⁻¹	[74]
Phenolic compounds	Tea beverages	SPE	Poly (glycidyl-co-ethylene dimethacrylate) nanohybrid modified with β -CD	100	HPLC-FD GC-MS	0.7–17 μ g L ⁻¹ ,	[75]
Malachite Green	Seafood	SPE	β -CD/GO/MF	–	HPLC-UV	0.21 mg L ⁻¹	[76]
Plant growth regulators	Plant-derived tissues and Foods	SPE	β -CD/MAA-MIPs	200	GC-FID	0.012–0.023 μ g L ⁻¹	[77]
Isoflavones	Soy Juices	MEPS	β -CD/GO@SiO ₂	–	HPLC/MS/ MS	0.5–1.0 μ g L ⁻¹	[51]
Aflatoxins	Maize and animal feeds	μ -SPE	β -CD/porous graphene (β -CDPG)	15	HPLC-FLD	0.025–100 μ g kg ⁻¹	[78]
Sanshool acid amide compounds	Pepper oil resin	SPE	β -CD/MIP	–	HPLC-UV GC-MS	–	[79]

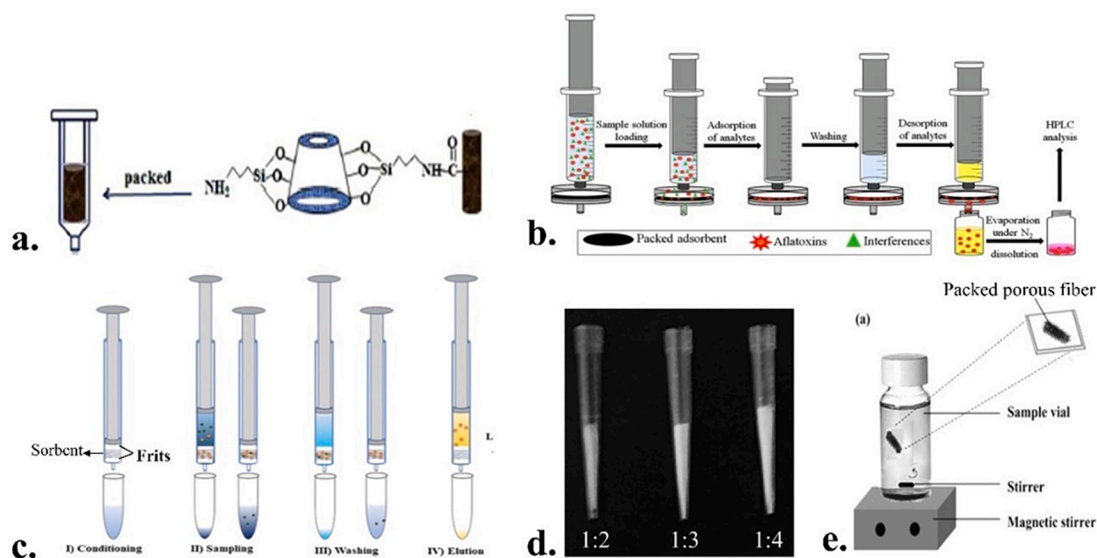


Fig. 4. Different formats of packed sorbent microextraction, employing cyclodextrin-functionalized extraction phases: **a.** SPE device packed with a β -Cyclodextrin-modified three-dimensional graphene oxide-wrapped melamine foam for extraction of flavonoids [72]; **b.** reusable syringe filter packed with a β -cyclodextrin decorated porous graphene nanohybrid to extract aflatoxins [78]; **c.** MEPS device adapted by the Lanças research group [83] and packed with β -Cyclodextrin, coupled to graphene oxide, supported the aminopropyl silica to extract isoflavones [51]; **d.** Pipette tips device packed with a β -cyclodextrin-silica hybrid monolithic column for the extraction of pesticides; **e.**, a packed device developed by the Lee research group [84] and packed with cyclodextrin-based polymer to extract steroids [85].

consumption of sorbent and solvents than those based on commercial SPE cartridges.

Some alternative sorbent-packed devices have been proposed to reduce the amount of sorbent consumed and provide miniaturized, environmentally friendly sample preparation methods. For example, Tezerji and coworkers developed a β -CD supported on porous graphene nanohybrid (β -CDPG) based μ -SPE setup [78], packing 15 mg of the supramolecular nanohybrid composite into a reusable filter holder (Fig. 4b). In the μ -SPE of aflatoxins from maize and animal feed samples, this setup allowed exploiting of the advantages of the recognition and enrichment capability of β -CD and the high specific surface area of the porous graphene, providing faster sample loading capacity than those possible with conventional SPE cartridges.

Da Silva et al. developed a cyclodextrin coupled to graphene oxide supported on an aminopropyl silica composite for the MEPS of isoflavones in soy beverages [51]. In this case, less than 4.0 mg of the β -CD/GO@SiO₂ sorbent was packed into a 1.0 mL polyethylene syringe and between two polypropylene frits (Fig. 4c). Despite the very low sorbent employed, this setup exhibits LODs comparable to those reported by previous SPE methods for isoflavones determination. Moreover, the setup allowed very profitable sorbent reuse (over 50 times) and speedy sample preparation, requiring less than 5 min for the whole sample preparation process.

DPX has also been particularly useful in testing cyclodextrin-functionalized monolithic extraction media. Chen and coworkers prepared hybrid monolithic columns by directly polymerizing the acryloyl β -CD as the monomer into pre-treated 200 μ L pipette tips (Fig. 4d). [74]. The monolith retained the β -CD molecular recognition capacity, and the developed pipette-tip setup allowed the sensitive determination of carbendazim and carbaryl from just 3.0 mL of extracts of leafy vegetables. The same research group recently described the preparation of acryloyl β -cyclodextrin-silica (A- β -CD-silica) hybrid monolithic columns through a sol-gel approach [71]. Sample preparation was performed by connecting the packed tip to a 5 mL gastight syringe to promote condition, sampling, washing, and elution with the syringe plunger. Assessed in the extraction of methyl parathion and fenthion in lettuce samples, this setup provided high enrichment factors and suitable sensitivity.

Another interesting cyclodextrin sorbent-packed approach combined

the packed-bed format and SBSE. Manaf and coworkers developed a porous fiber-packed to extract steroids from human urine, employing polymeric β -cyclodextrin [85]. First, sorbents were prepared by cross-linking β -cyclodextrin (β -CD) using two cross-linker units at variable reactant mole ratios. Subsequently, 5 mg of sorbent was packed into a folded porous membrane, and the extractions were performed by the suspension of the extractant device in the sample solution under stirring conditions (Fig. 4e). The developed setup demonstrated suitable extraction capabilities and outstanding selectivity due to the combination of the large molecule's exclusion properties of the porous membrane and the variable hydrophilic-lipophilic balance and accessible binding sites for forming an inclusion complex of the β -CD polymer sorbents.

CDs in coated devices-based techniques

SPME is undoubtedly the most widely adopted technique for utilizing extractive devices coated with a thin layer of sorbent [86]. The predominant approach involves using metallic or silica fibers with diameters below 500 μ m. Nonetheless, the emergence of SPME has also spurred the advancement of other coated-device-based techniques, such as in-tube SPME and SBSE. These alternatives replace the SPME fiber with open tubular columns or magnetic stir bars as extraction media (Fig. 5).

A recent study introduced a β -CD-functionalized silica-based SPME fiber (CDS) to enrich phenols [87]. Scanning electron microscopy analysis revealed the porous surface of CDS, resulting in increased extraction capacity. Additionally, this novel material exhibited shorter extraction and desorption times than commercial polydimethylsiloxane (PDMS) and polyacrylate (PA) fibers, leading to faster and more efficient analysis. The CD-fiber demonstrated superior enrichment of phenols, particularly for the target analyte 4-methyl phenol (4-MP), suggesting that phenols with sizes and shapes compatible with the β -CD cavity yield improved enrichment results.

Other examples of the application of cyclodextrins in SPME include their functionalization with carbon nanotubes (CNTs) in hollow fiber devices [88] for the extraction of plant hormones - 1-naphthalene acetic acid (NAA) and 2-naphthoxyacetic acid (2-NOA)- in tomato samples.

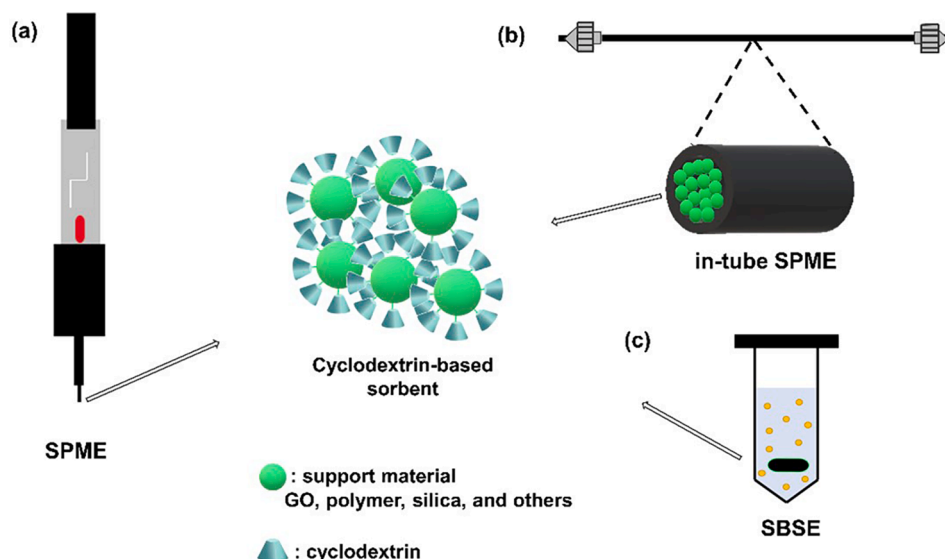


Fig. 5. Schematic representation of coated devices-based microextraction. (a) holder of SPME; (b) fiber in-tube SPME; (c) SBSE.

The results demonstrate that the enrichment factor for CNTs- β -CD was much higher ($EF = 275$ and 283 for NAA and 2-NOA, respectively) than that presented by a conventional hollow fiber ($EF = 34$ and 95) and another of CNTs ($EF = 153$ and 118).

Further examples include the modification of nanocellulose with β -cyclodextrin (CD-NC) [89], employing an amidation reaction (Fig. 6a) for the extraction of veterinary antibiotic danofloxacin (DAN) in milk samples. According to the authors, the interactions that lead to the inclusion complex's formation are hydrogen bonds, Van der Waals forces, and hydrophobic interactions. Pre-concentration data for DAN were generated to compare the new sorbent CD-NC with carboxylated and amidated nanocellulose, obtaining a value of 98.97%, 0.16%, and 0.12%, respectively (Fig. 6b). These results demonstrate the effectiveness of CD-NC in extracting the antibiotic and presenting a recovery rate of 76.40%.

Open tubular columns coated with cyclodextrin-based sorbents offer another coated devices-based sample preparation option. This technique, known as in-tube solid-phase microextraction (in-tube SPME), involves switching an extractive column to an analytical column through a commutative valve. This configuration allows direct loading of raw samples into the extraction column, washing out matrix interferences, and transferring analytes to the analytical column in an online and automated setup [90]. In-tube SPME is a versatile and advantageous approach for the automated treatment of complex samples. However,

after consulting the literature, no reports were found regarding using cyclodextrin-based sorbents in-tube SPME for food analysis. It is worth noting that there are no practical barriers to preparing capillary open tubular columns coated with cyclodextrin-based phases. Such devices have been previously developed for analytical [91] and extractive purposes [92]. Considering the exceptional selectivity of cyclodextrins, the development of open tubular columns for the online treatment of beverage and food extracts represents an intriguing research area. Given the outstanding selectivity of cyclodextrins, This could bring new insights into food surveillance and nutraceutical analysis.

Cyclodextrins have also been reported using the SBSE technique in another approach. Lei and coworkers [93] created a stir bar coated with polyaniline and α -cyclodextrin (PANI/ α -CD) for the enrichment of polychlorinated bisphenol (PCBs). The extraction performance of PANI- α -CD was compared with a commercial PDMS and a PANI stir bar through the peak area for each analyte. A good performance of the authors' sorbent was observed due to the material's hydrophobic interactions, inclusion complex, and π - π electron conjugation. In another report, Hu et al. [94] developed a PDMS/ β -CD stir bar for extracting bisphenol A and estrogens (estriol, 17β -estradiol, 17α -ethynylestradiol, and estrone) in samples of drinking water. Before applying the sorbent to the actual samples, the authors sought to understand the selectivity of the PDMS/ β -CD. For this, the new material's extraction capacity was compared with a conventional PDMS stir bar, using estrogens and PAHs

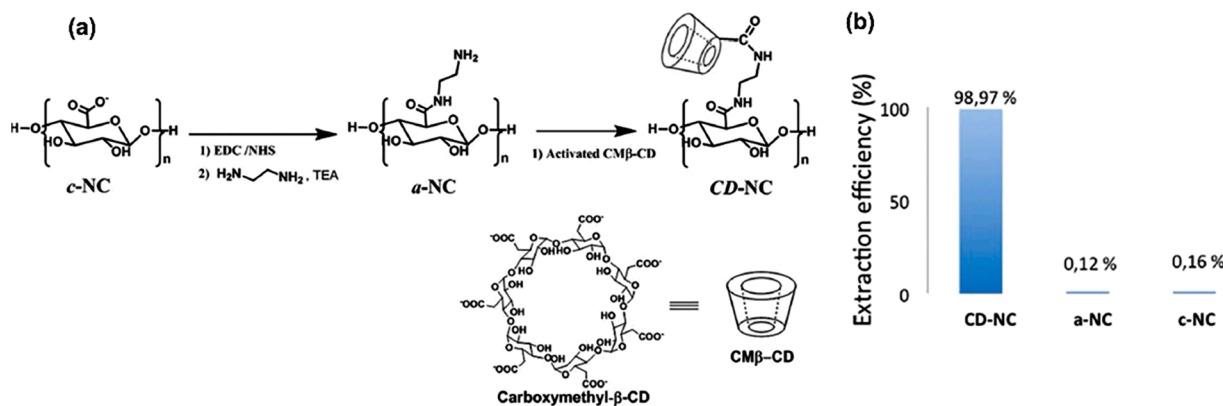


Fig. 6. (a) Synthesis of the CD-NC through the amidation reaction. (b) The extraction efficiency of CD-NC compared to amidated nanocellulose (a-NC) and carboxylated nanocellulose (c-NC). Adapted from [89] with permission of the Royal Society of Chemistry, 2015.

(fluoranthene and pyrene) standards PDMS/ β -CD provided enhanced extraction of estrogens, for instance, about three times for estriol. However, it showed a slight decrease in the extraction of PHAs (e.g., 185 ng of fluoranthene were extracted by PDMS/ β -CD compared to 186 ng for the PDMS stir bar). Therefore, the presence of β -CD in the sorbent increases the selectivity for polar compounds (estrogens and bisphenol A) more than for non-polar compounds.

Hu et al. [94] observed that cyclodextrin-based sorbents increase selectivity for polar compounds. This conclusion was also observed by Faraji et al. [95]. The authors used the silica functionalized with β -CD) to enrich phenolic compounds via SBSE [87]. The CDS stir bar was compared with the conventional PDMS stir bar. The results showed a higher extraction capacity of the CDS than the PDMS precisely because of its porous surface that increases enrichment and cyclodextrins' presence, making the material more polar. This CDS stir bar was tested on spiked drinking water samples with 15 $\mu\text{g/L}$ of each analyte, obtaining excellent extraction repeatability with RSDs of 3.1- 6.5% for seven replicates.

Functionalized membranes represent another type of coated device extensively employed in current applications. A notable example is the work by Hu et al. [96], where they devised a polydimethylsiloxane-cyclodextrin membrane (PDMS/ β -CD) via the sol-gel approach. This membrane was explicitly designed to enrich PAHs and phenols. In contrast, PDMS membranes have a higher sorption capacity (due to the high surface area) than conventional SPME fiber with the same material. In this study, the PDMS/ β -CD membrane showed a good capacity for enriching PAHs compared to the PDMS membrane, mainly for naphthalene, once it has an adequate size and shape to form a complex of inclusion with β -CD. Furthermore, this membrane also showed promising results for the extraction of phenols, under low pH conditions (where they are in neutral form) and high salt concentration, through the formation of hydrogen bonds or inclusion complexes. Therefore, the PDMS/ β -CD membrane proved versatile in enriching non-polar and polar compounds.

Table 2

Selected applications of dispersive sorbent-based extraction using CDs-based sorbents.

Analytes	Matrix	Technique	Sorbent	mg	analysis	LOD	Refs.
<i>Food and beverage samples</i>							
Plant growth regulators	Fresh vegetables	MSPE	IL/ Fe_3O_4 @ SiO_2 /GO/ β -CD	60	HPLC-MS	0.01 – 0.018 $\mu\text{g kg}^{-1}$	[97]
Antioxidant ingredients	<i>Mori fructus</i>	MSPE	β -CD	75	HPLC-UV	30.0 ng ml^{-1}	[98]
Anthocyanins	L. Ruthenicum fruit	DSPE	β -CD	5	UPLC-DAD	0.117 – 0.390 $\mu\text{g L}^{-1}$	[99]
Pyrethroids	Juice	MSPE	MNP@PAMAM@CD@IL	50	HPLC-DAD	0.36 - 1.3 $\mu\text{g L}^{-1}$	[100]
Phthalic acid esters	Water, milk, and wine	MSPE	MIP@m SiO_2 - β -CD@ Fe_3O_4	30	GC-MS	1.0–5.0 $\mu\text{g L}^{-1}$	[101]
Tetracyclines	Bovine milk	MSPE	Fe_3O_4 @ SiO_2 @GO- β -CD	10	HPLC-UV	1.8 - 2.9 $\mu\text{g L}^{-1}$	[102]
Lysozyme	Egg white	DSPE	Carboxyl-functionalized porous β -CD polymer P-CDP-COO-	30	HPLC-UV	-	[103]
Catechins	Green tea	DSPE	Fe_3O_4 /Amino-chitosan/ β -CD	-	HPLC-UV	0.22 - 0.18 $\mu\text{g g}^{-1}$	[104]
Phenolic compounds	Red beet extract	DSPE	β -CD	-	UV-VIs	-	[105]
Beta-carotene	Carrots	MSPE	Fe_3O_4 /Chitosan/ β -CD	60	HPLC-UV	0.21 $\mu\text{g mL}^{-1}$	[106]
Polycyclic aromatic hydrocarbons	Rice	MSPE	Fe_3O_4 /poly(β -cyclodextrin-ionic liquid) nanocomposites	20	GC-MS	0.01 - 0.18 $\mu\text{g kg}^{-1}$	[55]
Sudan dyes	Foodstuffs and environmental water	MSPE	Fe_3O_4 / porous β -CD polymers (PCDPs)	4–8	HPLC-DAD	0.025 – 0.054 $\mu\text{g L}^{-1}$	[107]
Carbaryl and carbofuran	Apple	MSPE	Fe_3O_4 @polyhedral oligomeric silsesquioxanes/ β -CD	15	HPLC-UV-VIs	0.5 - 0.7 $\mu\text{g kg}^{-1}$	[108]
Sulfonamides	Meat samples	MSPE	Fe_3O_4 @COF@Au- β -CD	20	HPLC-MS/MS	0.8 - 1.6 $\mu\text{g kg}^{-1}$	[109]
Phytohormones	Oilseeds	MSPE	Ti_3C_2 MXene/ β -CD/ Fe_3O_4 Fe_3O_4 @ Ti_3C_2 @ β -CD	5	UPLC-MS/MS	0.89 – 13.62 ng L^{-1}	[110]
17 β -estradiol	Milk	DSPE	β -CD/monodisperse RAM/ MIP	100	HPLC-UV	2.08 $\mu\text{g L}^{-1}$	[111]
Carbendazim	Fruits	DSPE	β -CD/MIP	20	HPLC-UV	0.03 mg L^{-1}	[112]
Plant growth regulators	Vegetables	MSPE	Fe_3O_4 @ SiO_2 /GO/ β -CD	100	GC-MS	0.04 - 0.28 $\mu\text{g kg}^{-1}$	[113]
Benzoylurea insecticides	Honey, tomato, and water samples	MSPE	β -CD polymer@ Fe_3O_4	16	HPLC-UV	0.02 - 0.8 $\mu\text{g kg}^{-1}$	[114]
Ofplant growth regulators	Fresh vegetable	MSPE	Fe_3O_4 @ SiO_2 /GO/ β -CD	80	UHPLC-QTrap-MS/MS	0.04 - 0.29 $\mu\text{g kg}^{-1}$	[115]

CDs in dispersive solid phase extractions

One of the most remarkable trends in exploiting CDs functionalized sorbents is their use in DSPE, particularly with magnetic material (MSPE). DSPE is a fast and efficient technique based on the dispersion of the extractant material into the sample solution under stirring conditions (or ultrasonic force convection). DSPE overcomes the drawbacks of the clogging and high backpressures that limit the sorbent-packed devices' lifetime and provide more effective interaction between the sorbent and the analytes than the coated devices-based techniques, resulting in fast and efficient extractions. Nevertheless, DSPE procedures typically involve centrifugation and filtration stages for sorbent separation. Therefore, MSPE has become a profitable alternative to exploit the advantages of the DSPE while excluding the filtration or centrifugation steps once an external magnetic field can easily retrieve the sorbent. Table 2 summarizes some selected recent publications describing applications of cyclodextrin functionalized sorbent for the treatment of food samples employing both DSPE and MSPE techniques.

Dispersive solid-phase extraction

In recent years, DSPE with cyclodextrin functionalized sorbents has been used in the treatment of food/beverage samples to effectively extract different analytes such as natural products [99], hormones [111], pollutants [112], and enzymes [103]. DSPE is a simple technique, easily implementable with raw cyclodextrin materials. For example, Zhang et al. extracted anthocyanins plant materials by mixing milled material with an extraction solvent and an aqueous solution containing β -CD [99]. A 1.65% β -CD solution and a liquid/solid ratio of 15:1 were enough to produce excellent extraction, providing a sustainable and reliable approach.

Cyclodextrin-based DSPE has also been utilized employing diverse polymeric materials, including carboxyl-functionalized porous β -cyclodextrin polymer (P-CDP-COO-) [103], MIPs [112], monodisperse restricted access media-molecularly - imprinted polymers (RAM-MIP)

[111], and some natural polymers derived sorbent such as amino-based chitosan cyclodextrin derivatives [104]. An alternative DSPE extraction with a cyclodextrin sorbent was reported by Yang and coworkers [116], who prepared effervescent tablets with the β -CD/ATP sorbent to extract pyrethroids without an external force such as vortex, stirring, or ultrasonication. The effervescence works as a dispersive mechanism so that this setup could be suitable for in-field environmental extractions. However, a centrifuge could be required to retrieve the extract sorbent.

Magnetic solid-phase extraction

MSPE is currently considered the preferred technique for evaluating innovative cyclodextrin-functionalized sorbents. Fe_3O_4 nanoparticles (NPs) are widely used to synthesize magnetic cyclodextrin-functionalized sorbents among magnetic carriers. These NPs can be directly functionalized with cyclodextrins or coated on silica or other polymers and then grafted with cyclodextrins and additional functionalized moieties.

An example of magnetic sorbent prepared by direct co-precipitation of MNPs in the presence of CDs was introduced by Wang et al. [117]. The

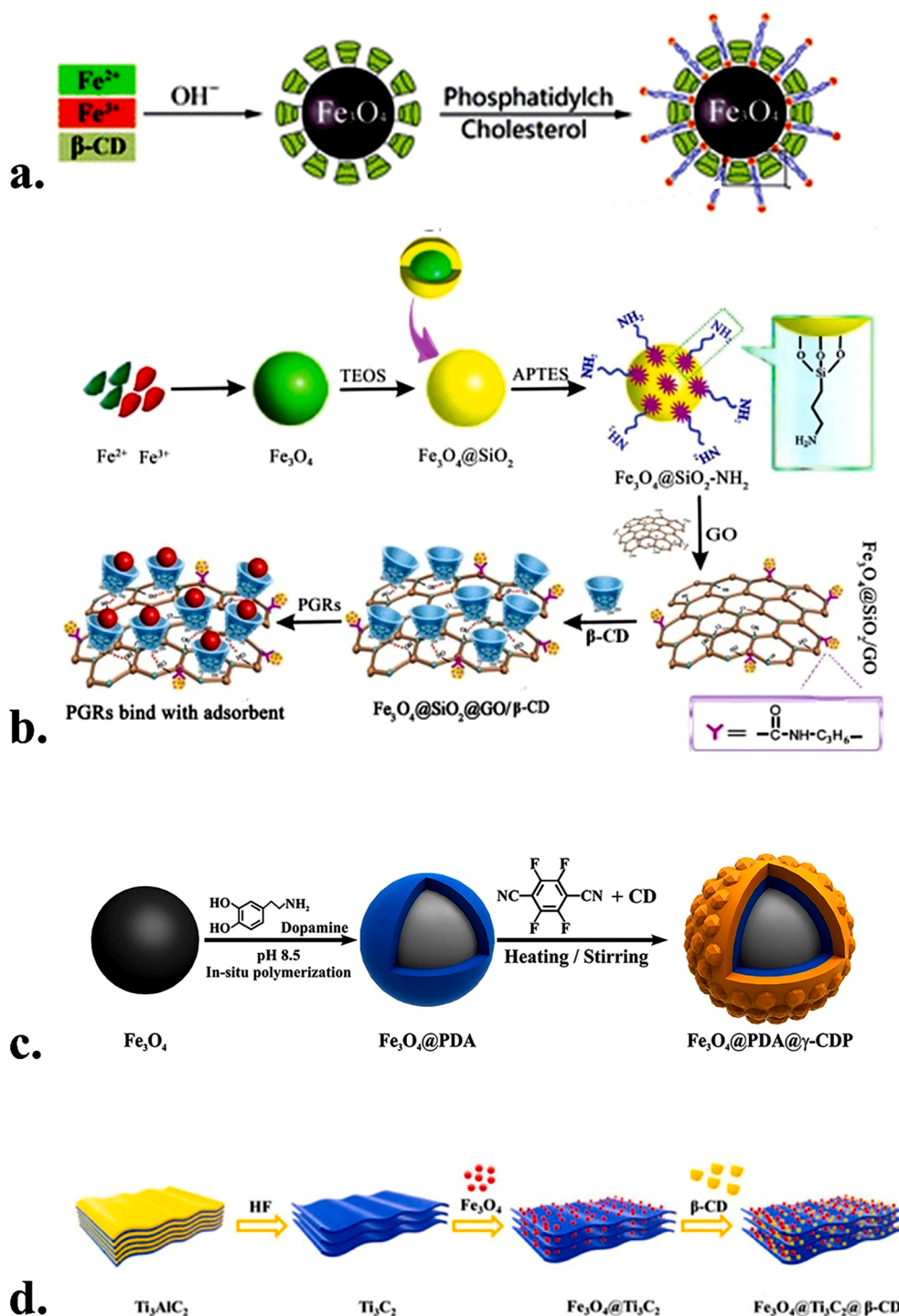


Fig. 7. Some examples of magnetic cyclodextrin functionalized sorbents. (a) cyclodextrin–lipid bilayer (β -CD-LB) magnetic nanocomposite [117]; (b) Magnetic β -CD modified graphene oxide adsorbent [115]; (c) PDA/ β -CD magnetic composite [118]; (d) magnetic Ti_3C_2 MXene sorbent functionalized with β -cyclodextrin [119].

authors reported the preparation of a cyclodextrin-lipid bilayer (β -CD-LB) magnetic nanocomposite by co-precipitation of a mixture of Fe^{2+} and Fe^{3+} in the presence of a NaOH solution containing β -CD (Fig. 7a) [117]. The $\text{Fe}_3\text{O}_4@$ β -CD-LB nanocomposite was assessed by extracting nine organochlorine pesticides (OCPs), exhibiting suitable stability and reusability.

Another common approach to prepare magnetic cyclodextrin-functionalized sorbents is to coat Fe_3O_4 NPs with a porous layer of silica and subsequently covalently bond cyclodextrins to the coated NPs. For example, Shahrehabak and coworkers reported the preparation of a poly(β -cyclodextrin-ester)-functionalized silica-coated magnetic nanoparticles (MNPs-CDP) for MSPE of the malachite green (MG) and crystal violet (CV) from water samples [120]. The synthesis of Fe_3O_4 NPs was followed by salinization, amination, and treatment with (3-glycidyloxypropyl) trimethoxy silane (EPO). To graft the poly(β -cyclodextrin-ester) (β -CDP) to the magnetic NPs surface, 0.5 g $\text{Fe}_3\text{O}_4@$ SiO_2 -GP was added to a mechanically stirred solution of 7 mL dry DMF containing 2.0 g β -CDP at 80 °C under reflux condition for 24 h under N_2 atmosphere. Then, the separation of the mixture was done with a magnet. The resulting $\text{Fe}_3\text{O}_4@$ SiO_2 /CDP sorbent performed excellently in extracting MG and CV. Chen's research group has employed a similar strategy to synthesize magnetic β -CD modified graphene oxide adsorbents [113,115]. In this case, the Fe_3O_4 NPs were coated with tetraethoxysilane (TEOS) and posteriorly derivatized with aminopropyl silica (APTES, (3-Aminopropyl) triethoxysilane) to obtain $\text{Fe}_3\text{O}_4@$ SiO_2 - NH_2 (Fig. 7b). GO was then covalently anchored to the amino groups, and the $\text{Fe}_3\text{O}_4@$ SiO_2 /GO/ β -CD sorbent was obtained through hydrogen bonding between oxygen functional groups of GO and hydroxyl of β -CD. Due to the synergy between β -CD and GO, the synthesized sorbent exhibits the ability to retain compounds, particularly those with aromatic structures, through π - π and hydrophobic interactions and the formation of host-guest inclusion complexes. As a result, it demonstrates excellent performance in the MSPE of plant growth regulators [113,115] and tetracyclines from bovine milk [102].

Another innovative sorbent was developed by Xu and coworkers, who incorporated a deep eutectic solvent (DES) composed of trifluoroacetamide and benzyl trimethylammonium chloride into β -CD previously grafted onto magnetic beads [15]. In this case, Fe_3O_4 NPs were first coated with APTES, then functionalized with β -CD, and finally impregnated with DES. The MB-NH2@ β -CD@DES sorbent demonstrated suitable performance in the extraction of trypsin from bovine pancreas crude extract, attributed to the diverse interaction mechanisms provided by the cyclodextrin moieties and the DES. β -CD can form an inclusion complex with trypsin's hydrophobic and aromatic residues, while its hydrophilic outer wall provides hydrogen bond binding sites. Additionally, the DES contributed to the extraction through dipole-dipole interactions, hydrophobic interactions, and π - π stacking interactions.

Alternatively to the coating with silica, Fe_3O_4 NPs can also be coated with polymers and further functionalized with cyclodextrins. For instance, Huang and coworkers initially coated the magnetic NPs with polydopamine (PDA) and then subjected them to a reaction with γ -CD, tetrafluoroterephthalonitrile, and K_2CO_3 (Fig. 7c) [118]. The resulting $\text{Fe}_3\text{O}_4@$ PDA/ γ -CDP sorbent exhibits good hydrophilicity, aqueous dispersibility, chemical stability, and a suitable mesoporous structure with a large specific surface area. Similarly, Ma et al. [108] coated Fe_3O_4 nanoparticles with a combination of polyhedral oligomeric silsesquioxanes (POSS) and β -cyclodextrin through surface polymerization. This resulted in forming a porous sandwich structure over the Fe_3O_4 , and the introduction of octa vinyl-POSS increased the polymers' specific surface area and long-term stability. The β -CD@POSS@ Fe_3O_4 composite excellently extracted carbaryl and carbofuran from apple samples. In another outstanding example, Boon et al. [55] prepared magnetic nanoparticles coated with poly(β -cyclodextrin functionalized ionic liquid) as a sorbent for MSPE polycyclic aromatic hydrocarbons (PAHs) in rice samples. The β -CD-Vinyl-TDI was grafted on the surface of

Fe_3O_4 nanoparticles using the polymerization-induced self-assembly (PISA) method. The obtained sorbent showed an adequate extraction capability due to the combined merits of β -CD-IL, TDI, and MNP, allowing the development of a method with relatively lower consumption of sorbent and desorption volume compared to other similar studies.

Natural polymers have also been utilized in the preparation of magnetic cyclodextrin-functionalized sorbents. For instance, Dai and Row extracted carotenes from carrot samples using an eco-friendly magnetic chitosan β -CD biopolymer [106]. The magnetic chitosan composite was synthesized by the in situ chemical precipitation of Fe^{2+} and Fe^{3+} in an alkaline solution with chitosan. Subsequently, it was functionalized through β -CD esterification and amidation reactions. This sorbent exhibited higher extraction recovery than β -cyclodextrin, chitosan, and other commercial sorbents.

Other magnetic cyclodextrin-based sorbents have been obtained by integrating MNPs, CDs, and modern hyper porous materials, such as COFs. For example, Guoliang et al. prepared a β -Cyclodextrin-functionalized magnetic covalent organic framework ($\text{Fe}_3\text{O}_4@$ COF@Au- β -CD) [109]. Fe_3O_4 NPs were coated with a covalent organic framework (COF) through a solvothermal method using benzidine 1,3,5-triformylphloroglucinol in two sequential stages. Gold NPs were first immobilized onto the $\text{Fe}_3\text{O}_4@$ COF surface, followed by functionalized with thiolated- β -CD through Au-S bonding. By combining the properties of COF and the recognition ability of β -CD, the developed sorbent exhibited excellent extraction capabilities in the MSPE of sulfonamides from meat samples.

Another approach for preparing cyclodextrin magnetic sorbents does not involve coating Fe_3O_4 NPs with a specific material. Instead, it relies on the physical immobilization of the MNPs onto the surface of 2D materials. For example, Lou et al. prepared a magnetic Ti_3C_2 MXene sorbent functionalized with β -CD [110]. Firstly, Fe_3O_4 was immobilized over the Ti_3C_2 , and the $\text{Fe}_3\text{O}_4@$ Ti_3C_2 was grafted with β -CD, obtaining a highly selective sorbent for the extraction of phytohormones from oil seeds (Fig. 7d). Yazdanpanah and Nojavan employed a similar approach to synthesize a magnetic β -CD-carbon nanotube composite [119]. This composite was prepared by reacting oxidized carbon nanotubes with cyclodextrin in the presence of hydrazine hydrate and subsequent attachment to the Fe_3O_4 NPs via chemical co-precipitation of Fe^{2+} and Fe^{3+} . The magnetic CNT-CD composite performed better than the M-CNT in extracting PAHs from aqueous media.

Matrix solid-phase dispersion extraction

Matrix solid-phase dispersion extraction technique (MSPD) is a commonly used dispersive technique for directly treating solid samples. It can be particularly beneficial in analyzing food samples, which often exist in solid form and undergo cryogenic drying and grinding during sample pretreatment. For instance, Du et al. [98] ground a portion of dried Mori Fructus powders and β -CD into a mortar at room temperature to extract multiple antioxidants. Afterward, the mixture was put into an MSPD column and eluted with an ionic liquid solution. Compared with other methods, the extraction by ionic liquid-assisted trace β -CD matrix solid-phase dispersion extraction method was more straightforward and environment-friendly due to shorter extraction time, less reagent, and less sample consumption.

Concluding remarks

Cyclodextrin-based sorbents play a relevant role in modern sample preparation techniques because of their interesting physicochemical properties. An external hydrophilic surface and an internal hydrophobic one form them. This distinct characteristic allows CDs to form inclusion complexes in a host-guest-like mechanism with many targets in food analysis. This means that CDs can interact with a broad range of compounds, from hydrophobic lipids to very polar food additives or contaminants. In addition, cyclodextrins have been combined with other

classes of sorbents, including MIPs, MOFs, COFs, and carbon-based materials [7,17]. This review discussed different preparation procedures, highlighting the range of highly selective CD-based sorbents that can be produced. We also pointed out the possibility of adapting them to various sample preparation devices such as fibers, membranes, particles, monoliths, nanocomposites, and magnetic materials [28].

In most cases, chemically-modified or hybrid CD-based materials exhibit superior mechanical and chemical resistance compared to unmodified materials. Within such a context, applications herein discussed comprised various sorbent-based sample preparation techniques, such as DSPE, MSPE, SPE, μ -SPE, SPME, and SDBS [28]. On one hand, packed and coated-based approaches (e.g., SPME or MEPS) report great accuracy, precision, and pre-concentration factors. However, the complex coating or packing process and specific hardware requirements can be challenging. On the other hand, dispersive-based approaches (i.e., DSPE and MSPE) offer a straightforward procedure, requiring only the dispersion of the sorbent in a solution. That being said, CD-based sorbents extract a very decent range of targets, including pesticides, mycotoxins, other harmful substances, and bioactive compounds such as flavonoids, vitamins, and antioxidants from food and dietary supplements.

Despite the abovementioned attractive properties, cyclodextrins are still considered an environmentally friendly class of sorbent. This comes mainly from the fact that they are naturally occurring compounds, or when it is necessary to produce them synthetically, byproducts of reaction or waste can be further used, creating a sustainable chain. Some works also report their functionalization with other materials in aqueous media without using toxic chemicals. Therefore, the combination of cyclodextrins with other compounds to generate hybrid and greener materials is a trend expected to keep increasing soon. Though, we still need more attention to this field as cyclodextrins are not so applied when compared, for example, to more traditional, not-greener materials such as C18 or other commercially available ones. Finally, integrating such sorbents in miniaturized and automated techniques holds promise for the future of sample preparation, enabling high-throughput and environmentally-friendly analytical methods that require minimal sample amounts.

Author statement

The authors declare that this manuscript has not been (and is not) submitted to any other journal. The authors agree with its publication if accepted and have no conflict of interest.

Availability of data and materials

Not applicable.

CRedit authorship contribution statement

Edvaldo Vasconcelos Soares Maciel: Conceptualization, Writing – original draft. **Natalia Gabrielly Pereira dos Santos:** Conceptualization, Writing – original draft. **Deyber Arley Vargas Medina:** Conceptualization, Writing – original draft. **Fernando Mauro Lanças:** Funding acquisition, Conceptualization, Supervision, Writing – original draft.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Data availability

Data will be made available on request.

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