

PRELIMINARY CRYSTALLOGRAPHIC DATA OF THE $\text{Ln}(\text{PF}_6)_3 \cdot 4\text{TDTD} \cdot 4\text{H}_2\text{O}$ COMPLEXES-*Mueller, R.A. (1), Ulbrich, H.H. (2), Vicentini, G. (1) and Giesbrecht, E. (1)

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Vicentini et al (J. Inorg. Nucl. Chem., 37, 2021, 1975) synthesized a series of complexes $\text{Ln}(\text{PF}_6)_3 \cdot 4\text{TDTD} \cdot 4\text{H}_2\text{O}$ ($\text{Ln} = \text{La-Lu}$, Y; TDTD = 1,4 - dithiane - 1,4 - dioxide) as very fine powders. Recrystallized single crystals are always complexly twinned and show the same D-S powder pattern as the original powders. Precession, Weissenberg and rotation methods were used to determine the crystallographic characteristics of the compounds with $\text{Ln} = \text{La}$, Ce, Yb. The La and Ce compounds are monoclinic with β close to 90° , and P2, Pm, P2/m as possible space groups (diffraction symbol is $2/mP \dots$). The Yb compound shows orthorhombic symmetry, space group Pnca (Pbcn is the standard orientation). Doubling of two axes increases the unit cell content from 1 in the monoclinic to 4 in the orthorhombic crystals. D-S powder films were made of all the recrystallized material (Cr K α , Si as internal standard) and a few lines computer-indexed. Cell constants were calculated using conventional least-squares computer methods. All complexes with $\text{Ln} = \text{Y}$, La to Tm, have comparable powder patterns and are thus monoclinic, while the Lu pattern is identical to that of the orthorhombic Yb. Cell parameters, change from $a = 9.05(1)$, $b = 9.12(2)$, $c = 12.44(1)$ Å, $V = 1026(2)$ Å³ (Y compound) to $a = 8.93(1)$, $b = 9.31(2)$, $c = 12.41(2)$ Å, $V = 1031(2)$ Å³ (Tm compound) while β changes from $92^\circ 06'(6)$ to $91^\circ 51'(10)$; corresponding data for the Yb compound are $17.19(4)$, $17.75(3)$, $13.37(2)$ Å, $V = 1020(3)$ Å³ (V per molecule). There is, within the low resolution of the cell parameters, an irregular tendency for the cell parameters to decrease, probably controlled by a small overall lanthanide contraction effect (Siekierski, J. Inorg. Nucl. Chem., 33, 377, (1971)).

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